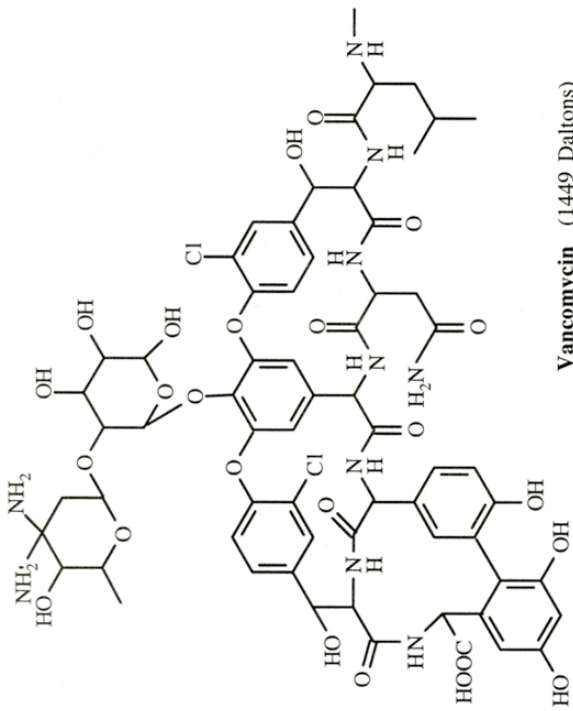
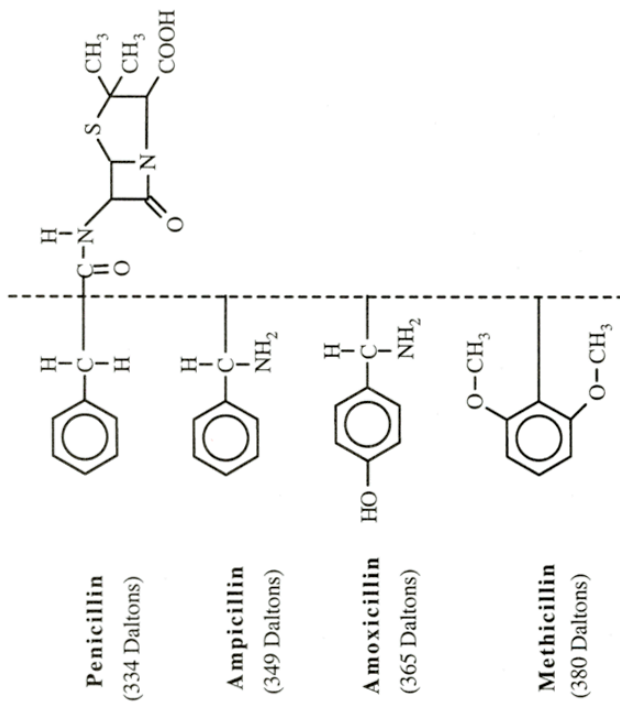
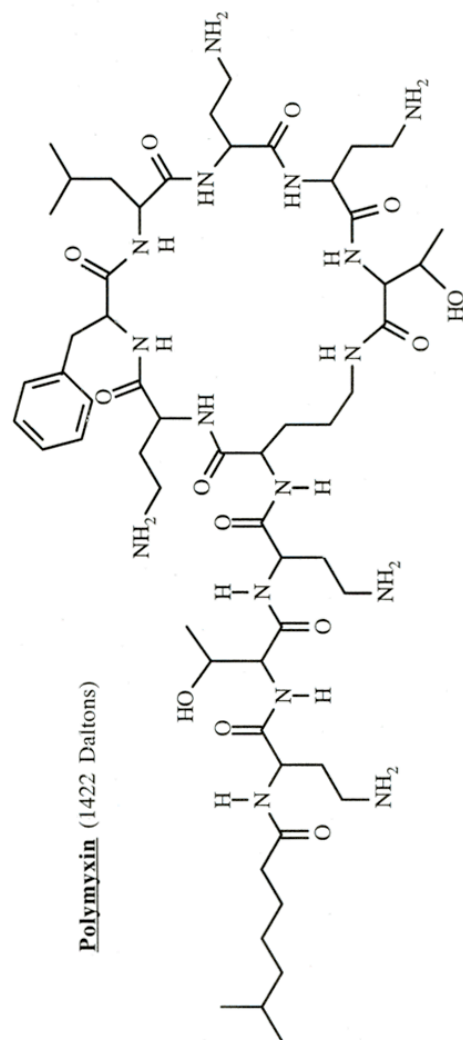


# REFERENCE MATERIAL FOR THE MICB 201 FINAL EXAM

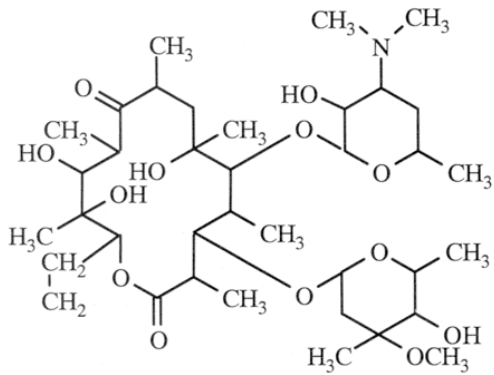
## β-lactam antibiotics



## Cephalosporin antibiotics

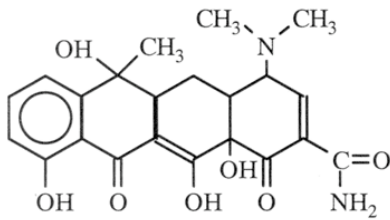


### Macrolide antibiotics



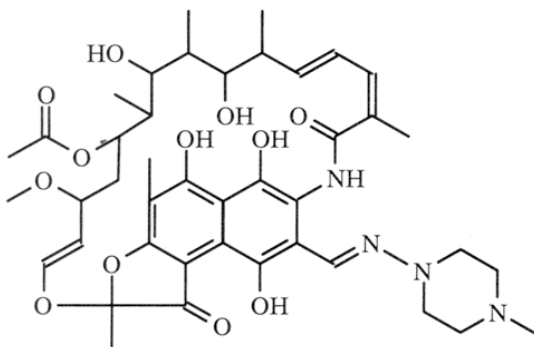
**Erythromycin** (733 Daltons)

### Tetracycline antibiotics

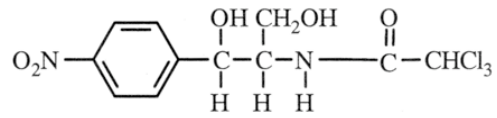


**Tetracycline** (444 Daltons)

### Rifamycin antibiotics

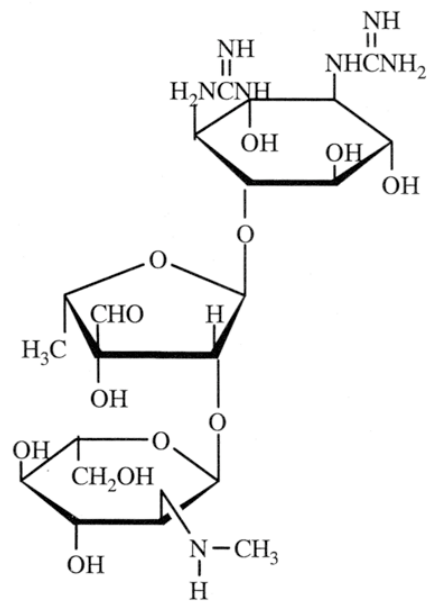


**Rifampin** (822 Daltons)



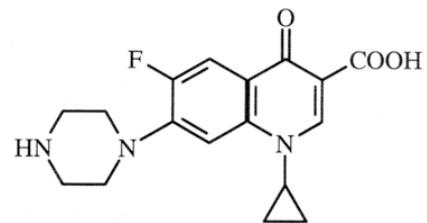
**Chloramphenicol** (323 Daltons)

### Aminoglycoside antibiotics



**Streptomycin** (581 Daltons)

### Quinolone antibiotics



**Ciprofloxacin** (331 Daltons)

<b>Antibiotic class</b>	<b>Producing Organism Domain Specifics</b>	<b>Affects</b>	<b>Mode of action</b>	<b>Side effects</b>	<b>Other comments</b>
<b><math>\beta</math>-lactams</b>	Eukarya Filamentous fungus ( <u>Penicillium</u> sp.) Bacteria <u>Streptomyces</u> sp.	Bacteria	Binds to and inhibits the transpeptidase enzyme involved in PG-X-linking, disrupting cell wall synthesis	Allergic reactions	Most important class of antibiotics clinically

Examples:

Penicillin: Natural antibiotic

Ampicillin, amoxicillin, methicillin: Chemically modified (= semi synthetic) penicillins

<b>Antibiotic class</b>	<b>Producing Organism Domain Specifics</b>	<b>Affects</b>	<b>Mode of action</b>	<b>Side effects</b>	<b>Other comments</b>
<b>Cephalosporins</b>	Eukarya Filamentous fungus ( <u>Cephalosporium</u> sp.)	Bacteria	Binds to and inhibits the transpeptidase enzyme involved in PG-X-linking, disrupting cell wall synthesis	Allergic reactions	Very important class of antibiotics

Example:

Ceftriaxone: Chemically modified form of a natural cephalosporin antibiotic

<b>Antibiotic</b>	<b>Producing Organism Domain Specifics</b>	<b>Affects</b>	<b>Mode of action</b>	<b>Side effects</b>	<b>Other comments</b>
<b>Vancomycin</b>	Bacteria A species of <u>Streptomyces</u> a member of the "Actinomycetes" (filamentous Gram +ve bacteria)	Bacteria	Binds to amino acid portion of the the disaccharide pentapeptide unit blocking the transpeptidase X-linking reaction, disrupting cell wall synthesis	Serious side effects are rare	

<u>Antibiotic class</u>	<u>Producing Organism Domain Specifics</u>	<u>Affects</u>	<u>Mode of action</u>	<u>Side effects</u>	<u>Other comments</u>
<b>Tetracyclines</b>	Bacteria <u>Streptomyces</u> , see vancomycin above.	Bacteria	Binds to the 30S ribosomal subunit, disrupting translation	Not serious in low doses	Primary use is to eliminate the bacterium that causes acne ( <u>Propionibacterium acne</u> )

<u>Antibiotic class</u>	<u>Producing Organism Domain Specifics</u>	<u>Affects</u>	<u>Mode of action</u>	<u>Side effects</u>	<u>Other comments</u>
<b>Aminoglycosides</b>	Bacteria Species of the genus <u>Streptomyces</u> , see vancomycin above.	Bacteria	Binds to the 30S ribosomal subunit, disrupting translation	Serious	Last resort antibiotics when less toxic ones fail

Example:

Streptomycin: One of the first natural antibiotics discovered.  
Still used to treat tuberculosis (Mycobacterium tuberculosis) even though side effects are serious

<u>Antibiotic class</u>	<u>Producing Organism Domain Specifics</u>	<u>Affects</u>	<u>Mode of action</u>	<u>Side effects</u>	<u>Other comments</u>
<b>Macrolides</b>	Bacteria Species of the genus <u>Streptomyces</u> , see vancomycin above.	Bacteria	Binds to the 50S ribosomal subunit, disrupting translation	Not serious	Account for 10% of the antibiotics produced and used

Example:

Erythromycin: Natural antibiotic  
Used when a patient is allergic to the penicillin antibiotics

<b>Antibiotic</b>	<b>Producing Organism Domain Specifics</b>	<b>Affects</b>	<b>Mode of action</b>	<b>Side effects</b>	<b>Other comments</b>
<b>Chloramphenicol</b>	Bacteria A species of Streptomycetes, see vancomycin above	Bacteria	Binds to the 50S ribosomal subunit, disrupting translation	Serious	Last resort antibiotics when less toxic ones fail

<b>Antibiotic class</b>	<b>Producing Organism Domain Specifics</b>	<b>Affects</b>	<b>Mode of action</b>	<b>Side effects</b>	<b>Other comments</b>
<b>Rifamycins</b>	Bacteria Species of the genus Streptomycetes, see vancomycin above.	Bacteria	Binds to and inhibits RNA polymerase, disrupting transcription	Potential liver damage	

Example: Rifampin-Used to treat tuberculosis (*Mycobacterium tuberculosis*) despite risk of liver damage.

<b>Antibiotic class</b>	<b>Producing Organism Domain Specifics</b>	<b>Affects</b>	<b>Mode of action</b>	<b>Side effects</b>	<b>Other comments</b>
<b>Quinolone</b>	Not applicable-synthetic molecules	Bacteria	Binds to and inhibits an enzyme involved in DNA replication	Serious side effects are rare	

Example: Ciprofloxacin-Used to treated anthrax (*Bacillus anthracis*)

<b>Antibiotic</b>	<b>Producing Organism Domain Specifics</b>	<b>Affects</b>	<b>Mode of action</b>	<b>Side effects</b>	<b>Other comments</b>
<b>Polymyxin</b>	Bacteria <i>Bacillus polymyxa</i> , a Gram +ve, endospore-forming rod	Bacteria	Binds to LPS of OM, then somehow leads to disruption of the CM	Nervous system kidney damage	Last resort antibiotic when less toxic ones fail. Used in antibiotic ointments ("Polysporin") & eye drops

<b>Group</b>	<b>Growth in relation to temperature</b>	<b>Some habitats</b>
<b>Psychrophiles</b>	Growth between ~ 0-15°C Some grow at < 0°C where elevated solute concentration keeps water from freezing	The ocean (most ocean water never gets much above 5°C) Snow, glacial streams and run-off Under arctic/antarctic ice and icebergs
<b>Mesophiles</b>	Growth between ~15°C-50°C	Ubiquitous in soil and water and on plant surfaces in temperate and tropical latitudes Bodies of warm blooded animals
<b>Thermophiles</b>	Growth between ~50°C-80°C	Spill-over areas from hot springs Surface soils which are exposed to direct sunlight; temperatures can reach 50-70°C even in northern latitudes. Soil thermophiles that find their way into hot water heaters, hot tubs/saunas. Compost piles (they get quite hot, 70°C) during decomposition of organic matter
<b>Extreme (hyper) thermophiles</b>	Growth between ~80°C-100°C Some grow at > 100°C where elevated pressure keeps water from boiling	Geothermally heated mud pots, hot springs; temperatures at the boiling point of water Hydrothermal vent areas under pressure on ocean floor

<b>Group</b>	<b>Growth in relation to pH</b>	<b>Some habitats</b>
<b>Extreme acidophiles</b>	< 2	Sulfurous volcanic soils, volcanic hot springs
<b>Acidophiles</b>	~2-5	Stomachs of animals, volcanic soils
<b>Neutralophiles</b>	~5-8	Ubiquitous
<b>Alkaliphiles</b>	~8-11	Ubiquitous in soils
<b>Extreme alkaliphiles</b>	~11-14	Alkaline salt lakes

<u>Group</u>	<u>Growth in relation to [NaCl]</u>	<u>Some habitats</u>
<b>Nonhalophiles</b>	Do not require NaCl and grow optimally at [NaCl] less than that of seawater ~3% w/v NaCl	Ubiquitous
<b>Moderate halophiles</b>	Require NaCl and grow optimally near the [NaCl] characteristic of seawater ~3% w/v NaCl	<b>Oceans:</b> the [NaCl] in seawater is about 3% w/v NaCl (other dissolved salts are also present)
<b>Extreme halophiles</b>	Require a minimum concentration of ~ 10 % w/v NaCl in the environment and grow optimally at concentrations of NaCl characteristic of a saturated salt solution.	<b>Hypersaline lakes:</b> [NaCl] ~ 25-30 % w/v NaCl. Example: the <b>Dead Sea</b> (pH ~ 6, so not an alkaline salt lake like the Great Salt Lake)  Salt ponds
<b>Halotolerant nonhalophiles</b>	Do not require NaCl. Optimal growth < 3% w/v NaCl but still significant growth rates up to 10% w/v NaCl.	Ubiquitous Example: <i>S. aureus</i> and other skin bacteria. Skin is a salty environment; [NaCl] fluctuates with sweating

<b>Organism</b>		<b>Glucose-metabolizing pathway</b>
Bacteria	Some	Glycolysis (Embden-Myerhoff, EM pathway)
	Some	Entner-Doudorff (ED) pathway
	Some	Other
Archaea	Some	Archaeal variant of the EM pathway
	Some	Archaeal variant of the ED pathway
	Some	Archaeal variants of both the EM and ED pathways

<b>Bacteria</b>	<b>Reaction pathways</b>
Cyanobacteria	Calvin cycle
Purple bacterial lithoautotrophs	Calvin cycle
Purple bacterial photoautotrophs	Calvin cycle
Green sulfur bacteria	Reverse TCA cycle
Green nonsulfur bacteria	Hydroxypropionate pathway
Lithoautotrophic Planctomyces	Acetyl CoA pathway

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<u>TEA</u>								
<u>ENERGY SOURCE</u>	O <sub>2</sub>	NO <sub>3</sub> <sup>-</sup>	NO <sub>2</sub> <sup>-</sup>	Fe <sup>+3</sup>	SO <sub>4</sub> <sup>-2</sup>	S <sup>0</sup>	CO <sub>2</sub>	No TEA
<b><u>Organic</u></b>								
CHX	+	+	+	+	+	+	+	+
Special case: CH <sub>4</sub>	+	+	-	?	+	?	-	-
<b><u>Inorganic</u></b>								
H <sub>2</sub>	+	+	+	+	+	+	+	-
H <sub>2</sub> S	+	+	-	-	-	-	-	-
S <sup>0</sup>	+	+	-	-	-	-	-	-
Fe <sup>+2</sup>	+	+	-	-	-	-	-	-
NH <sub>4</sub> <sup>+</sup> (NH <sub>3</sub> )	+	-	+	-	-	-	-	-
NO <sub>2</sub> <sup>-</sup>	+	-	-	-	-	-	-	-
<b><u>Light</u></b>	NR	NR	NR	NR	NR	NR	NR	NR

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<u>Type</u>	<u>Group</u>	<u>Photosystem(s)</u>	<u>Electron sources</u>
<b>Type I</b>	Green sulfur bacteria	PS I-bacteriochlorophyll	H <sub>2</sub> , H <sub>2</sub> S
	Cyanobacteria	PS I-bacteriochlorophyll	H <sub>2</sub> S
	Gram-positive	PS I-bacteriochlorophyll	See below, Figure 11-7.
<b>Type II</b>	Purple bacteria	PSII-bacteriochlorophyll	H <sub>2</sub> , H <sub>2</sub> S, Fe <sup>+2</sup> , NO <sub>2</sub> <sup>-</sup> , CHX <sub>Red</sub>
	Green nonsulfur bacteria*	PSII-bacteriochlorophyll	H <sub>2</sub> , H <sub>2</sub> S, CHX <sub>Red</sub>
<b>Type I/ Type II</b>	Cyanobacteria	PSI and PSII-chlorophyll	H <sub>2</sub> O
	Chloroplasts in Eukarya	PSI and PSII-chlorophyll	H <sub>2</sub> O

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<u>Energy Source/TEA</u>	<u>Waste</u>	<u>Electrons transferred</u>	<u>Energy produced (kJ/reaction standard conditions)</u>	<u>Energy produced/2e<sup>-</sup></u>
<u>C</u> H <sub>4</sub> + 2 O <sub>2</sub>	→ CO <sub>2</sub> + 2 H <sub>2</sub> O	8	816	204
<u>H</u> <sub>2</sub> + 1/2 O <sub>2</sub>	→ H <sub>2</sub> O	2	237	237
H <sub>2</sub> <u>S</u> + 2 O <sub>2</sub>	→ SO <sub>4</sub> <sup>-2</sup> + 2 H <sup>+</sup>	8	800	200
<u>N</u> H <sub>4</sub> <sup>+</sup> + 2 O <sub>2</sub>	→ NO <sub>3</sub> <sup>-</sup> + H <sub>2</sub> O + 2 H <sup>+</sup>	8	349	87
2 <u>Fe</u> <sup>+2</sup> + 2 H <sup>+</sup> + 1/2 O <sub>2</sub>	→ 2 Fe <sup>+3</sup> + H <sub>2</sub> O	2	66	66

<u>Reactions</u>	<u>e<sup>-</sup> transferred</u>	<u>Energy produced (kJ/reaction, standard conditions)</u>	<u>Energy produced/2e<sup>-</sup></u>
<u>Energy Source/TEA</u>	<u>Waste</u>		
H <sub>2</sub> + 1/2 O <sub>2</sub>	→ H <sub>2</sub> O	2	237
H <sub>2</sub> + 2 Fe <sup>+3</sup>	→ 2 Fe <sup>+2</sup> + 2 H <sup>+</sup>	2	220
H <sub>2</sub> + NO <sub>3</sub> <sup>-</sup>	→ NO <sub>2</sub> <sup>-</sup> + H <sub>2</sub> O	2	163
4 H <sub>2</sub> + SO <sub>4</sub> <sup>-2</sup> + 2 H <sup>+</sup>	→ H <sub>2</sub> S + 4 H <sub>2</sub> O	8	152
4 H <sub>2</sub> + CO <sub>2</sub>	→ CH <sub>4</sub> + 2 H <sub>2</sub> O	8	131

<u>Reactions</u>	<u>e<sup>-</sup> transferred</u>	<u>Energy produced (kJ/reaction, standard conditions)</u>	<u>Energy produced/2e<sup>-</sup></u>
<u>Energy Source/TEA</u>	<u>Products</u>		
H <sub>2</sub> + NO <sub>3</sub> <sup>-</sup>	→ NO <sub>2</sub> <sup>-</sup> + H <sub>2</sub> O	2	163
5 H <sub>2</sub> + 2NO <sub>3</sub> <sup>-</sup> + 2 H <sup>+</sup>	→ N <sub>2</sub> + 6 H <sub>2</sub> O	10	1120
3 H <sub>2</sub> + SO <sub>4</sub> <sup>-2</sup> + 2 H <sup>+</sup>	→ S <sup>0</sup> + 4 H <sub>2</sub> O	6	124
4 H <sub>2</sub> + SO <sub>4</sub> <sup>-2</sup> + 2 H <sup>+</sup>	→ H <sub>2</sub> S + 4 H <sub>2</sub> O	8	152