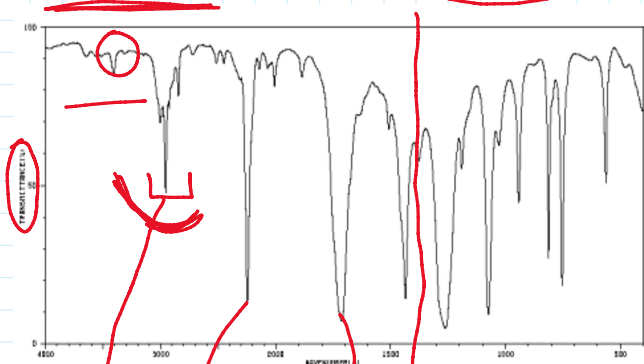
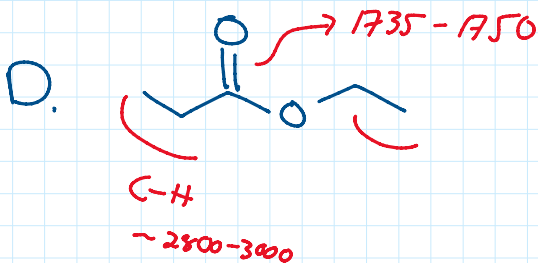
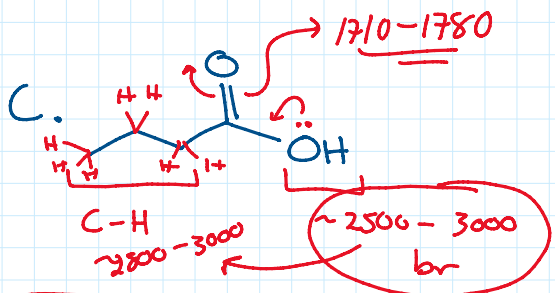
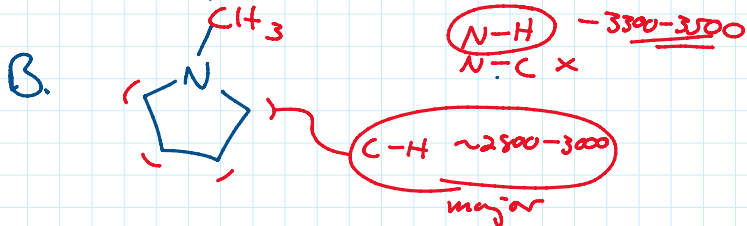
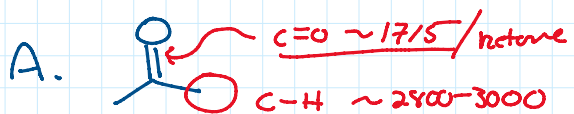
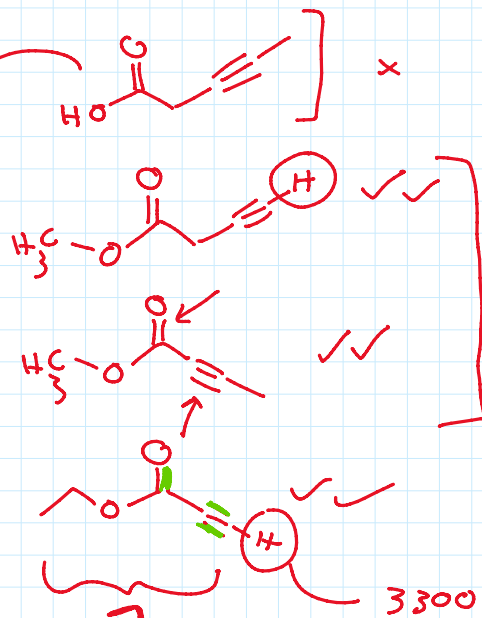
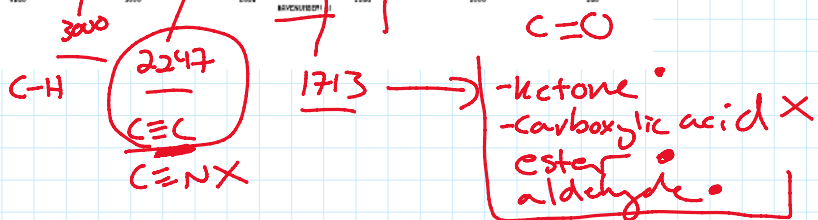


See posted WS1 problems

1. Predict the major IR absorption bands



2. Propose a structure for a compd w/ the formula: $C_5H_6O_2$ & the IR spec. shown.



$$DoU = \frac{2C + 2 + N - H - X}{2}$$

Degree of unsaturation

$$\frac{2 \cdot 5 + 2 - 6}{2}$$

$$\frac{10 + 2 - 6}{2} = \frac{6}{2} = 3$$

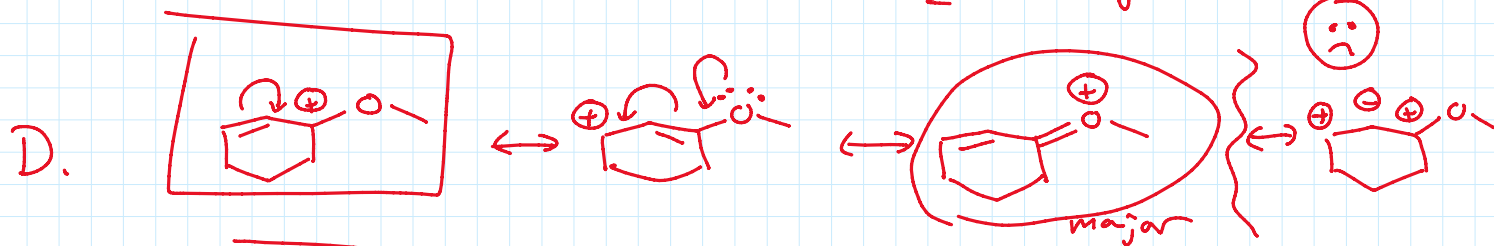
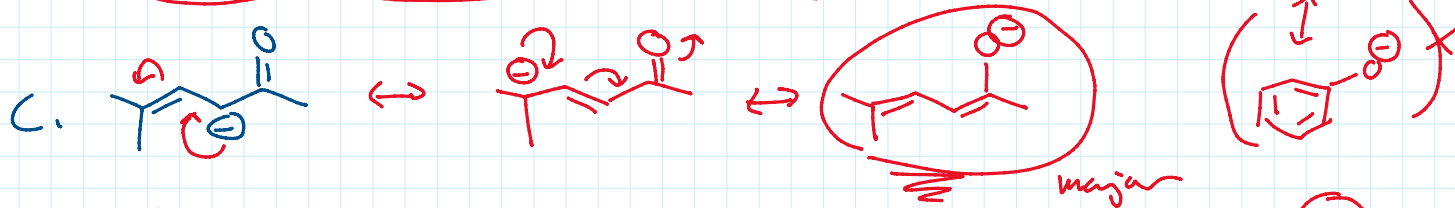
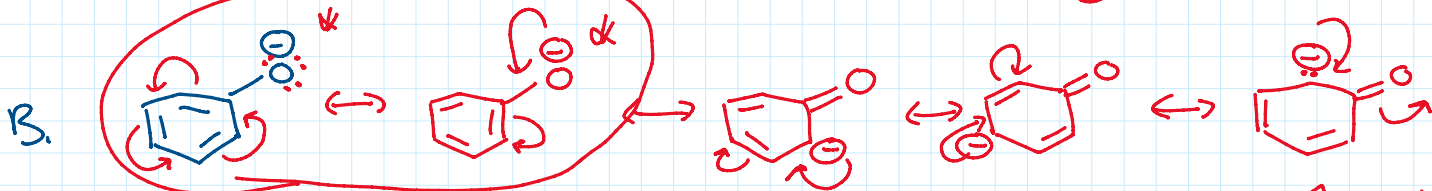
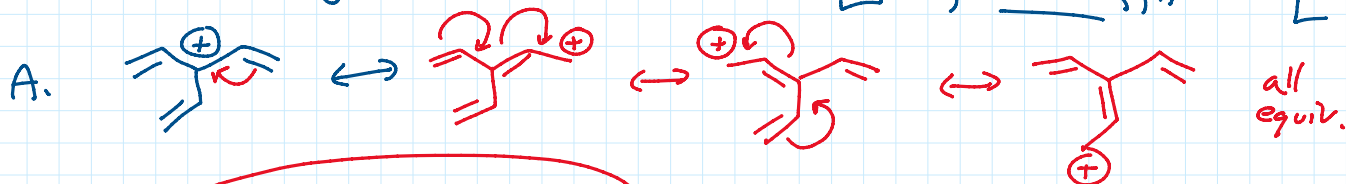
- DoU
- 0 H3C-CH3
 - 1 H2C=CH2
 - 2 H-C\equiv C-H

valid • Conserve charge
 • No atoms, single bonds
 major/minor
 • # cov. bonds
 • filled valence



3. Draw major resonance forms

- valid
- Conserve charge
 - No atoms, single bonds
 - Obey valency rules
 - LP, non-bonding, π
- major/minor
- # cov. bonds
 - filled valence
 - sep. of charge
 - EU



- unfilled valence
- covalent bonds

- \oplus on oxygen