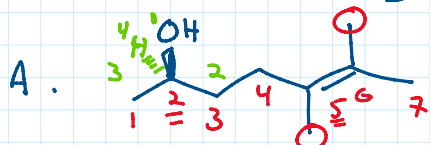
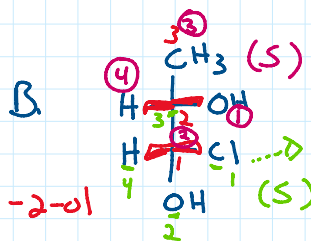


Workshop problems:

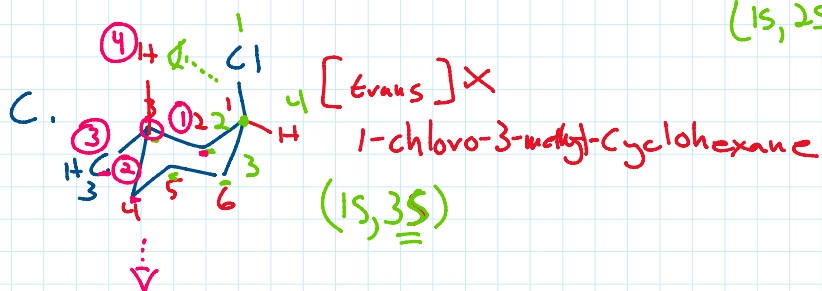
1. Name the following



(2E)-5,6-dimethylhept-5-en-2-ol

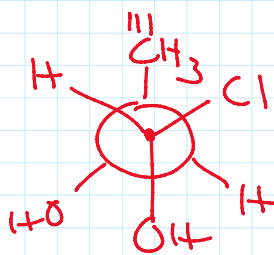
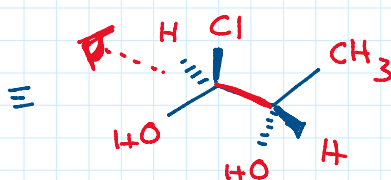
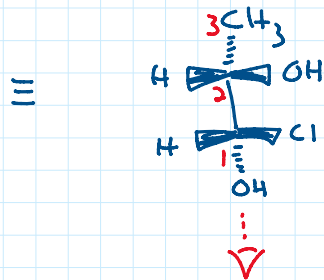
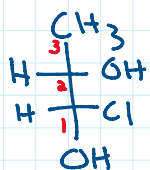


(1S,2S)-1-chloropropan-1,2-diol

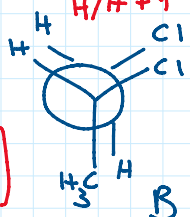
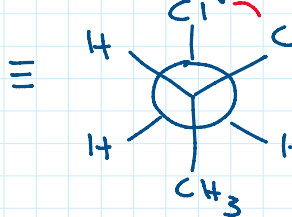
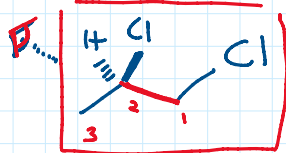


2. Draw a Newman projection along ~~C1-C2~~ bond of:

C1-C2



3. Draw Newman projection & calc energies



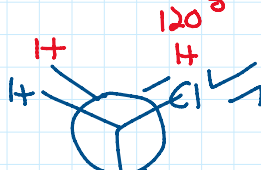
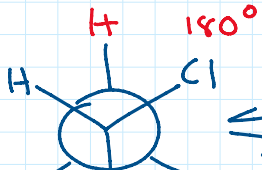
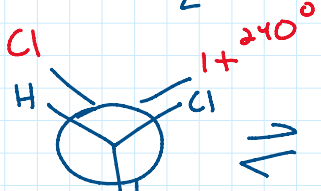
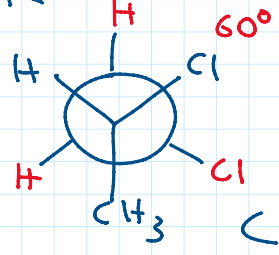
Cl/Cl +28

+38 kJ/mol

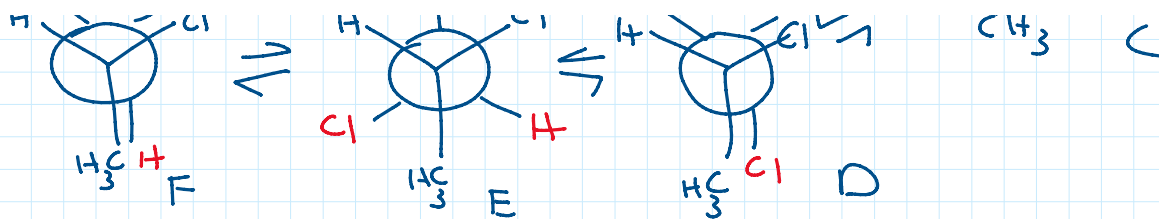
+2.5 kJ/mol

H/CH3 +6

H/H +4



Atoms	Interaction	Energy Penalty [kJ/mol]
H/H	Eclipsed	4.0
H/CH ₃	Eclipsed	6.0
H/CH ₃	Gauche	0
CH ₃ /CH ₃	Eclipsed	11.0
CH ₃ /CH ₃	Gauche	3.8
Cl/H	Eclipsed	8.3
Cl/CH ₃	Eclipsed	7.0
Cl/CH ₃	Gauche	1.2
Cl/Cl	Gauche	2.5
Cl/Cl	Eclipsed	28



A. $2.5 = +2.5 \text{ kJ/mol}$ 300°
 $\frac{\text{Cl/Cl}}{\text{S}}$

E. $1.2 = +1.2 \text{ kJ/mol}$ 180°
 $\frac{\text{Cl/CH}_3}{\text{S}}$

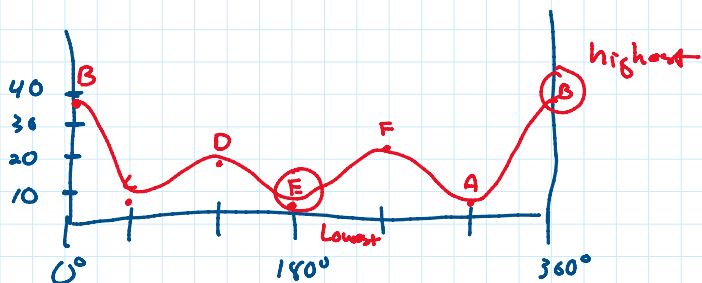
B. $28 + 4.0 + 6.0 = +38 \text{ kJ/mol}$ 0°
 $\frac{\text{Cl/Cl}}{\text{e}} \quad \frac{\text{H/H}}{\text{e}} \quad \frac{\text{H/CH}_3}{\text{e}}$

F. $6.0 + 8.3 + 8.3 = +22.6 \text{ kJ/mol}$ 240°
 $\frac{\text{CH}_3/\text{H}}{\text{e}} \quad \frac{\text{Cl/H}}{\text{e}} \quad \frac{\text{Cl/H}}{\text{e}}$

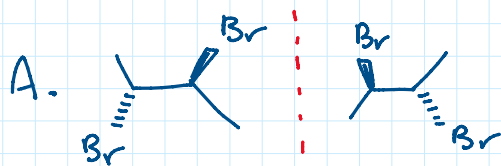
C. $2.5 + 1.2 = +3.7 \text{ kJ/mol}$ 60°
 $\frac{\text{Cl/Cl}}{\text{S}} \quad \frac{\text{Cl/Me}}{\text{S}}$

D. $7.0 + 4.0 + 8.3 = +19.3 \text{ kJ/mol}$ 120°
 $\frac{\text{Cl/CH}_3}{\text{e}} \quad \frac{\text{H/H}}{\text{e}} \quad \frac{\text{Cl/H}}{\text{e}}$

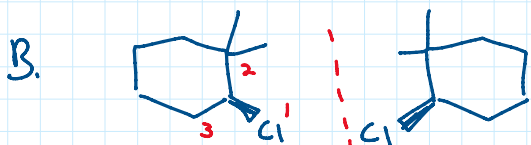
A	+2.5	300°	D	+19.3	120°
B	+38	0°	E	+1.2	180°
C	+3.7	60°	F	+22.6	240°



4. What is the relationship b/w the following?

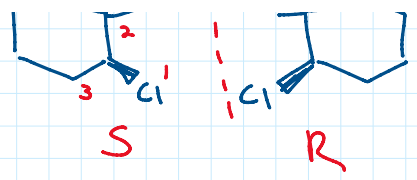


~~Enantiomer~~
Same Meso
 plane of Symmetry



enantiomers!

15.



enantiomers!