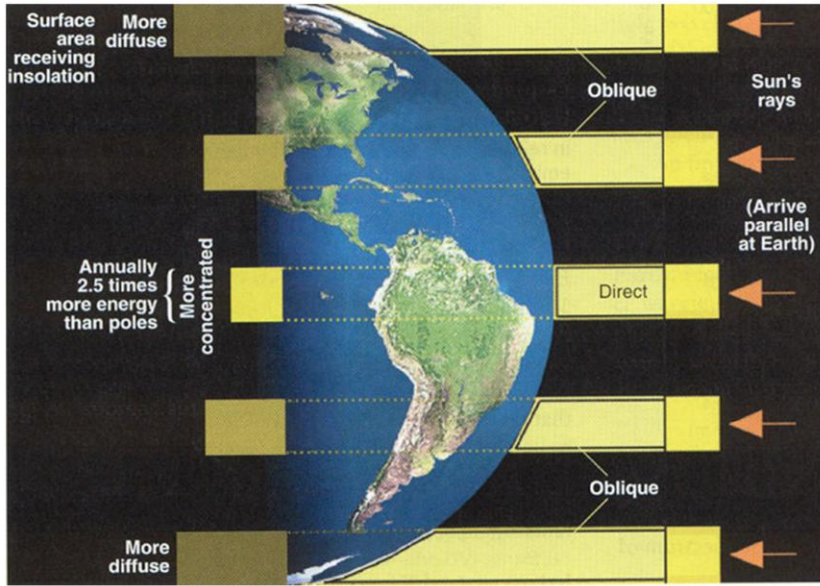


Reasons for variations in insolation with time and place on Earth

- **Earth's Revolution:** revolves around the sun (365.24 days for annual revolution)
- **Earth's Rotation:** 24 hrs; west to east; determines daylength (circle of illumination)
- **Tilt of Earth's axis:** 23.5° from plane of ecliptic (seasons)
- **Earth's Sphericity:** produces the uneven receipt of insolation from pole to pole

Insolation on a curved surface

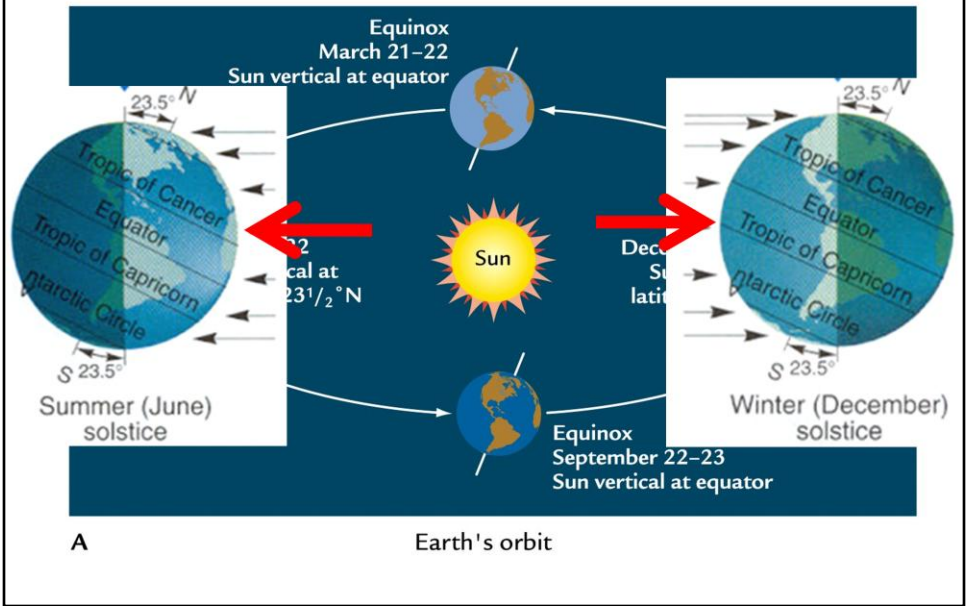


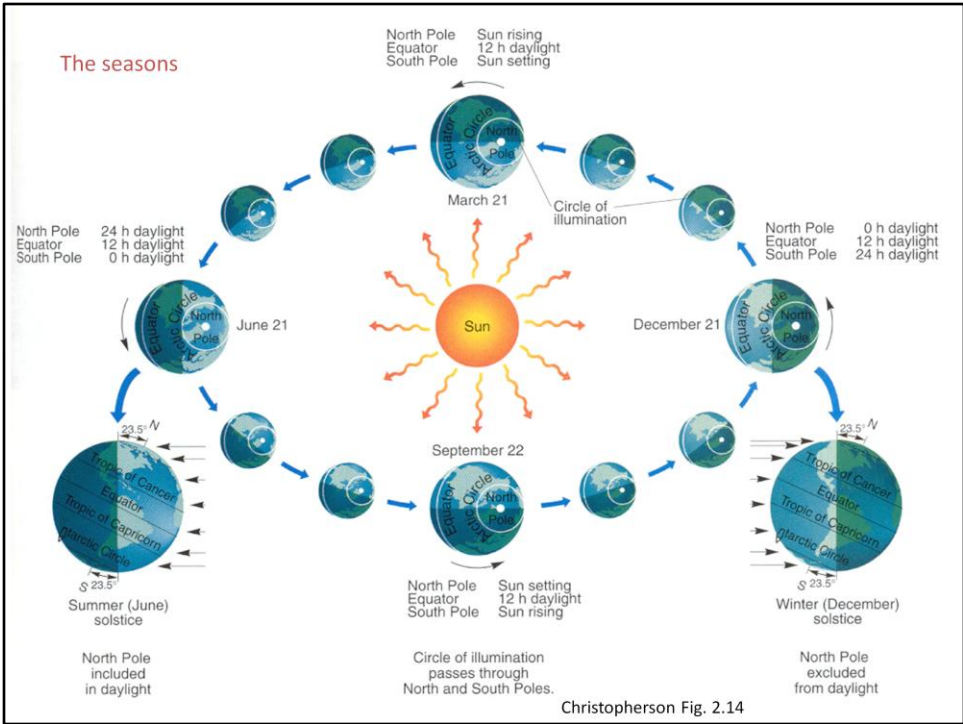
Christopherson Fig. 2.9

Seasonality

- Affects both:
 1. amount of solar radiation (angle of rays received)
 2. day length
- Related to the tilt of the earth's axis
 - Equinox (Sun over Equator):
 - Vernal March 20-21
 - Autumnal September 22-23
 - Solstice (Sun over 23.5°):
 - Winter December 21-22 (23.5° S)
 - Summer June 20-21 (23.5° N)

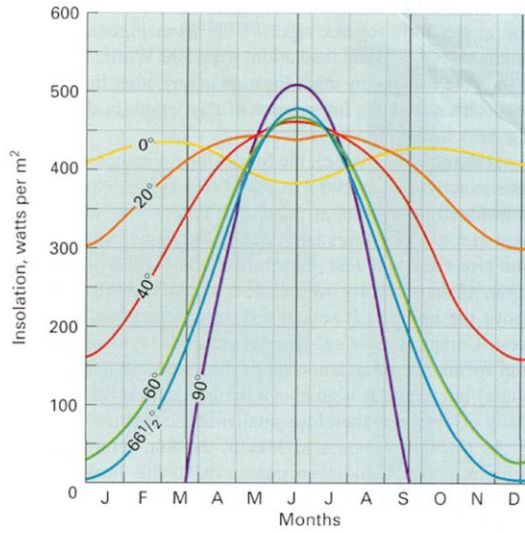
Equinoxes and Solstices

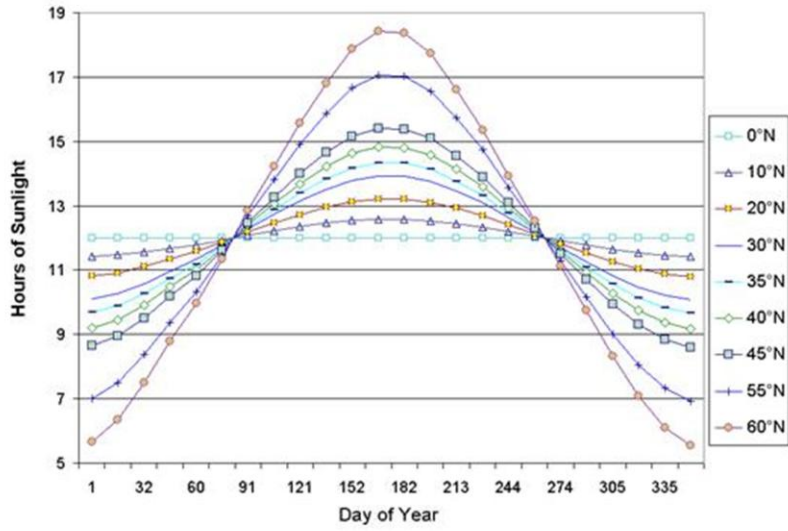




Insolation variations by latitude and season
at the upper atmosphere

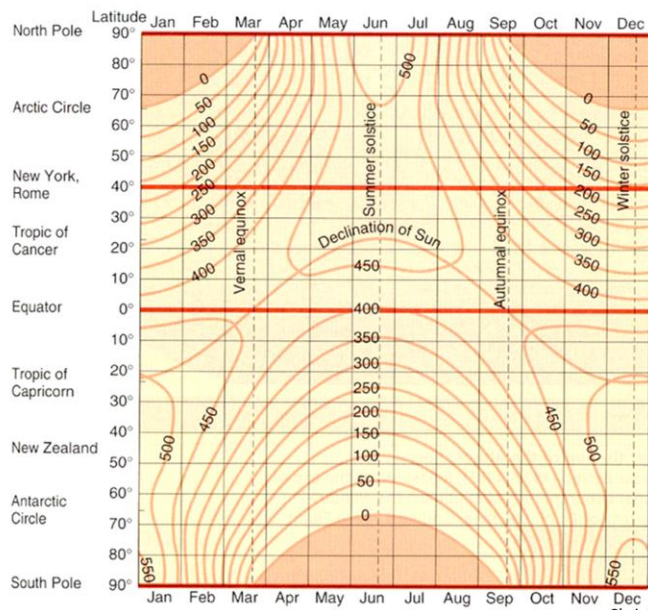
24 hour average values at given season and latitude



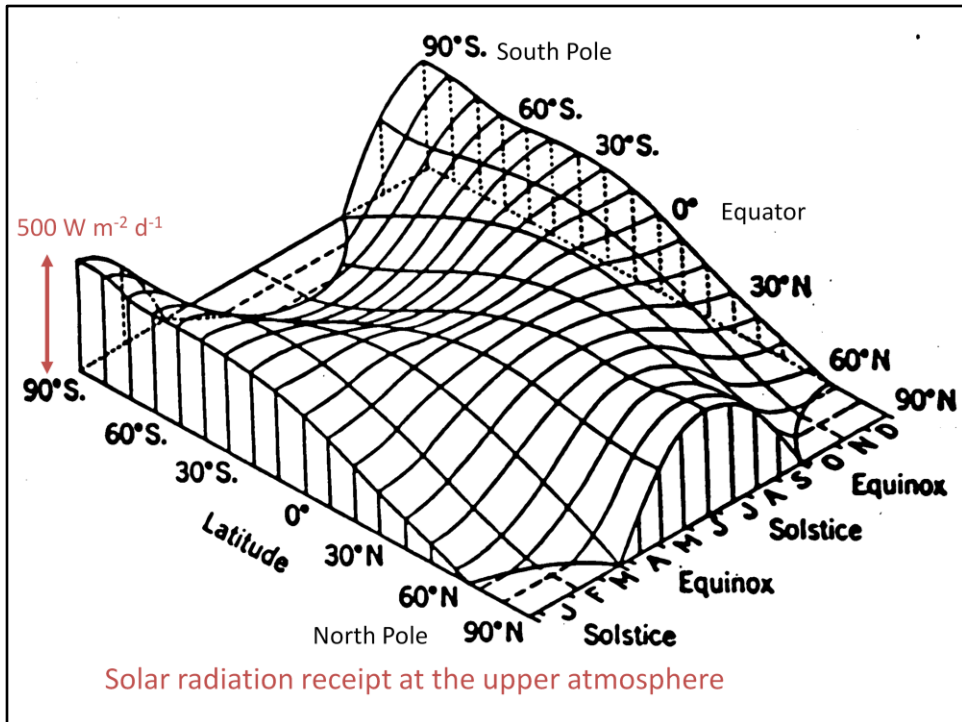


www.isws.illinois.edu/data/altcrops/giclim.asp

Radiation received in the upper atmosphere, by latitude and date



Christopherson Fig. 2.10



So what are the implications of
this uneven receipt of solar
radiation with latitude?

...why do we care?

Seasonal observations at a temperate bog



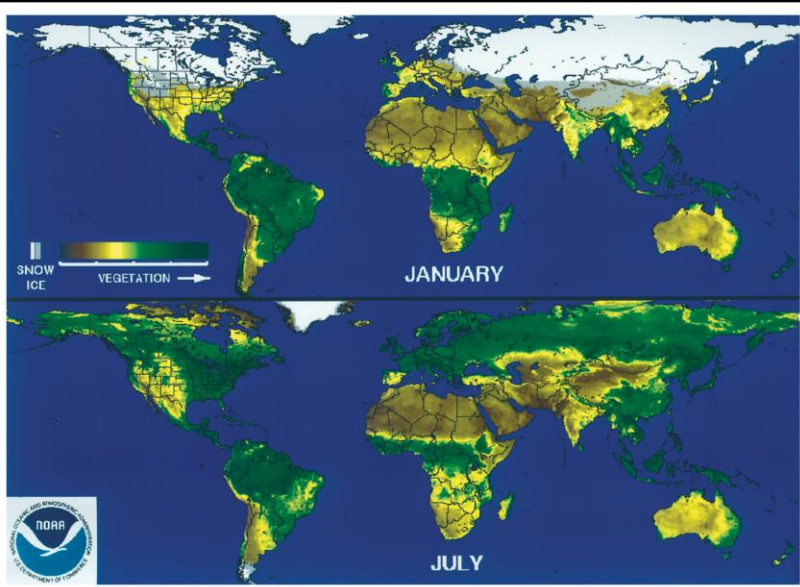
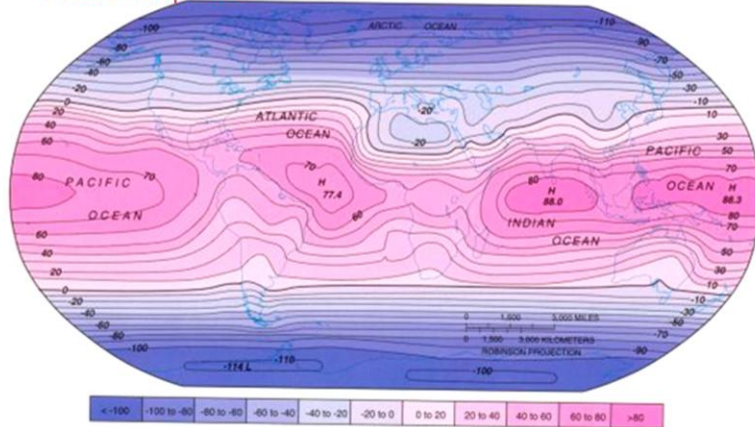


FIGURE 2.18 Seasonal change from orbit.
 Seasonal change is monitored and measured by sensors aboard polar-orbiting satellites. Compare and contrast similar regions for January and July: central Canada, China, Argentina, and Europe. Orbital remote sensing is tracking an earlier spring season worldwide. [Images courtesy of Garik Gutman, NOAA, National Environmental Satellite, Data, and Information Service.]

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Net radiation imbalance with latitude

Daily net radiation (W/m^2) patterns at the top of the atmosphere



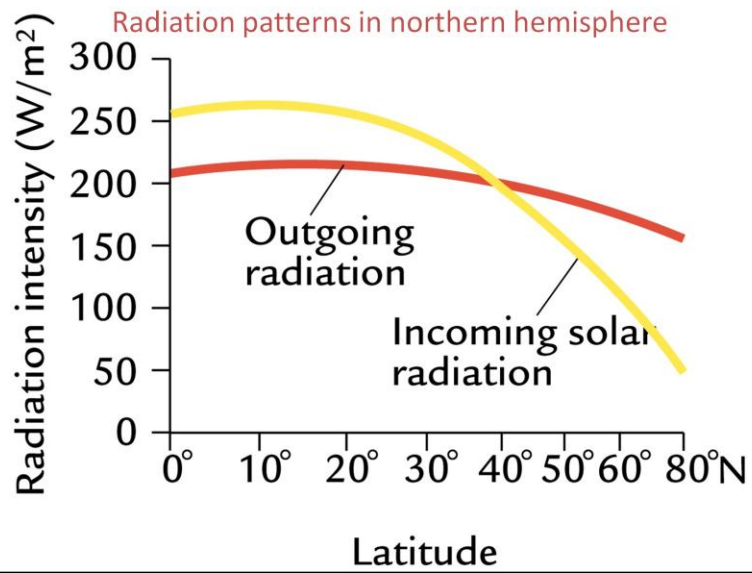
Net radiation = all incoming – all outgoing radiation

Positive at Tropics

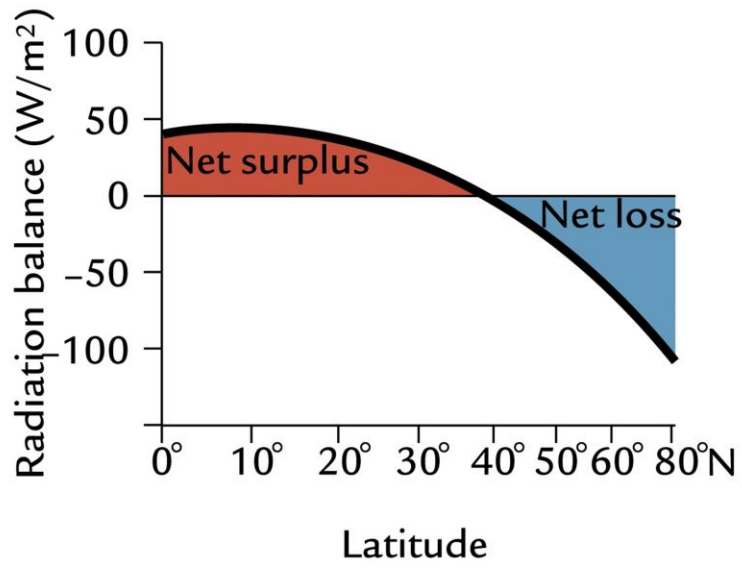
Negative at Poles

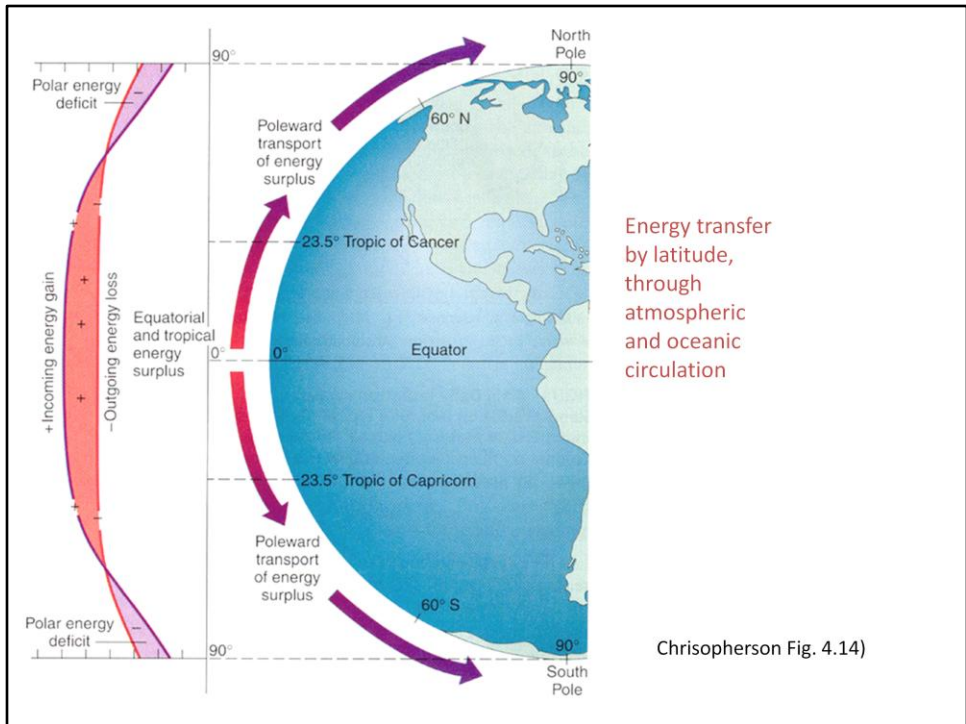
Christopherson Fig. 2.11

Net radiation imbalance driven by differences in incoming solar radiation

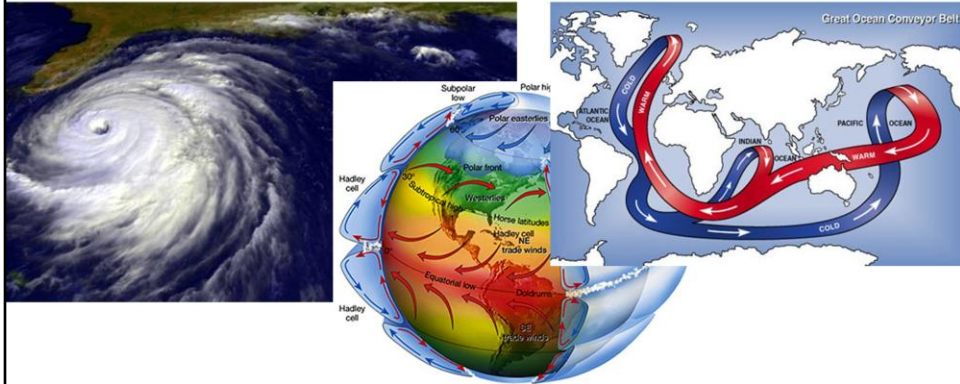


Radiation balance in northern hemisphere





This net energy imbalance
between tropics and poles
drives global circulation
patterns



*The Atmosphere:
Earth's security blanket!*



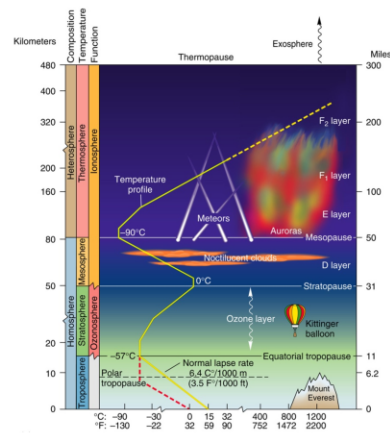
Earth's Modern Atmosphere

- Composition
- Temperature
- Function

Readings: Chapter 3 (pp. 63 – 75)

Atmospheric Profile

- Atmosphere extends ~480 km from the surface
- Beyond that the atmosphere is rarified (nearly a vacuum) - known as the *exosphere* extending outwards as far as 32,000 km
 - contains scarce helium and hydrogen atoms weakly bound by gravity

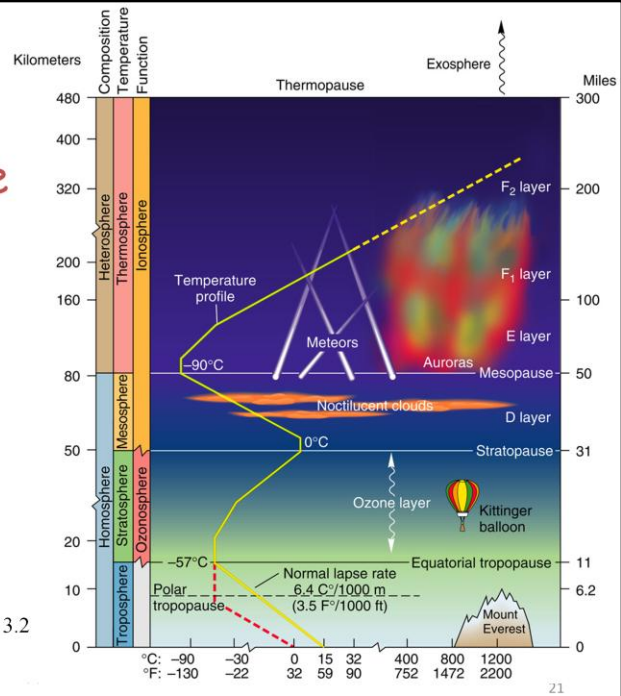


Atmospheric Profile

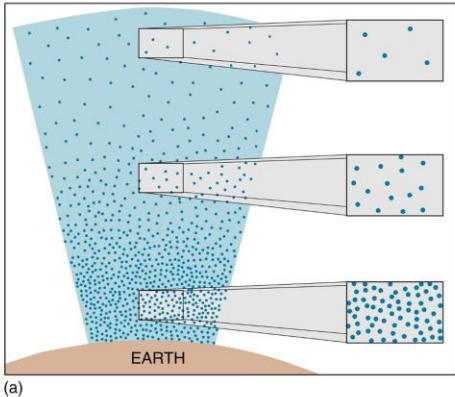
- The atmosphere can be divided into shells (layers) by
 - Composition
 - Temperature
 - Function
- Different layers called spheres
 - E.g. troposphere
- Top boundary of each layer called -
pause
 - E.e. tropopause

Profile of the Atmosphere

Figure 3.2

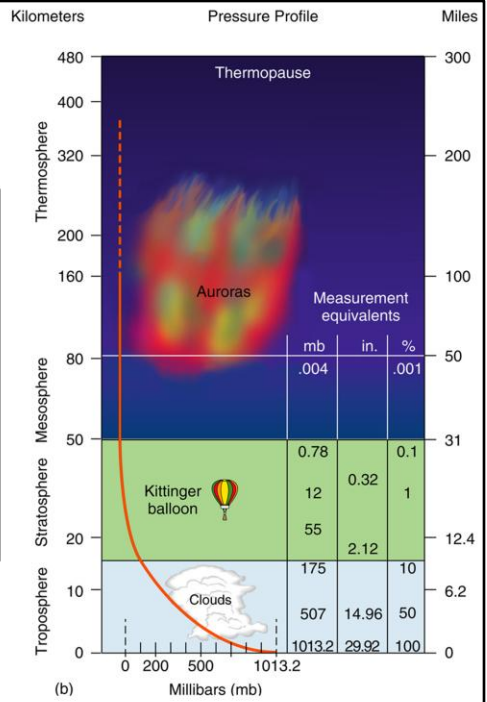


Atmospheric Pressure



(a)

Figure 3.3



(b)

1960: 31 km Experiencing the stratosphere

2012: 39 km



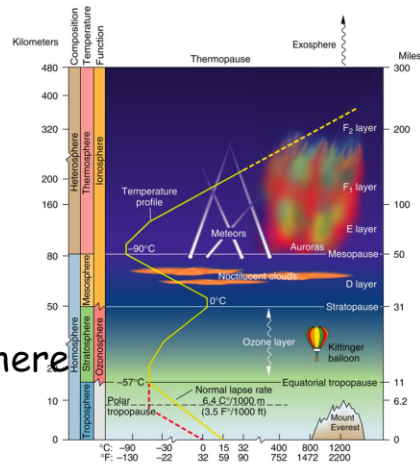
FIGURE 1 Stratospheric leap into history. Moments into Captain Joseph Kittinger's historic exploration of the atmosphere, captured by a remotely triggered camera. The lanyard cord is attached to his parachute pack and snaps to set a timer that opens a small stabilization chute 16 seconds into the free fall. The clouds are more than 26,000 m (85,000 ft) below him. He carries an instrument pack on his seat, his main chute, and pure oxygen for his breathing mask. [Volkmar Wentzel/NGS Image Collection, used by permission from the National Geographic Society, *National Geographic Magazine*, Dec. 1960, p. 855. All rights reserved.]



Felix Baumgartner jumps out of his capsule from more than 24 miles above the earth on Oct. 15, 2012. Baumgartner shattered the sound barrier Sunday while making the highest jump ever — a tumbling, death-defying plunge from a balloon to a safe landing in the New Mexico desert. Here's a look at how Baumgartner prepared for his feat. (Red Bull Stratos/AP)

Atmospheric Composition

- **Heterosphere** - outer atmosphere
 - gases are not mixed
 - H & He in outer layer
 - O & N in lower layers
 - > 80 km
- **Homosphere** - inner atmosphere
 - gases blended (mostly)
 - 0 - 80 km
 - ~99.999% of the atmosphere



Composition of the Earth's Atmosphere

Constituent	Formula	Content	Role
Nitrogen	N ₂	78.1 %	Inert
Oxygen	O ₂	21.0 %	Absorbs U-V radiation
Argon	Ar	0.9 %	Inert
Carbon dioxide	CO ₂	370 ppm	Absorbs IR radiation
Neon	Ne	18 ppm	Inert
Helium	He	5 ppm	Inert
Methane	CH ₄	1.8 ppm	Absorbs IR radiation
Nitrous oxide	N ₂ O	0.3 ppm	Absorbs IR
Carbon monoxide	CO	0.01 ppm	Reactive gas
Ozone	O ₃	0.01 ppm	Reactive gas, absorbs U-V and IR radiation
CFCs	Various	< 0.001 ppm	Absorbs IR, destroys O ₃
Water vapour	H ₂ O	0.01 - 4 %	Absorbs IR radiation

"Greenhouse" gases

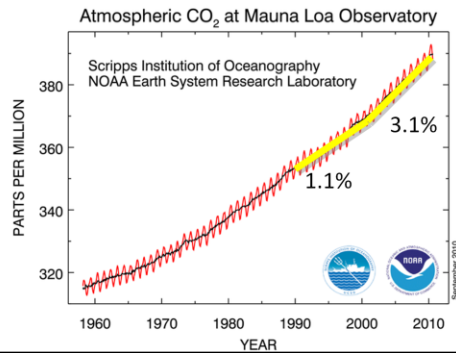
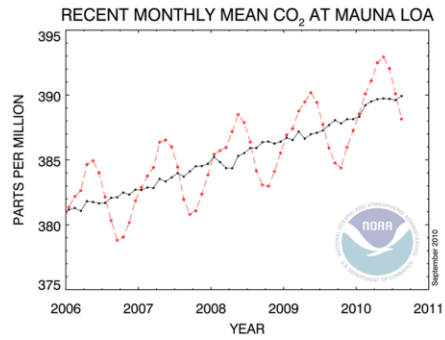
Variability in CO_2 concentration
1) over the past few years...

What causes the pattern?

2) over the past 50 years

Scenario → 1390 ppm by 2100

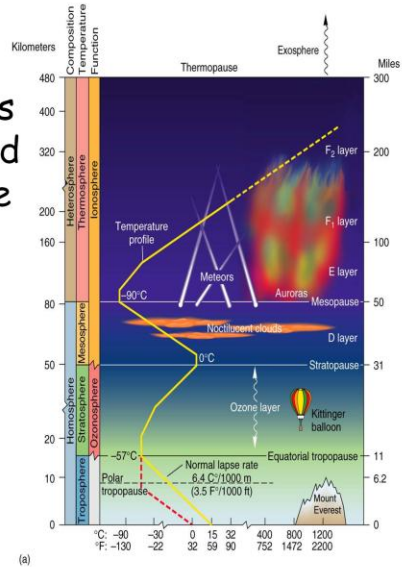
<http://www.esrl.noaa.gov/gmd/ccgg/trends/>



Atmospheric Profile - Temperature

• Earth's atmosphere contains four different layers defined according to air temperature

- Troposphere
- Stratosphere
- Mesosphere
- Thermosphere

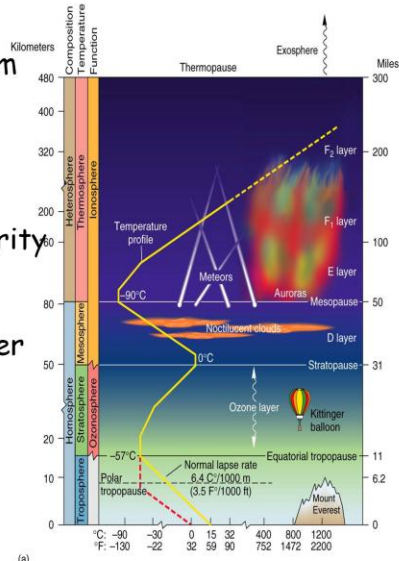


(a)

Atmospheric Profile - Temperature

Troposphere

- extends from the surface to ~ 18 km
- ~ 90 % of the total mass of the atmosphere is contained in this layer
- Unstable: the layer where the majority of our weather occurs
- maximum air temperature in the lower atmosphere occurs in this layer
- with increasing height, air temperature dec. at a rate of ~6.4 °C/1000 m (normal lapse rate)

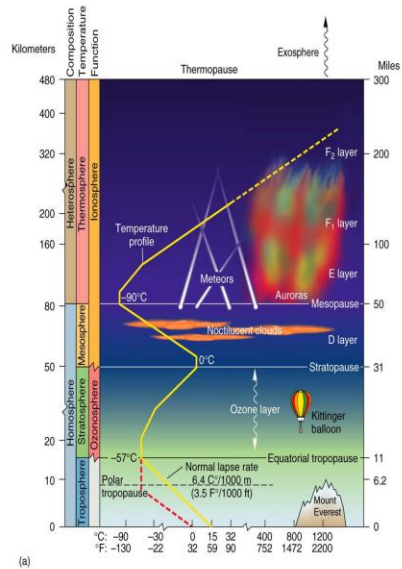


(a)

Atmospheric Profile - Temperature

Tropopause

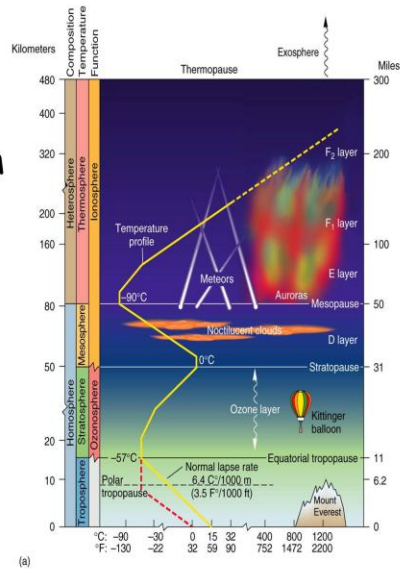
- extending from 8 to 18 km
- an isothermal layer
- temperature remains constant
- the layer where the jet stream occurs



Atmospheric Profile - Temperature

Stratosphere

- extends from 18 to 50 km
- temperature inc. with altitude
 - why?
 - ozone layer



(a)

Ozone layer

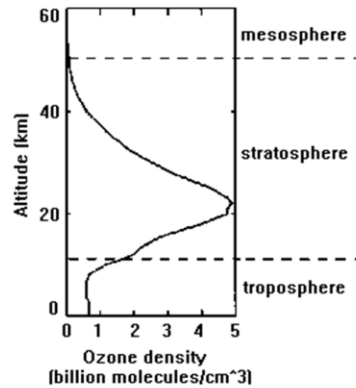
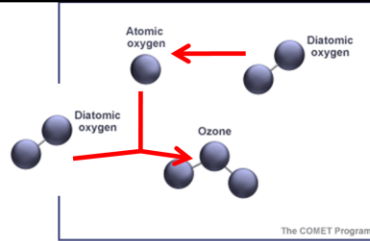
Ozone - molecule made up of three oxygen atoms joined together (O_3)

•concentrated in a layer of the stratosphere called the ozone layer, lies between 15-48 km above the Earth's surface

Why is ozone important?

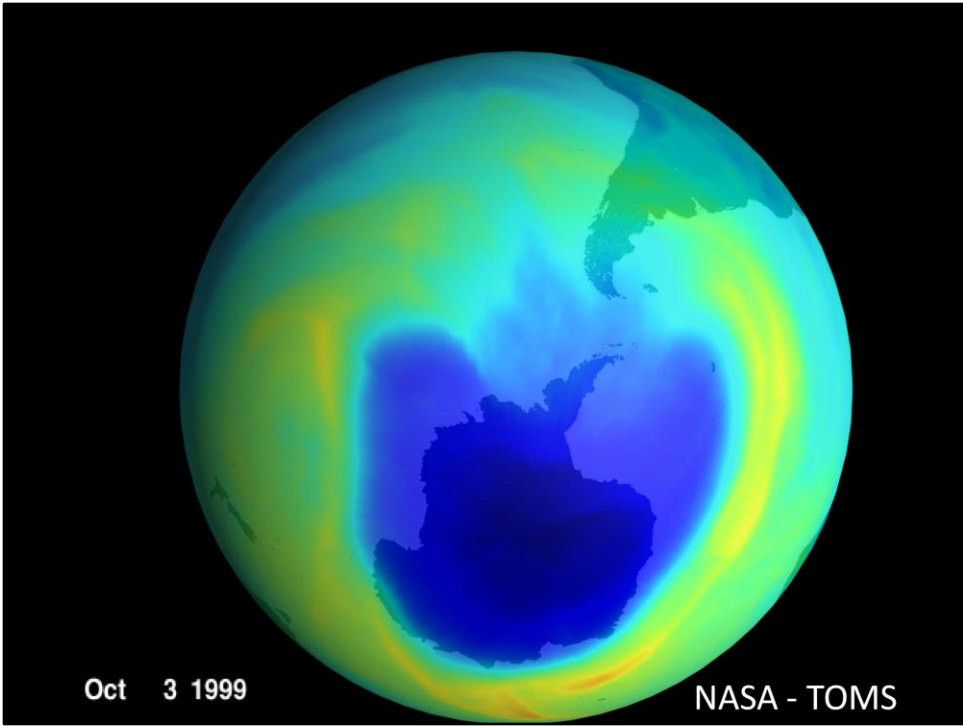
How is ozone formed and destroyed?

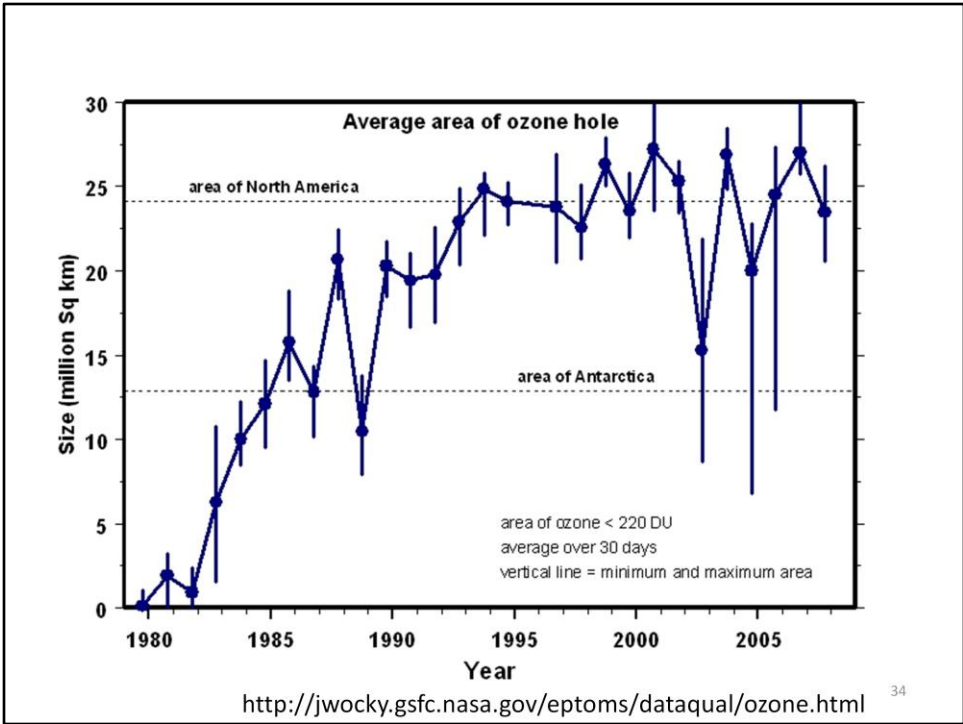
What is the Montreal Protocol?



Hair spray + 80s =







Weather.gov Forecast
City, ST GO

- » Active Weather Alerts
- » NOAA Organizations
- » Working With NOAA
- » Media & Constituents
- » NOAA In Your State
- » Emergency Information for NOAA Employees

- Related Links
- » [NOAA's South Pole Ozone Update Oct 1, 2012 \(YouTube video\)](#)
 - » [NOAA's South Pole Ozone](#)
 - » [NASA's Ozone Hole Watch](#)
 - » [Suomi National Polar-orbiting Partnership, NPP and launch](#)

- Media Contact
- » [Katy Human](#) (NOAA)
303-497-4747
 - » [Steve Cole](#) (NASA)
202-358-0918

 Tweet (185)

NOAA, NASA: Antarctic ozone hole second smallest in 20 years

October 24, 2012

Warmer air temperatures high above the Antarctic led to the second smallest seasonal ozone hole in 20 years, according to NOAA and NASA satellite measurements. This year, the average size of the ozone hole was 6.9 million square miles (17.9 million square kilometers). The ozone layer helps shield life on Earth from potentially harmful ultraviolet (UV) radiation that can cause skin cancer and damage plants.

The Antarctic ozone hole forms in September and October, and this year, the hole reached its maximum size for the season on Sept. 22, stretching to 8.2 million square miles (21.2 million square kilometers), roughly the area of the United States, Canada and Mexico combined. In comparison, the largest ozone hole recorded to date was in 2000 at 11.5 million square miles (29.9 million square kilometers).

The Antarctic ozone hole began making a yearly appearance in the early 1980s, caused by chlorine released by manmade chemicals called chlorofluorocarbons or CFCs. The chlorine can rapidly break apart ozone molecules in certain conditions, and the temperature of the lower stratosphere plays an important role.

"It happened to be a bit warmer this year high in the atmosphere above Antarctica, and that meant we didn't see quite as much ozone depletion as we saw last year, when it was colder," said Jim Butler with NOAA's Earth System Research Laboratory in Boulder, Colo.

Even 25 years after an international agreement was signed to regulate production of ozone-depleting chemicals, the ozone hole still forms each year. In fact, it could be another decade before scientists can detect early signs of Antarctic ozone layer recovery, according to a paper by NOAA researchers and colleagues published last year. The ozone layer above Antarctica likely will not return to its early 1980s state



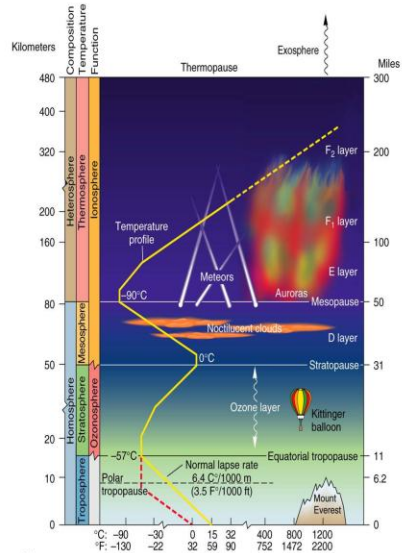
Staff at the South Pole get ready to release a balloon that will carry an ozone instrument up to 20 miles in the atmosphere, measuring ozone levels all along the way. NOAA image from 2011.

[Download here](#) (Credit: NOAA.)

Atmospheric Profile - Temperature

Mesosphere

- 50 - 80 km
- atmosphere reaches its coldest temperatures ($\sim -90^{\circ}\text{C}$) at a height of ~ 80 km

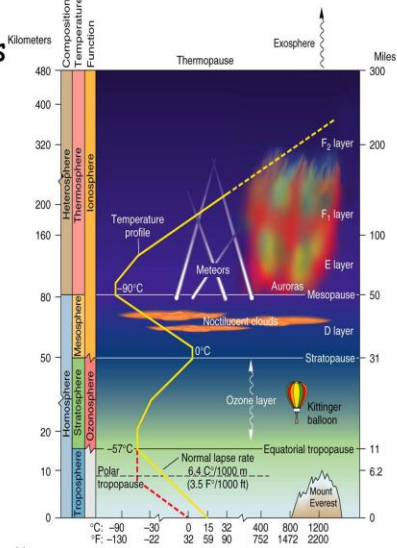


http://science.nasa.gov/spaceweather/nlcs/gallery2004_page1.html (a)

Atmospheric Profile - Temperature

Thermosphere

- last atmospheric layer - altitudes greater than 80 km
- intense solar radiation excites individual molecules to high levels of vibration - vibration is the energy of motion (kinetic energy) measured as **temperature** which rises to 1200 °C
- however, few molecules - v. low transfer of **sensible heat** (flow of kinetic energy from one molecule to another)



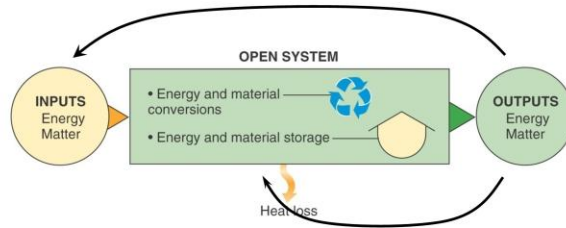


Atmosphere and Surface Energy Balances

•Radiant energy pathways

Readings: Chapter 4

1999 9 1 12:22



Solar energy passes through the troposphere to the Earth's surface

-**Input:** insolation

-**Output:** reflected shortwave radiation and emitted longwave radiation

-The difference is **net radiation**, the energy used to drive many of Earth's abiotic and biotic systems

At the surface ...

$$K \downarrow - K \uparrow + L \downarrow - L \uparrow = R_n = H + LE + G + \Delta S$$

Radiation balance

energy balance

$K \downarrow$ ($SW \downarrow$)

$K \uparrow$ ($SW \uparrow$)

$L \downarrow$ ($LW \downarrow$)

$L \uparrow$ ($LW \uparrow$)

R_n ($NET R$)

K = shortwave radiation

L = Longwave radiation

Rn = net radiation

H= Sensible heat

LE= latent heat (evaporation)

G= ground heating and cooling

40

$K \downarrow$ at the surface (insolation) is much less than the solar constant...

Solar radiation has many fates/pathways as it travels to the surface:

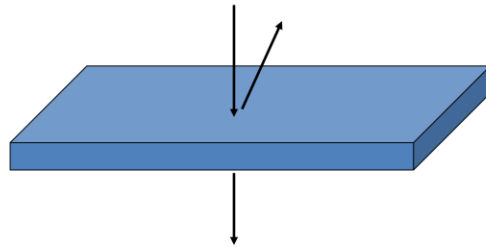
- Transmission
- Reflection
- Absorption
- Scattering
- Refraction

•Transmission, Absorption, Reflection of Solar Radiation

- Transmission: passage through the atmosphere, etc
- Absorption: assimilation by molecules & its conversion from one form of energy to another
- Reflection: energy that bounces back into space without being absorbed or changed in any way

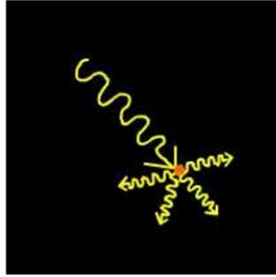
Energy that is not transmitted (T) is either reflected (R) or absorbed (A)

$$100\% = A + R + T$$



- Scattering

- Changing direction of light's movement, without altering its wavelengths
- Produces diffuse radiation (vs. direct)



You see scattering everyday!

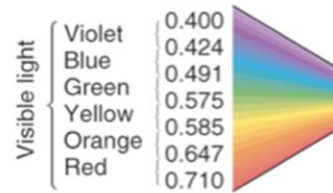
Why the sky is blue!



Why polluted skies are white



Why sunsets are red/orange



- Refraction

-Change in speed and direction of light

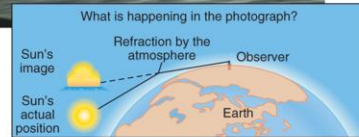
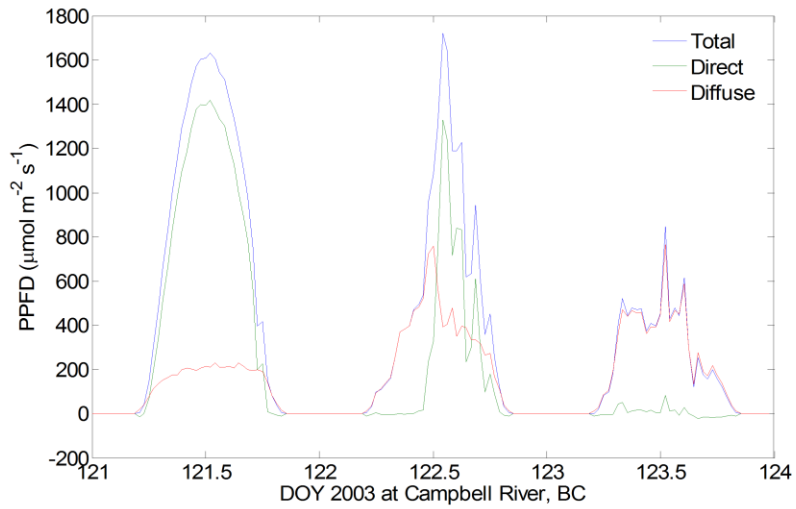


FIGURE 4.3 Sun refraction.

The distorted appearance of the Sun nearing sunset over the ocean is produced by refraction of the Sun's image in the atmosphere. Have you ever noticed this effect? [Photo by Robert W. Christopherson.]

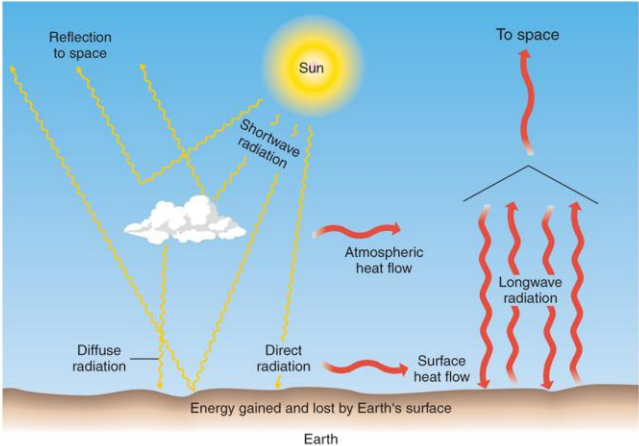
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Can you tell what type of weather occurred during these three days at Campbell River, BC?



Radiant energy pathways & principles

Shortwave energy Longwave energy



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- Insolation input ($K \downarrow$): All radiation received at Earth's surface - direct and indirect

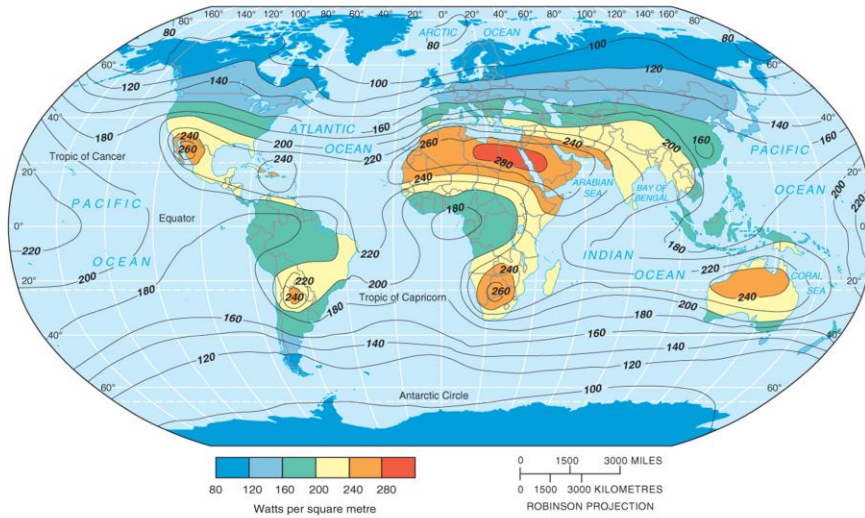
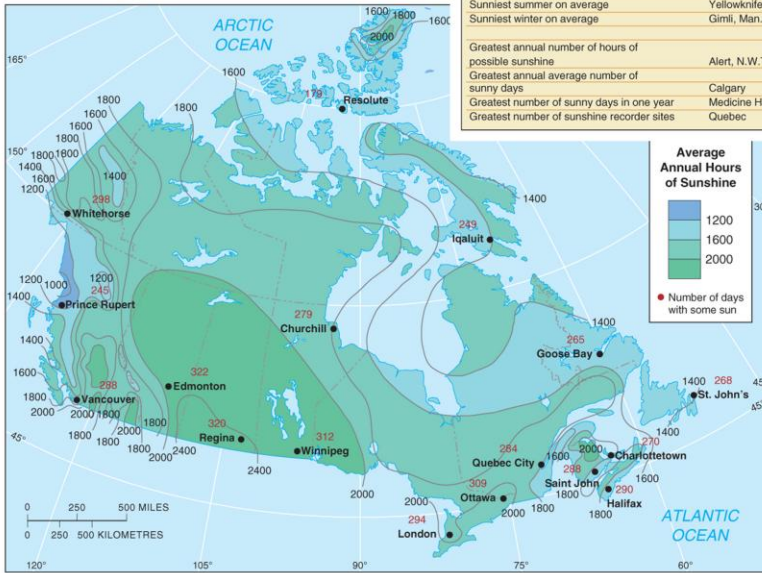


FIGURE 4.4 Insolation at Earth's surface.

Average annual solar radiation received on a horizontal surface at ground level in watts per square metre ($100 \text{ W/m}^2 = 75 \text{ kcal/cm}^2/\text{year}$) [After M. J. Budde, *The Heat Balance of the*

Sunshine Hours in Canada

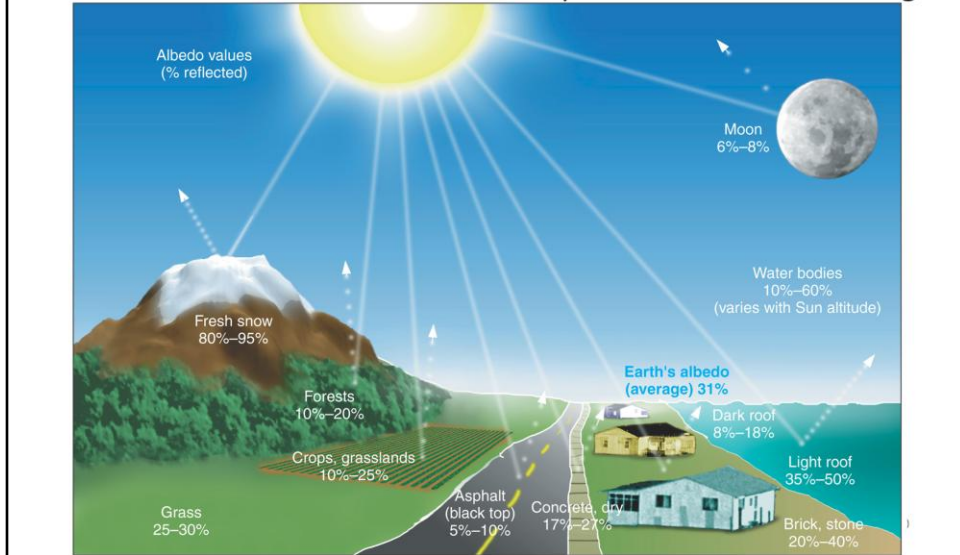


Sunniest Places in Canada

Sunniest month	Eureka, N.W.T.	621 hours	May 1973
Greatest average annual number of sunny hours	Estevan, Sask.	2537	
Major city with greatest annual average number of sunny hours	Saskatoon	2450	
Sunniest small town under 10,000 population	Coronation, Alta.	2490	
Sunniest provincial capital	Regina	2331	
Sunniest year on record	Manyberries, Alta.	2785	1976
Shortest spell of consecutive days without sun	Charlottetown	10 days	
Sunniest summer on average	Yellowknife	1065 hours	June, July, August
Sunniest winter on average	Gimli, Man.	376 hours	December, January, February
Greatest annual number of hours of possible sunshine	Alert, N.W.T.	4580	
Greatest annual average number of sunny days	Calgary	329	
Greatest number of sunny days in one year	Medicine Hat	346	1976
Greatest number of sunshine recorder sites	Quebec	84	

Albedo: reflected shortwave radiation/incoming shortwave radiation

- surface characteristic & can depend on sun altitude angle



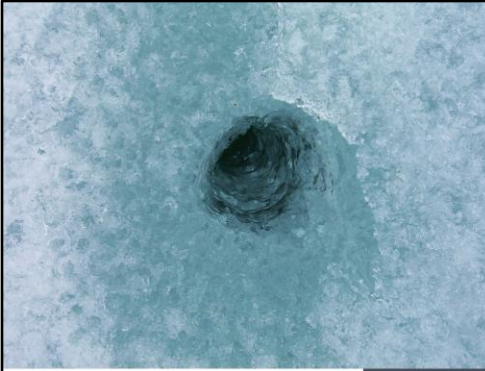
Albedo



Cloudberry leaf and ptarmigan droppings on Daring Lake ice, NWT, May 2006

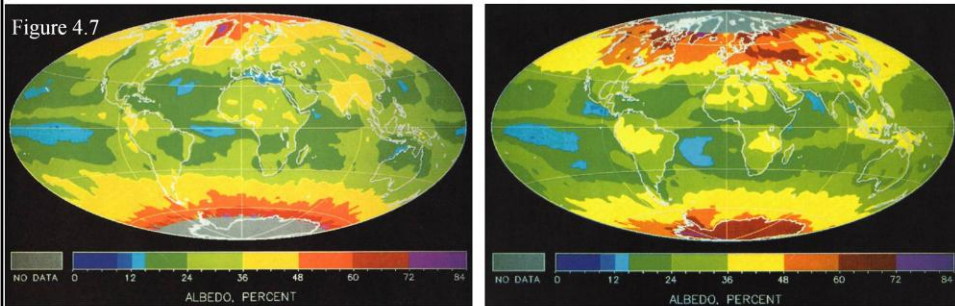
51

Water has a
lower albedo
than ice



Seasonal and Spatial Variations in Albedo

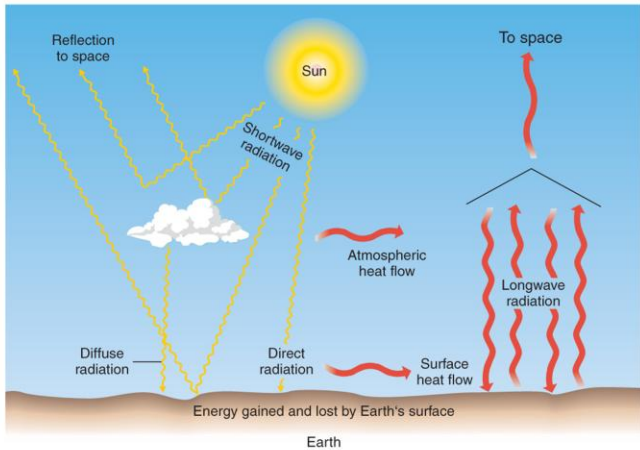
Which image is likely taken in January?



Radiant energy pathways & principles

$$R_n = K \downarrow - K \uparrow + L \downarrow - L \uparrow$$

Shortwave energy Longwave energy



Longwave radiation

How much longwave radiation do objects emit?

- All objects above absolute zero emit radiation

$$L = \varepsilon \sigma T^4$$

T is temperature
in Kelvin

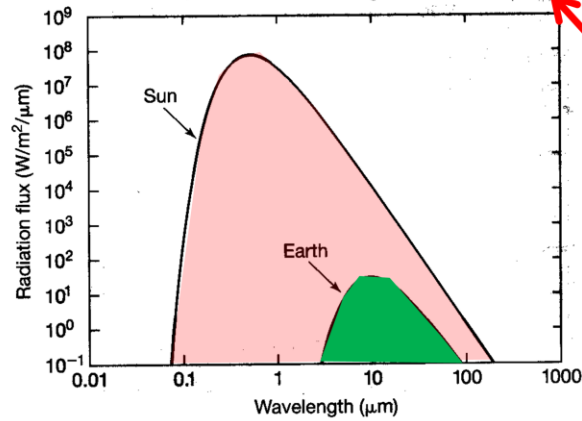
σ is the Stefan-Boltzmann constant
= $5.67 \times 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$

ε is the **emissivity** of the object

Radiation: the basics

Hotter objects emit more total radiation- curve for black body

$$\text{Total radiation emitted} = \epsilon \sigma T^4$$



Deviation from black body

Emissivity describes how readily the object emits L
($\epsilon = 1$, object is a perfect blackbody and emits the full amount of L possible based on temp)

<u>Surface</u>	<u>Emissivity</u>
plant leaves	0.94-0.99
soil	0.93-0.96
water	0.96
animals	0.98-0.99
clear sky	0.5 - 0.75
overcast sky	~0.95

Sky emissivity depends on:

- water vapour content
- cloud cover
- air temperature

Quinn is a pretty good
black body, $\varepsilon \sim 0.99$

...so is Abby, $\varepsilon \sim 0.99$

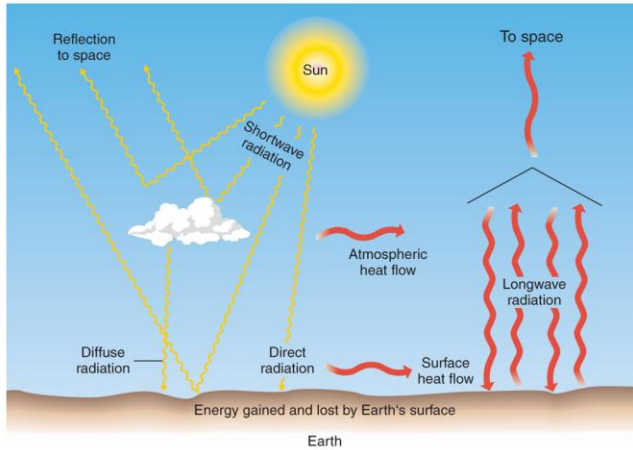


ε does not depend on
colour, skin, or fur!

Radiant energy pathways & principles

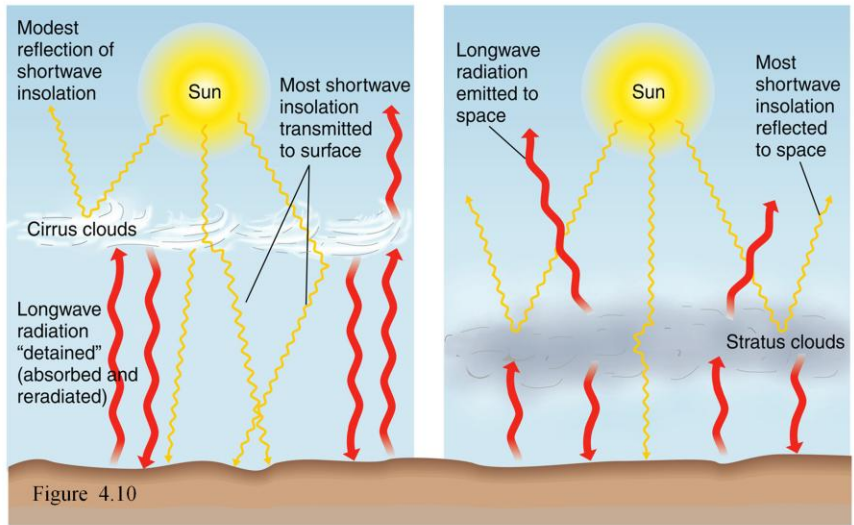
$$R_n = K \downarrow - K \uparrow + L \downarrow - L \uparrow$$

Shortwave energy Longwave energy



$L \downarrow$

•atmospheric gases & dust, clouds emit L in all directions (including down!)



(a) High clouds: net greenhouse forcing and atmospheric warming

(b) Low clouds: net albedo forcing and atmospheric cooling

Fig 4.11

The effects of jet contrails



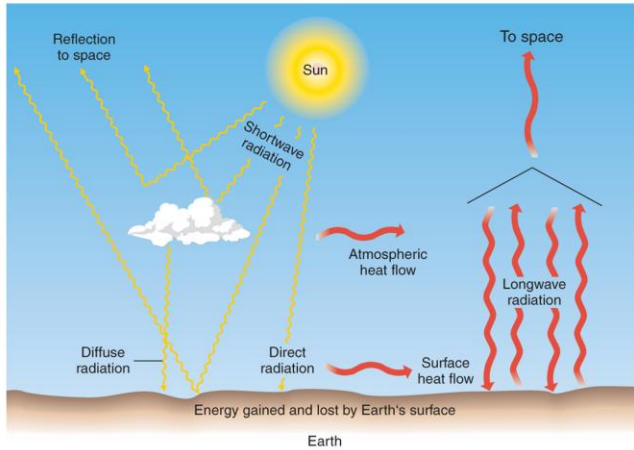
(a)

Fig 4.10c

Radiant energy pathways & principles

$$R_n = K \downarrow - K \uparrow + L \downarrow - L \uparrow$$

Shortwave energy Longwave energy



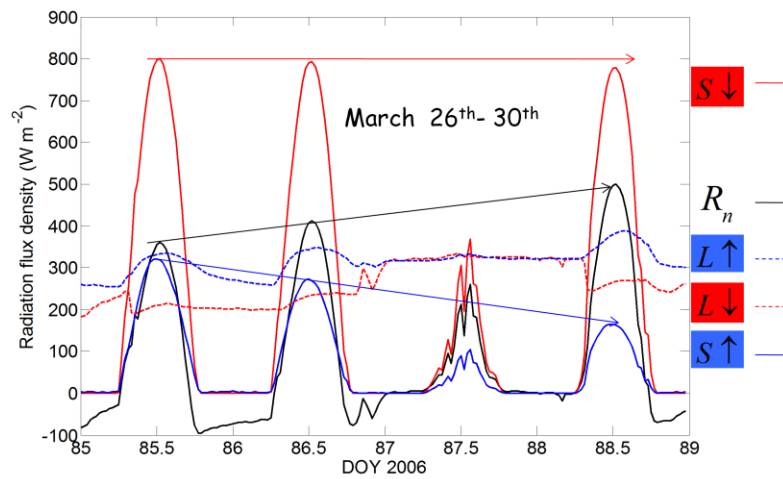
$L \uparrow$

•Surface objects emit L in all directions

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Local radiation budgets at the surface (microclimatology)

Mer Bleue bog, March 26 - March 29, 2006



How do changes in the weather & in the surface
affect radiation budget?

Global and local patterns of energy budgets

•Energy Balance

- Radiation budget
- Energy budget
- Urban heat island



Readings: Chapters 4

Radiation Budget

$$R_n = \text{inputs} - \text{outputs}$$

$$R_n = K\downarrow - K\uparrow + L\downarrow - L\uparrow \quad (\text{W m}^{-2})$$

R_n = net radiation (NET R)

$K\downarrow$ = incoming shortwave radiation (insolation, $SW\downarrow$)

$K\uparrow$ = outgoing shortwave radiation ($SW\uparrow$)

$L\downarrow$ = incoming longwave radiation ($LW\downarrow$)

$L\uparrow$ = outgoing longwave radiation ($LW\uparrow$)

all units in W m^{-2}

Sources? Pathways?

K_{\downarrow} From the sun - some gets absorbed & reflected before reaching the surface - includes both diffuse & direct

K_{\uparrow} Reflected off clouds & particles in atm, off ground surface objects - depends on albedo & sun angle.
 $K_{\uparrow} = \alpha K_{\downarrow}$

L_{\downarrow} Emitted from clouds & other atm constituents. At the surface, will also include L from overhead/nearby trees, buildings, etc. $L_{\downarrow} = \epsilon \sigma T^4$

L_{\uparrow} Emitted from surface objects. $L_{\uparrow} = \epsilon \sigma T^4$

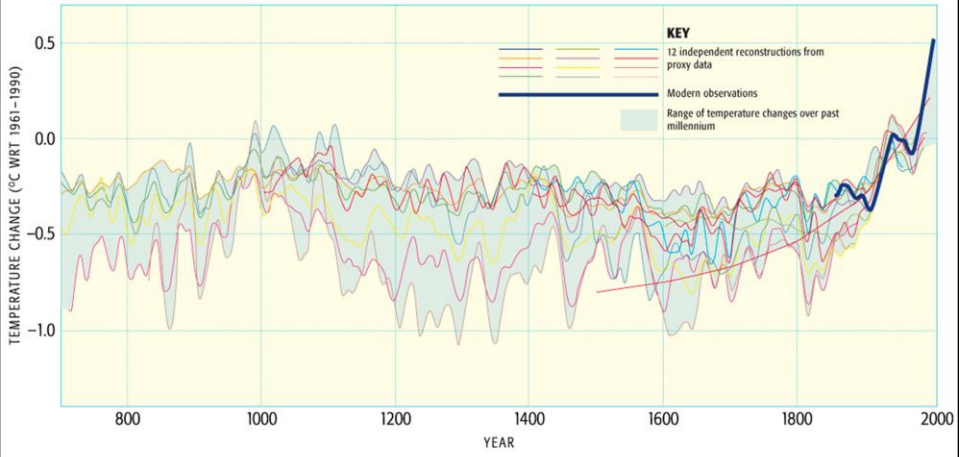
Radiation Budget

Top of the atmosphere
→ is $R_n > 0$, < 0 , or $= 0$?

At Earth's surface
→ is $R_n > 0$, < 0 , or $= 0$?

Dynamic equilibrium?

NORTHERN HEMISPHERE TEMPERATURE CHANGES OVER THE PAST MILLENNIUM



Radiation Budget

$$R_n = \text{inputs} - \text{outputs}$$

$$R_n = K\downarrow - K\uparrow + L\downarrow - L\uparrow$$

R_n = net radiation (NET R)

$K\downarrow$ = path changes in Earth's orbit around sun, etc.

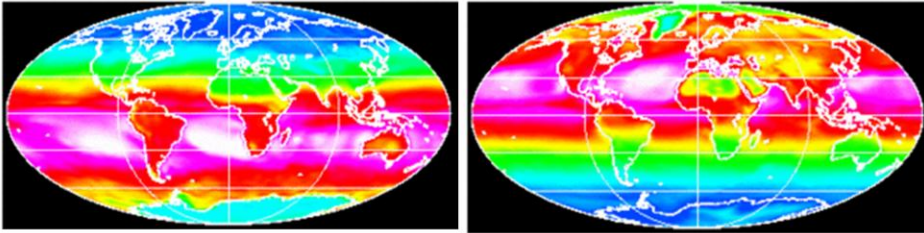
$K\uparrow$ = changes in albedo in atmosphere

$L\downarrow$ = GHGs

$L\uparrow$ =

all units in $W\ m^{-2}$

R_n at earth's surface



Source: NASA Surface Radiation Budget Project)

Latitudinal differences driven by differences
in shortwave solar input (sun angle)

http://www.eoearth.org/article/Energy_balance_of_Earth

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What happens to R_n ?

Energy is absorbed:

- Surface materials and atmosphere both **absorb energy**
- Molecules convert radiation from one form into another
- Absorbing surface re-radiates longwave radiation
- Absorbed (mostly shortwave) radiation is converted to **heat** or to chemical energy

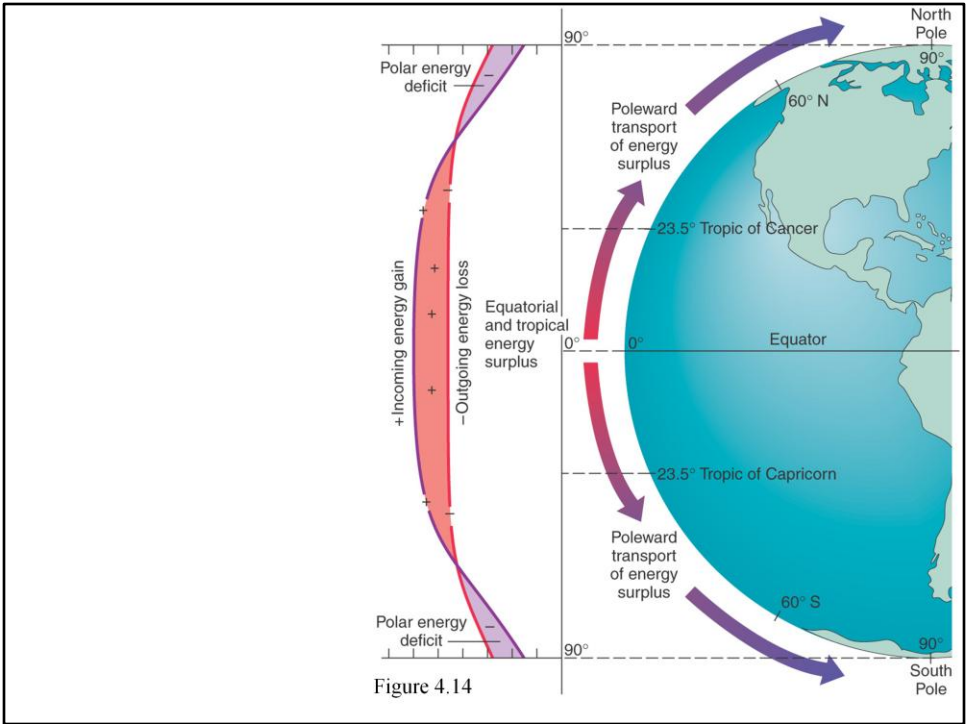


Figure 4.14

Temperature: measure of the average kinetic energy (motion) of individual molecules in matter

Heat: Form of energy that flows from one object (system) to another along a gradient

→ we *feel* the effect of temperature due to the transfer of sensible heat to our bodies

Heat Transfer

Radiation is 1 of 3 means by which energy is transferred.

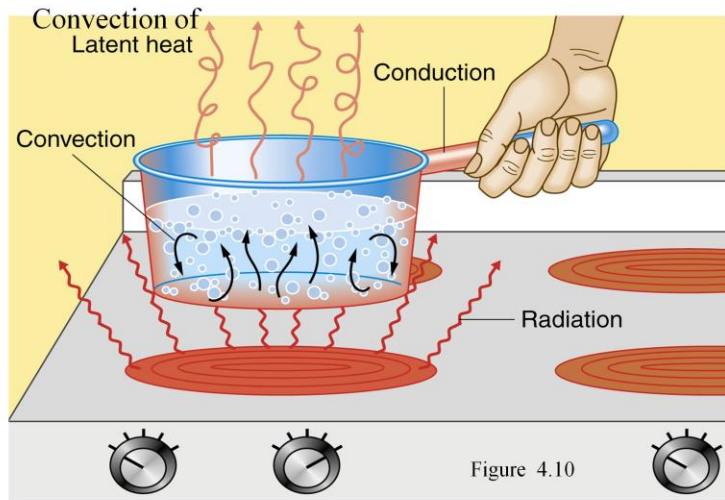
Radiation: Energy traveling through air or space (due to rapid oscillations of electromagnetic fields)

Conduction: Molecule to molecule transfer

Convection: Energy transferred by mass flow (of water, air)

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Heat Transfer



Convective energy transfer:

Horizontal → Advection

Vertical → Convection

- Sensible heat flux (H):

Flow of heat between the surface & atm. by convection.

- Latent heat flux (LE):

Flow of latent heat between the surface & atm. by convection. ie. evaporation/condensation

Conductive energy transfer:

- Ground heat flux (G):

Flow of heat into/out of ground by conduction.

**Important
elements of
the surface
energy
budget**

The physics of evaporation

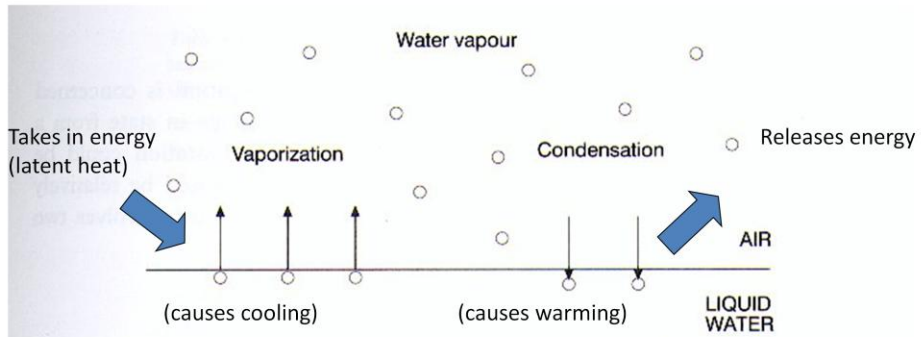


Figure 4.1 Evaporation is the net balance between the rate of vaporization of water molecules into the atmosphere, and the condensation rate of molecules from the atmosphere.

Surface Energy Budget

$$\text{Input} = \text{Outputs} + \text{Storage} \quad (\text{W m}^{-2})$$

$$R_n = H + LE + G + S$$

Type of energy transfer

R_n = net radiation (NET R)

H = sensible heat flux

LE = latent heat flux

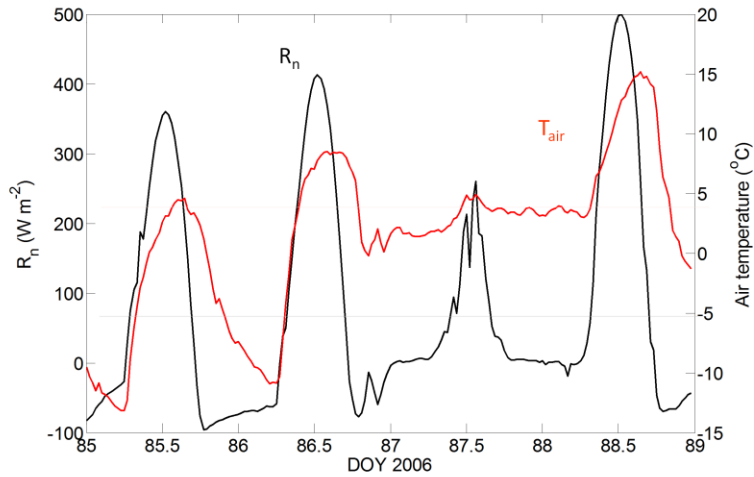
G = ground heat flux (or M)

S = change in stored energy



Why is there a temperature lag between peak net radiation and air temperature?

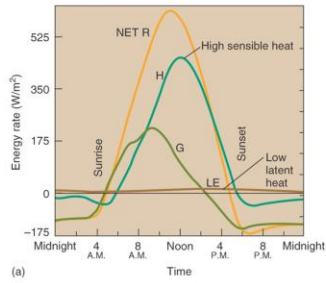
Mer Bleue bog, March 26 - March 29, 2006



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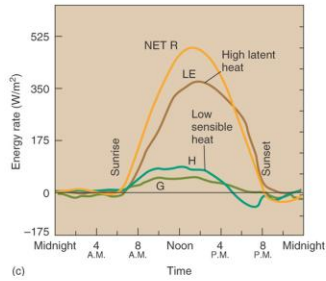
Contrasting Surface Energy Budgets

El Mirage, CA:
hot, dry



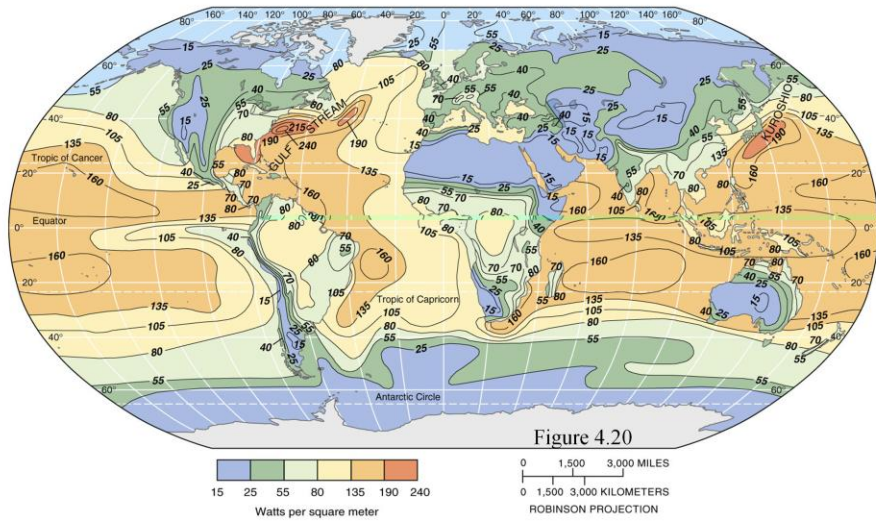
(b)

Pitt Meadows, BC: moist, with
moderate
temperatures



(d)

Global Energy Fluxes: Latent Heat



Global Energy Fluxes: Sensible Heat

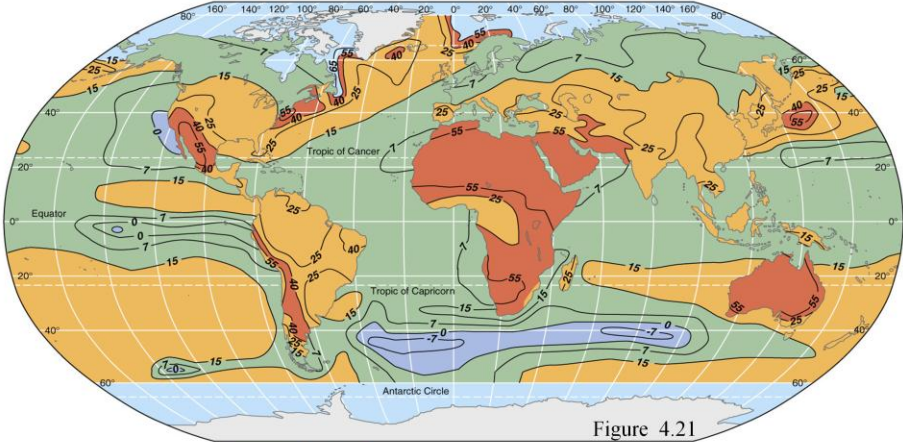
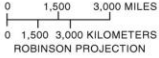
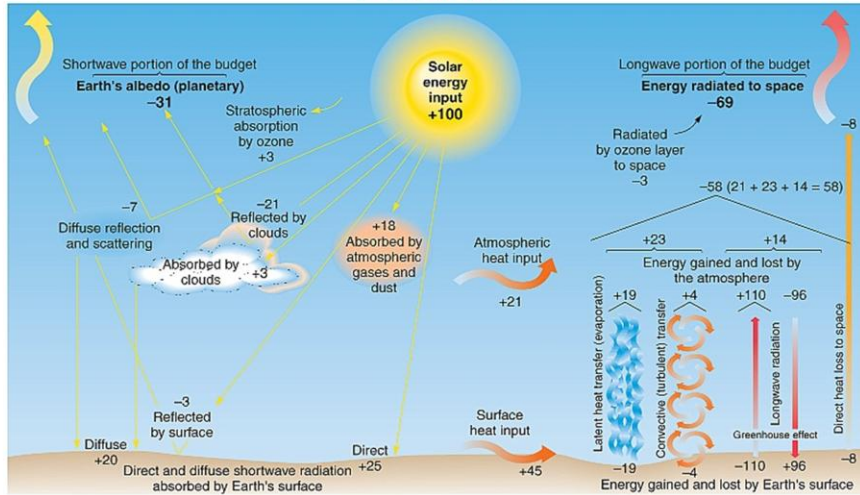


Figure 4.21



Earth-Atmosphere Radiation/Energy Balance

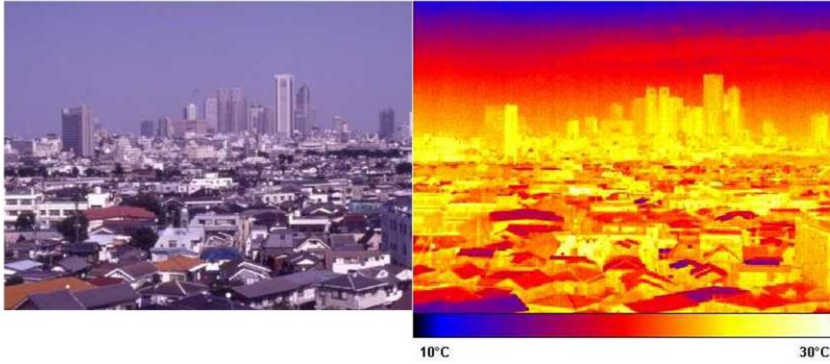


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$$\text{Inputs (K)} = \text{Outputs (K + L)}$$

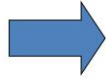
$$100 = 31 + 69$$

Urban microclimate: Urban heat island



Tokyo, Japan - M. Roth (National University of Singapore)

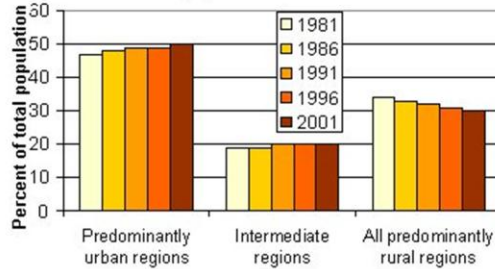
Cities are dramatic examples of the anthropogenic impact on terrestrial ecosystems and surface climates.



Urbanization & Microclimatology

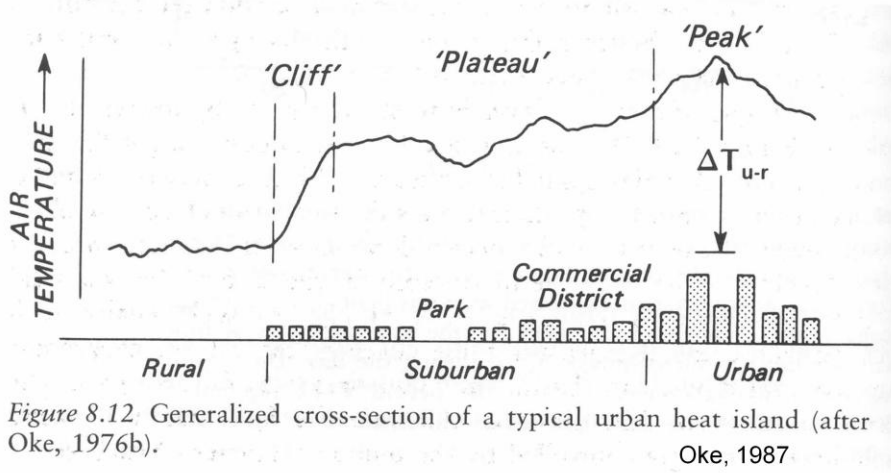
- 70% of Canadian population lives in urban areas (2011)
- 50% of global population lives in urban areas (2007, where urban is city with > 20 000 inhabitants)
- By 2025, 68% of global population is expected to live in urban areas. <http://www.unhabitat.org/>

Figure 1. The population of rural regions in Canada has steadily declined as a share of total population since 1981

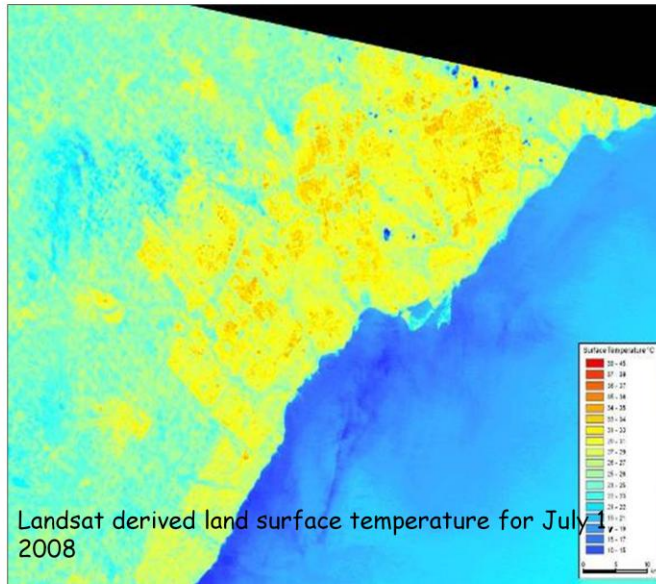


Source: Statistics Canada. Census of Population, 1981 - 2001.

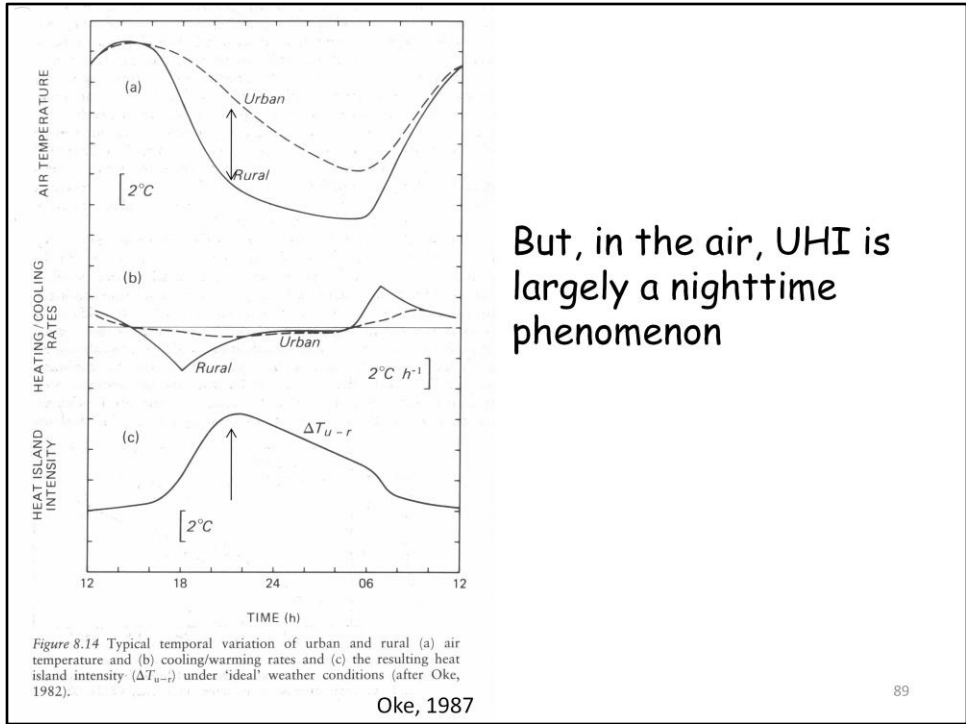
Urban Heat Island



Urban Heat Island (surface level): GTA



M. Maloley 2010



What causes the
urban heat island
effect?



What causes urban heat island effect?

Altered energy budget
terms leading to +ive ΔT_{u-r}

Related features of
urbanization

1. Increased absorption of short-wave radiation	Canyon geometry - increased surface area and multiple reflections
2. Increased long-wave radiation from sky	Air pollution - greater absorption and re-emission
3. Decreased long-wave radiation loss	Canyon geometry - reductions of sky view factor
	Building and traffic heat loss
	Construction materials - increased impermeability- remove H ₂ O storage
	Construction materials - increased thermal admittance
	Canyon geometry - reduction of wind speed

Oke 1987

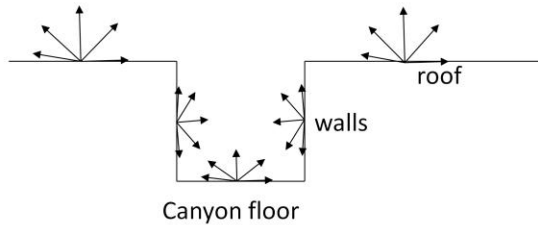
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Long-wave radiation balance

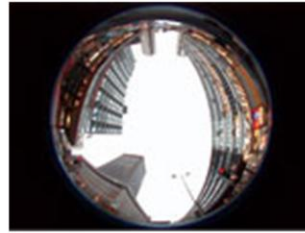
Restricted sky view factor (the amount of sky 'seen' by the surface)

ie. much of the L emitted is absorbed & re-emitted to the surface.

Reduces total *outgoing longwave radiation*



Important factor limiting radiative cooling at night!



How can we limit the urban heat island effect?

Altered energy budget
terms leading to +ive ΔT_{u-r}

Related features of
urbanization

- | | |
|-------------------------------------------------|-------------------------------------------------------------------|
| 1. Increased absorption of short-wave radiation | Canyon geometry - increased surface area and multiple reflections |
| 2. Increased long-wave radiation from sky | Air pollution - greater absorption and re-emission |
| 3. Decreased long-wave radiation loss | Canyon geometry - reductions of sky view factor |
| 4. Anthropogenic heat source | Building and traffic heat loss |
| 5. Decreased evapotranspiration | Construction materials - increased impermeability |
| 6. Increased heat storage | Construction materials - increased thermal admittance |
| 7. Decreased total turbulent heat transport | Canyon geometry - reduction of wind speed |

Oke 1987

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