

**CHM 1321 (prof. S. Gambarotta) 2<sup>nd</sup> Midterm April 2– 2009**

**Your Name:** \_\_\_\_\_ **Student #:** \_\_\_\_\_

**Your course TA (Steve, Derek, Marc):** \_\_\_\_\_

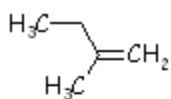
<b>Exercise</b>	<b>key</b>	<b>Exercise</b>	<b>key</b>
<b>1</b>		<b>16</b>	
<b>2</b>		<b>17</b>	
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<b>12</b>		<b>27</b>	
<b>13</b>			
<b>14</b>			
<b>15</b>			

1. Please deliver only this page. You can keep the text. Scratch paper is at the end.
2. Solution key will be posted today.
3. Marks will be posted on Virtual Campus gradebook ASAP.
4. This is an Open BOOK exam. You can use your models and textbook.

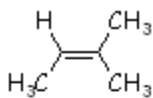
- Heating tert-butyl chloride with 1.0 M NaOH in a mixture of water and methanol would yield mainly:
  - $(\text{CH}_3)_3\text{COH}$  through an  $\text{S}_{\text{N}}1$  reaction.
  - $(\text{CH}_3)_3\text{COCH}_3$  through an  $\text{S}_{\text{N}}1$  reaction.
  - $(\text{CH}_3)_3\text{COH}$  through an  $\text{S}_{\text{N}}2$  reaction.
  - $(\text{CH}_3)_3\text{COCH}_3$  through an  $\text{S}_{\text{N}}2$  reaction.
  - $\text{CH}_2=\text{C}(\text{CH}_3)_2$  through an E2 reaction.
  
- Which of these compounds would give the largest E2/ $\text{S}_{\text{N}}2$  product ratio on reaction with sodium ethoxide in ethanol at  $55^\circ\text{C}$ ?
  - $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{Cl}$
  - $$\begin{array}{c} \text{CH}_3\text{CH}_2\text{CHCH}_2\text{CH}_2\text{Cl} \\ | \\ \text{CH}_3 \end{array}$$
  - $$\begin{array}{c} \text{CH}_3\text{CHCH}_2\text{CH}_2\text{Cl} \\ | \\ \text{CH}_3 \end{array}$$
  - $$\begin{array}{c} \text{CH}_3\text{CH}_2\text{CHCH}_2\text{CH}_3 \\ | \\ \text{Cl} \end{array}$$
  - $(\text{CH}_3)_3\text{CCH}_2\text{Cl}$
  
- Elimination reactions are favored over nucleophilic substitution reactions:
  - at high temperatures.
  - when tert-butoxide ion is used.
  - when  $3^\circ$  alkyl halides are used as substrates.
  - when nucleophiles are used which are strong bases and the substrate is a  $2^\circ$  alkyl halide.
  - in all of these cases.



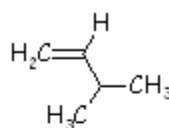
6. Neopentyl alcohol,  $(\text{CH}_3)_3\text{CCH}_2\text{OH}$ , cannot be dehydrated to an alkene without rearrangement. What is the chief product of dehydration?



I



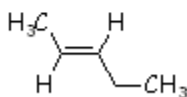
II



III



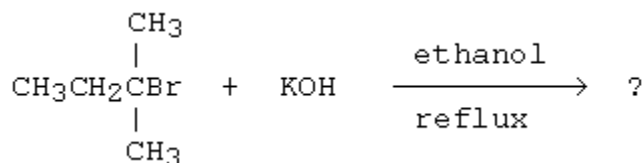
IV



V

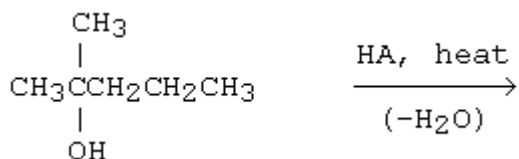
- A) I  
B) II  
C) III  
D) IV  
E) V

7. Which compound listed below would you expect to be the major product of this reaction?



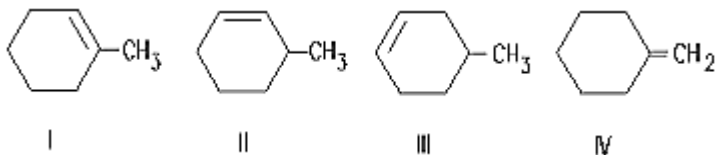
- A)  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3\text{CH}_2\text{COH} \\ | \\ \text{CH}_3 \end{array}$
- B)  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3\text{CH}_2\text{C}-\text{O}-\text{CH}_2\text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$
- C)  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3\text{CH}_2\text{C}=\text{CH}_2 \end{array}$
- D)  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3\text{CH}=\text{CCH}_3 \\ | \\ \text{CH}_3 \end{array}$
- E)  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_2=\text{CHCHCH}_3 \\ | \\ \text{CH}_3 \end{array}$

8. Which product(s) would be produced by acid-catalyzed dehydration of the following alcohol?



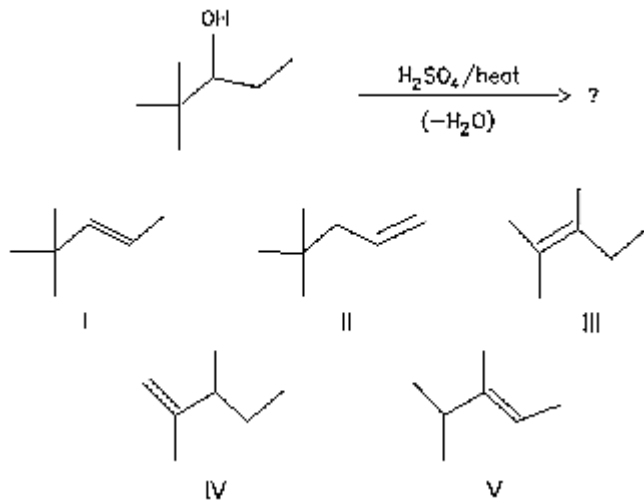
- A)  $\begin{array}{c} \text{CH}_2 \\ || \\ \text{CH}_3\text{CCH}_2\text{CH}_2\text{CH}_3 \end{array}$
- B)  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_2=\text{CCH}_2\text{CH}_2\text{CH}_3 \end{array}$  and  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3\text{C}=\text{CHCH}_2\text{CH}_3 \end{array}$
- C)  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3\text{CHCH}=\text{CHCH}_3 \end{array}$  and  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3\text{CHCH}_2\text{CH}=\text{CH}_2 \end{array}$
- D)  $\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3\text{CHCH}_2\text{CH}=\text{CH}_2 \end{array}$
- E)  $\begin{array}{c} \text{CH}_3 \quad \text{CH}_3 \\ | \quad | \\ \text{CH}_3\text{CH}_2\text{CH}_2\text{C}-\text{O}-\text{CCH}_2\text{CH}_2\text{CH}_3 \\ | \quad | \\ \text{CH}_3 \quad \text{CH}_3 \end{array}$

9. Which product (or products) would be formed in appreciable amount(s) when trans-1-bromo-2-methylcyclohexane undergoes dehydrohalogenation upon treatment with sodium ethoxide in ethanol?



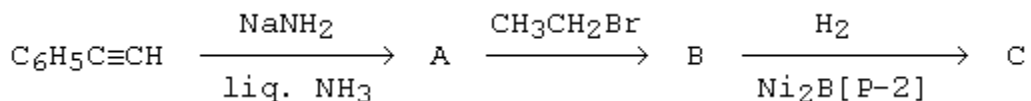
- A) I  
 B) II  
 C) III  
 D) IV  
 E) More than one of these

10. Which alkene would you expect to be the major product of the following dehydration?



- A) I  
 B) II  
 C) III  
 D) IV  
 E) V

11. The structure of the product, C, of the following sequence of reactions would be:

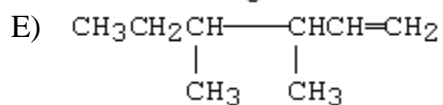
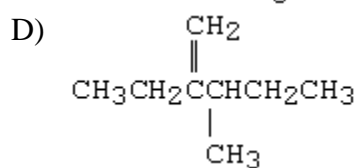
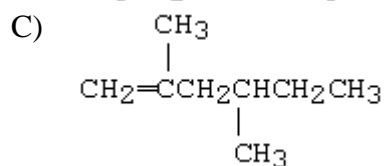
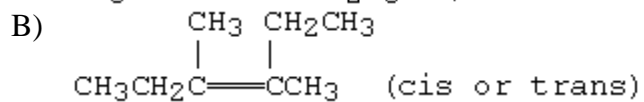
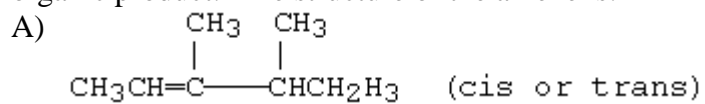


- A) cis- $\text{CH}_3\text{CH}_2\text{CH}=\text{CHC}_6\text{H}_5$   
 B) cis- $\text{CH}_3\text{CH}=\text{CHC}_6\text{H}_5$   
 C) trans- $\text{CH}_3\text{CH}_2\text{CH}=\text{CHC}_6\text{H}_5$   
 D)  $\text{C}_6\text{H}_5\text{C}\equiv\text{CCH}_2\text{CH}_2\text{Br}$   
 E)  $\text{C}_6\text{H}_5\text{C}\equiv\text{CCH}_2\text{CH}_3$

12. For which of the following is cis-trans isomerism impossible?

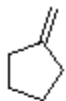
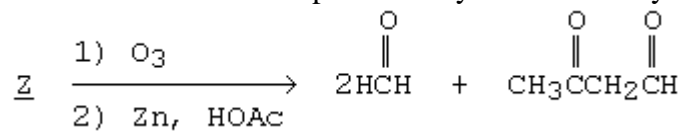
- A) 2-Hexene  
 B) 3-Methyl-2-pentene  
 C) 3-Hexene  
 D) 2-Methyl-2-butene  
 E) 2-Pentene

13. An alkene adds hydrogen in the presence of a catalyst to give 3,4-dimethylhexane. Ozonolysis of the alkene followed by treatment with zinc and acetic acid gives a single organic product. The structure of the alkene is:

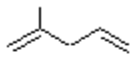


14.

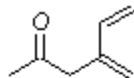
What is the structure of the compound Z that yields 2 mol of formaldehyde,  $\text{HCHO}$ , and 1 mol of  $\text{CH}_3\text{C}(=\text{O})\text{CH}_2\text{C}(=\text{O})\text{H}$  upon ozonolysis followed by treatment with zinc in acetic acid?



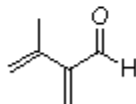
I



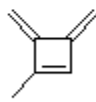
II



III



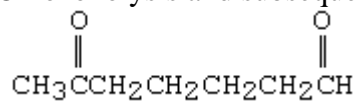
IV



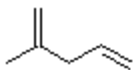
V

- A) I
- B) II
- C) III
- D) IV
- E) V

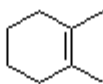
15. Compound, C, has the molecular formula  $C_7H_{12}$ . On catalytic hydrogenation, 1 mol of C absorbs 1 mol of hydrogen and yields a compound with the molecular formula  $C_7H_{14}$ . On ozonolysis and subsequent treatment with zinc and acetic acid, C yields only:



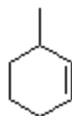
The structure of C is:



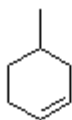
I



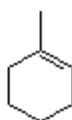
II



III



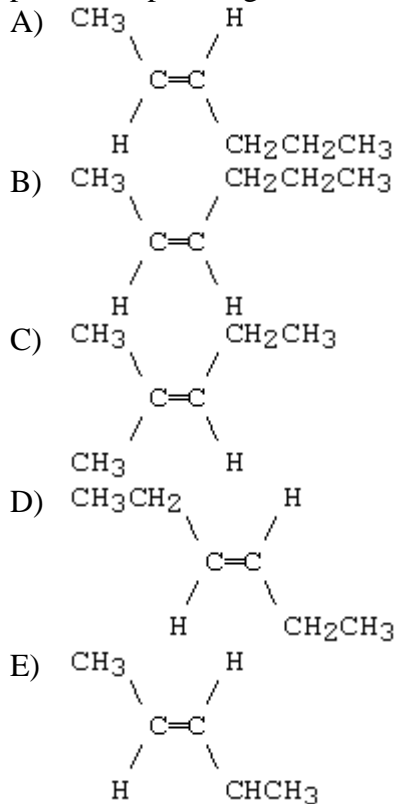
IV



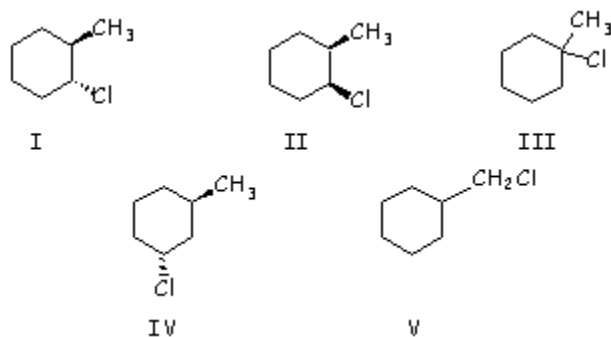
V

- A) I  
 B) II  
 C) III  
 D) IV  
 E) V

16. Which alkene would yield only  $\text{CH}_3\text{CH}_2\text{COOH}$  on oxidation with hot alkaline potassium permanganate?

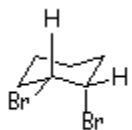


17. Treating 1-methylcyclohexene with HCl would yield primarily which of these?

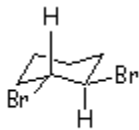


- A) I  
 B) II  
 C) III  
 D) IV  
 E) V

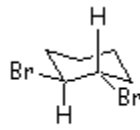
18. The addition of bromine to cyclohexene would produce the compound(s) represented by structure(s):



I



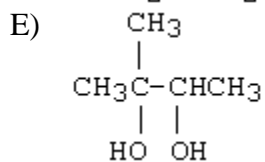
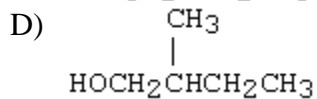
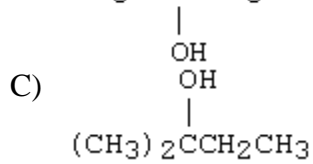
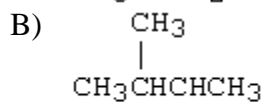
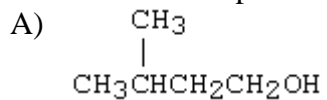
II



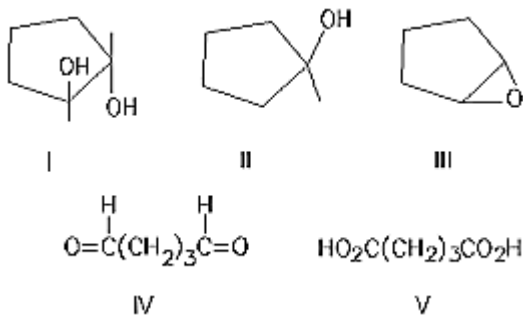
III

- A) I alone  
 B) II alone  
 C) II and III  
 D) III alone  
 E) I and II

19. What is the chief product of the acid-catalyzed hydration of 2-methyl-2-butene?

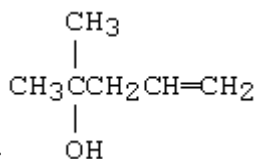


20. Which compound is reasonably anticipated as a byproduct in the hydroxylation of cyclopentene with cold alkaline permanganate?



- A) I  
 B) II  
 C) III  
 D) IV  
 E) V

21.

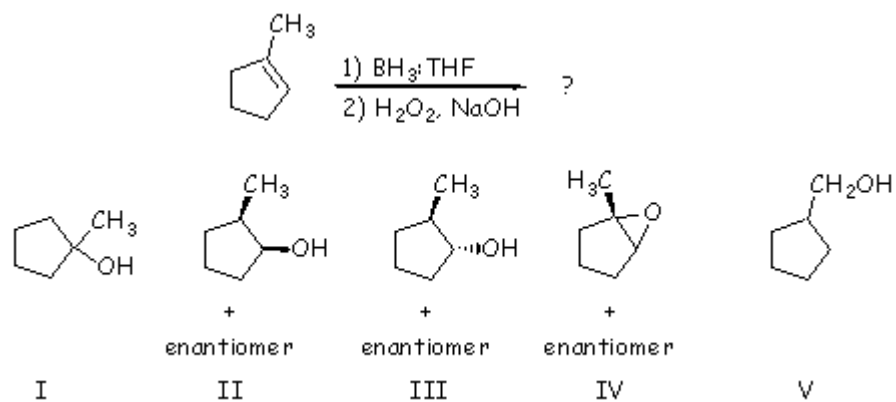


The correct IUPAC substitutive name for

is:

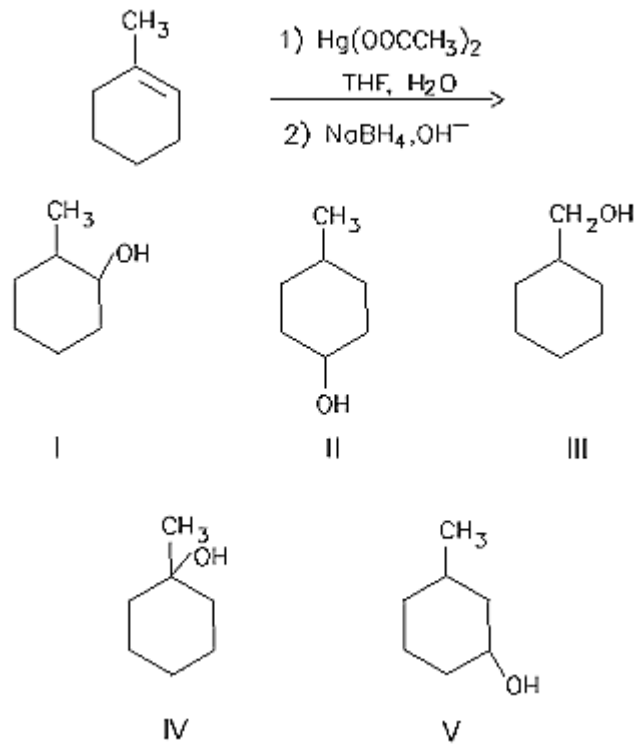
- A) 4-Penten-2-methyl-2-ol  
 B) 4-Methyl-1-penten-2-ol  
 C) 2-Methyl-4-penten-2-ol  
 D) 4-Methyl-1-penten-4-ol  
 E) 4-Hydroxy-4-methyl-1-pentene

22. Which product(s) would you expect to obtain from the following sequence of reactions?



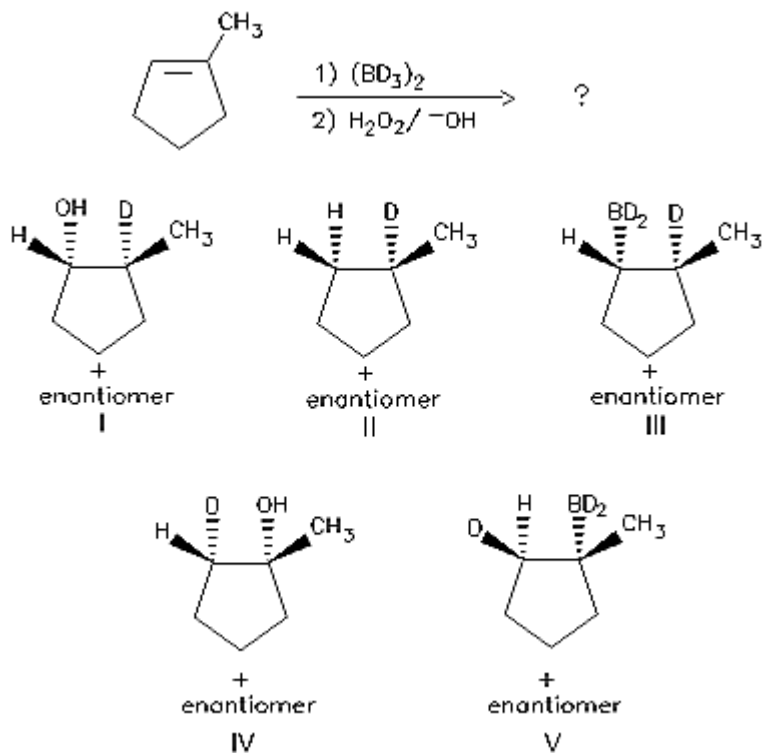
- A) I
- B) II
- C) III
- D) IV
- E) V

23. Select the structure of the major product formed from the following reaction.



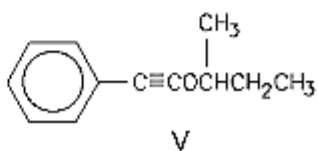
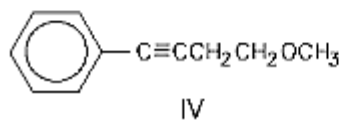
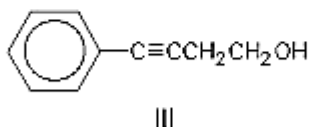
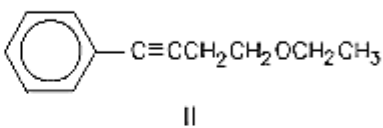
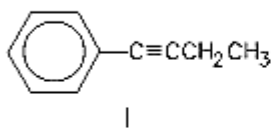
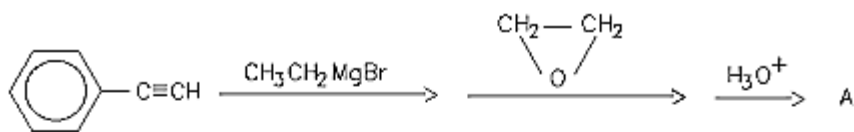
- A) I
- B) II
- C) III
- D) IV
- E) V

24. What product would you expect from the following reaction?



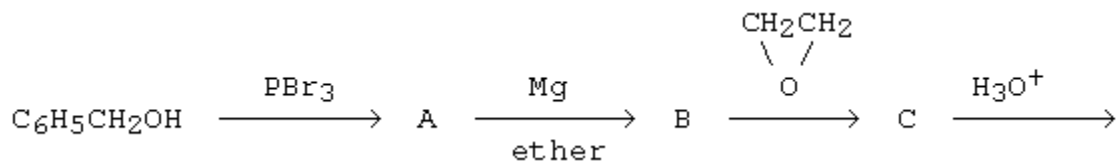
- A) I
- B) II
- C) III
- D) IV
- E) V

25. What is the product, A, that would be obtained from the following reaction sequence?



- A) I  
B) II  
C) III  
D) IV  
E) V

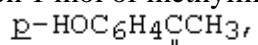
26. The final product, D, in the following reaction sequence,



would be?

- A)  $\text{C}_6\text{H}_5\text{CH}_2\text{OCH}_2\text{CH}_3$   
B)  $\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2\text{OH}$   
C)  $\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$   
D)  $\text{C}_6\text{H}_5\text{CH} \begin{array}{c} | \\ \text{Br} \end{array} \text{CH}_2\text{CH}_2\text{OH}$   
E)  $\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2\text{OCH}_3$

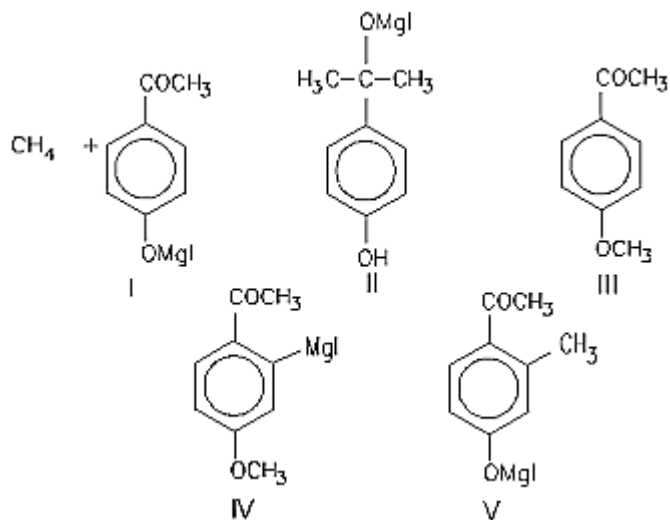
27. The principal product(s) formed when 1 mol of methylmagnesium iodide reacts with 1



mol of p-hydroxyacetophenone, i.e.



is/are:



- A) I
- B) II
- C) III
- D) IV
- E) V

## Answer Key

1. E
2. D
3. E
4. B
5. A
6. B
7. D
8. B
9. B
10. C
11. A
12. D
13. B
14. B
15. E
16. D
17. C
18. C
19. C
20. E
21. C
22. C
23. D
24. A
25. C
26. C
27. A