

MIDTERM #2 - SOLUTIONS

CHM 2120 3X

Prof: Mark Dornan

Last name: _____

First name: _____

Student #: _____

Date: July 14th 2015**Notes:**

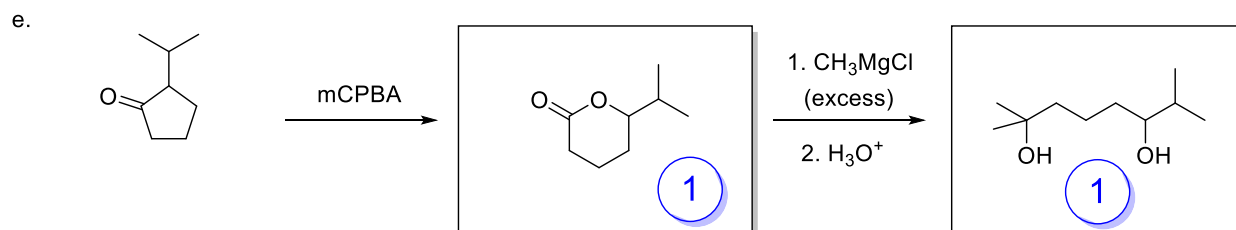
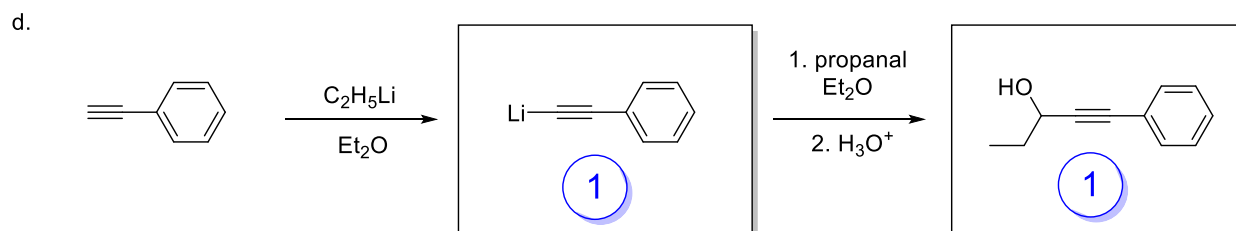
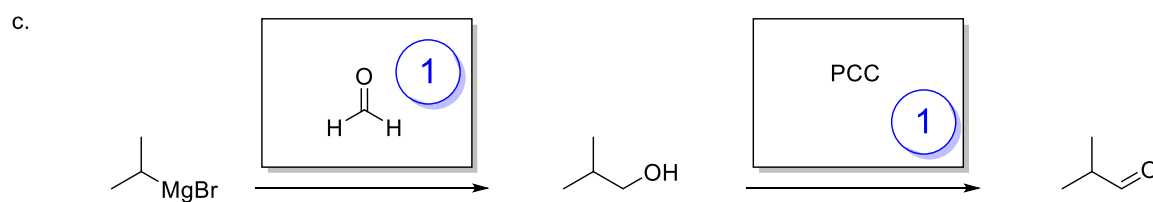
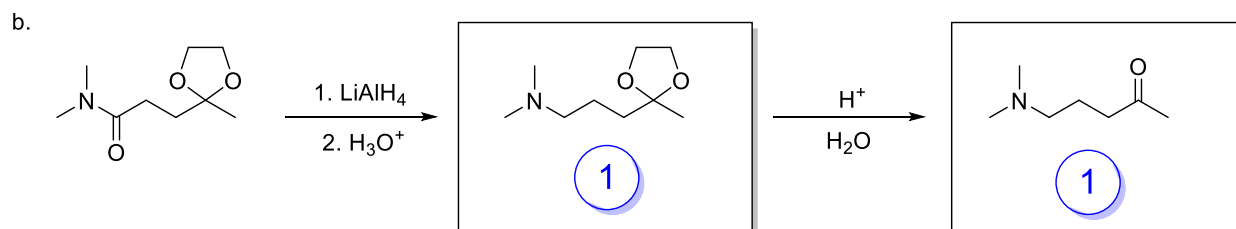
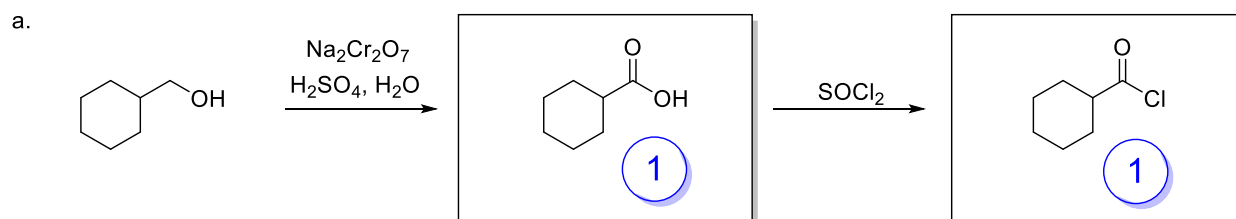
- Approximate total number of points: 64
- There are 9 pages in this exam
- Molecular models are allowed
- Re-grade requests for exams written in pencil will not be considered

1a	2a	3b	4b	5b	6b	7b	8	1b	2b	3a	4a	5a	6a	7a	0		
1 H															2 He		
3 Li	4 Be										5 B	6 C	7 N	8 O	9 F	10 Ne	
11 Na	12 Mg										13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	89 Ac	104 Rf	105 Ha	106 106												

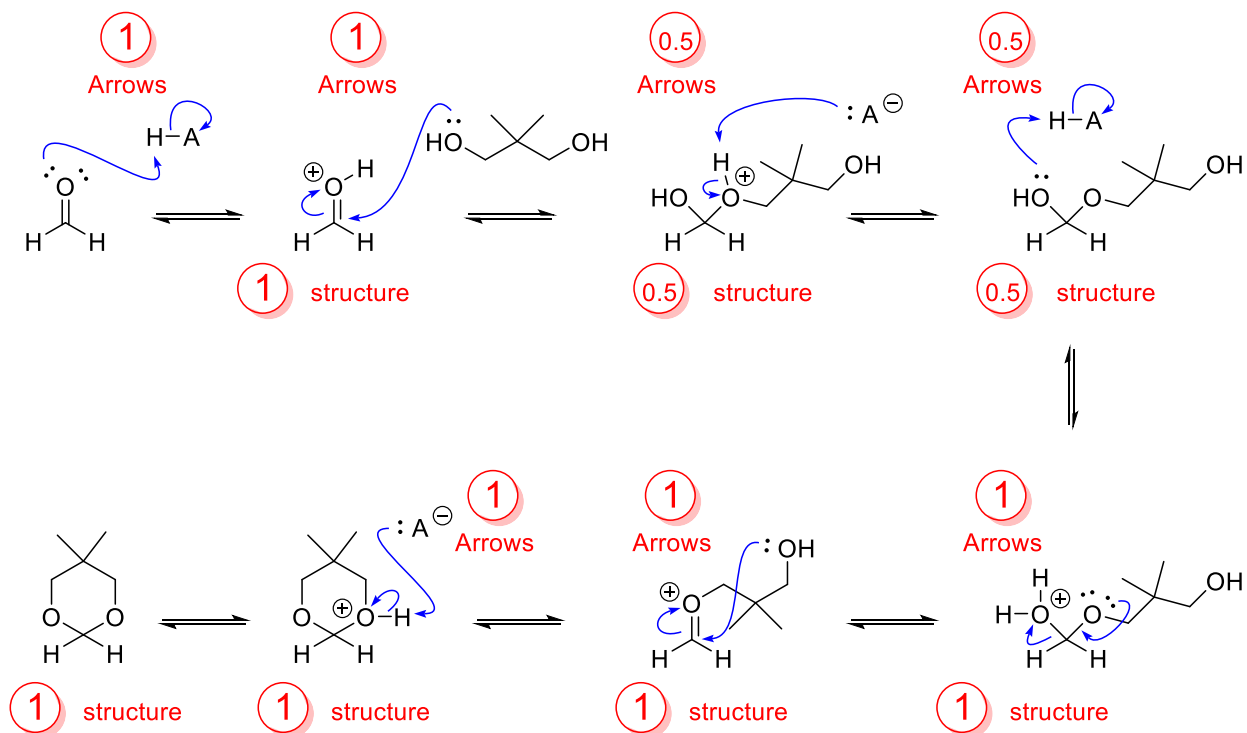
pK_a table

Acid	pK _a
HBr	-9
ROH ₂ ⁺	-2
H ₃ O ⁺	-1.75
RCO ₂ H	5
R ₃ NH ⁺	10-11
H ₂ O	15.75
ROH	16-18
RC(O)CH ₃	20
HC≡CH	25
H ₂	36
R ₂ NH	35-40
CH ₂ =CH ₂	44
RCH ₃	55

1. Fill in the missing structures or reagents (10 marks).



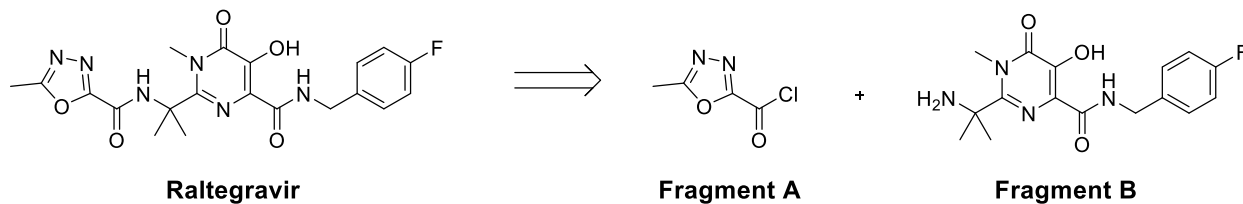
2. Provide a detailed mechanism for the formation of the major product of the reaction of 2,2-dimethyl-1,3-propanediol with acetone in the presence of an acid catalyst. (12 marks)



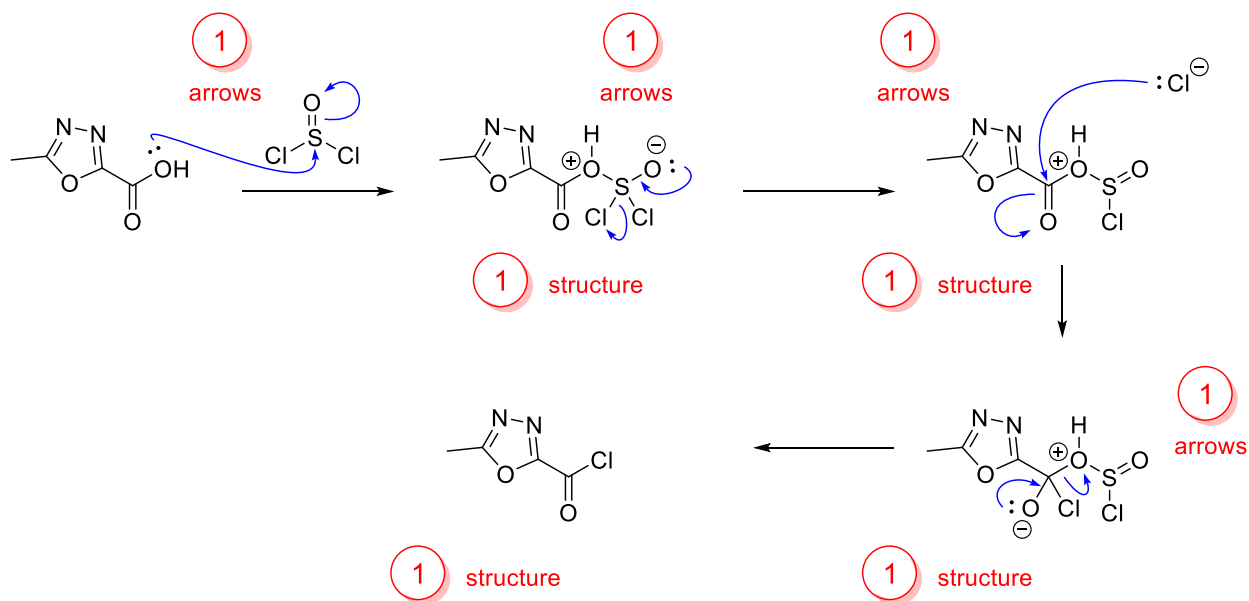
Doesn't matter which acid catalyst is used.

Doesn't matter what they use to do the proton shuttle (A- or another molecule of the diol)

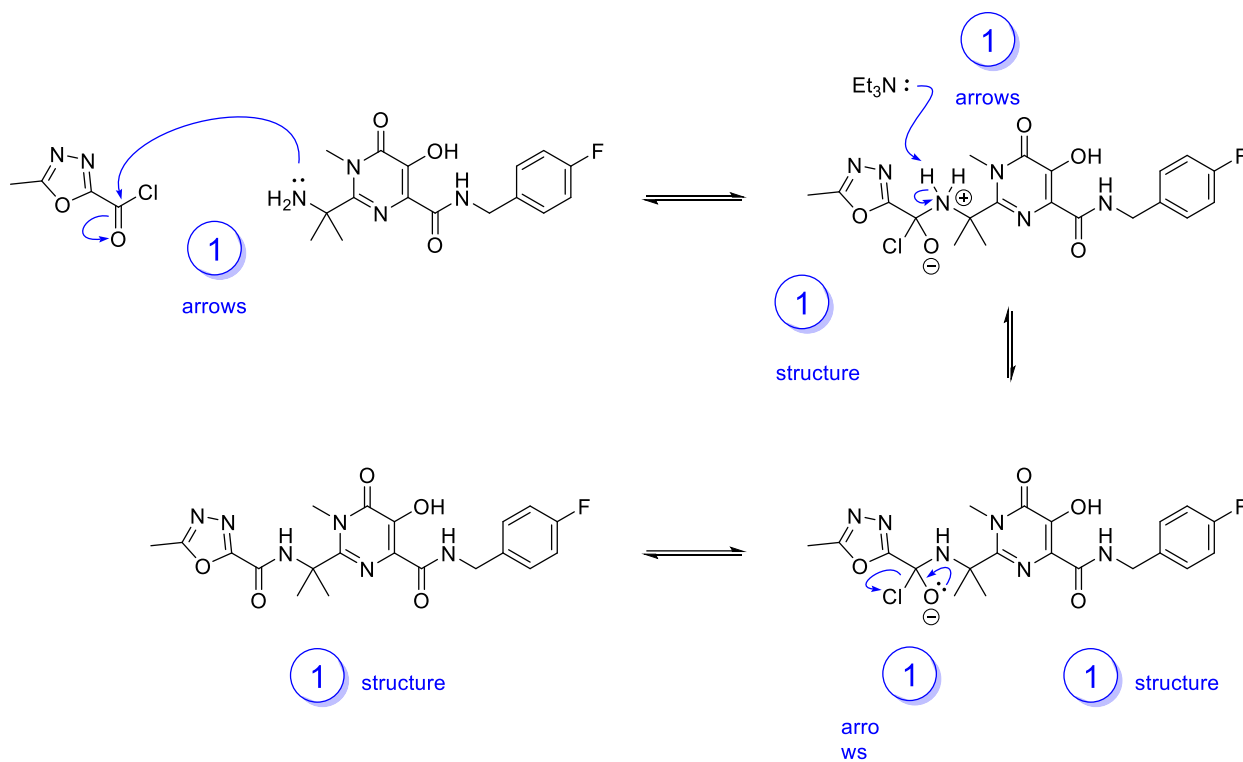
3. Raltegravir is an HIV-integrase inhibitor developed by Merck which is used to treat HIV infection in children. The last step of the synthesis involves an amide bond forming reaction between the two starting materials shown below.



- a. Draw a detailed mechanism for the formation of Fragment A from the corresponding carboxylic acid using any additional reagents necessary (8 marks).

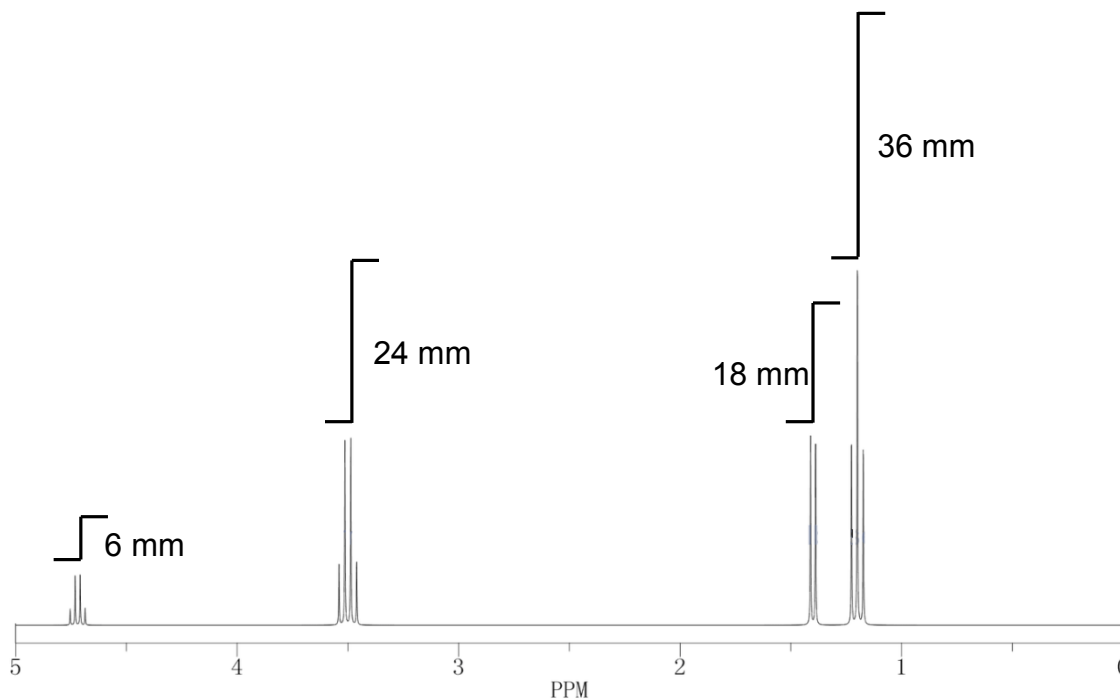


b. Draw a detailed mechanism of the formation of Raltegravir using Fragment A, Fragment B and any additional reagents necessary. (6 marks)



Any base acceptable. Can also use another equiv of amine as base

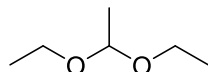
4. An unknown compound has the formula $C_6H_{14}O_2$. Fill in the table (5 marks) and elucidate the structure of the molecule by analyzing the following 1H NMR spectrum. (5 marks)



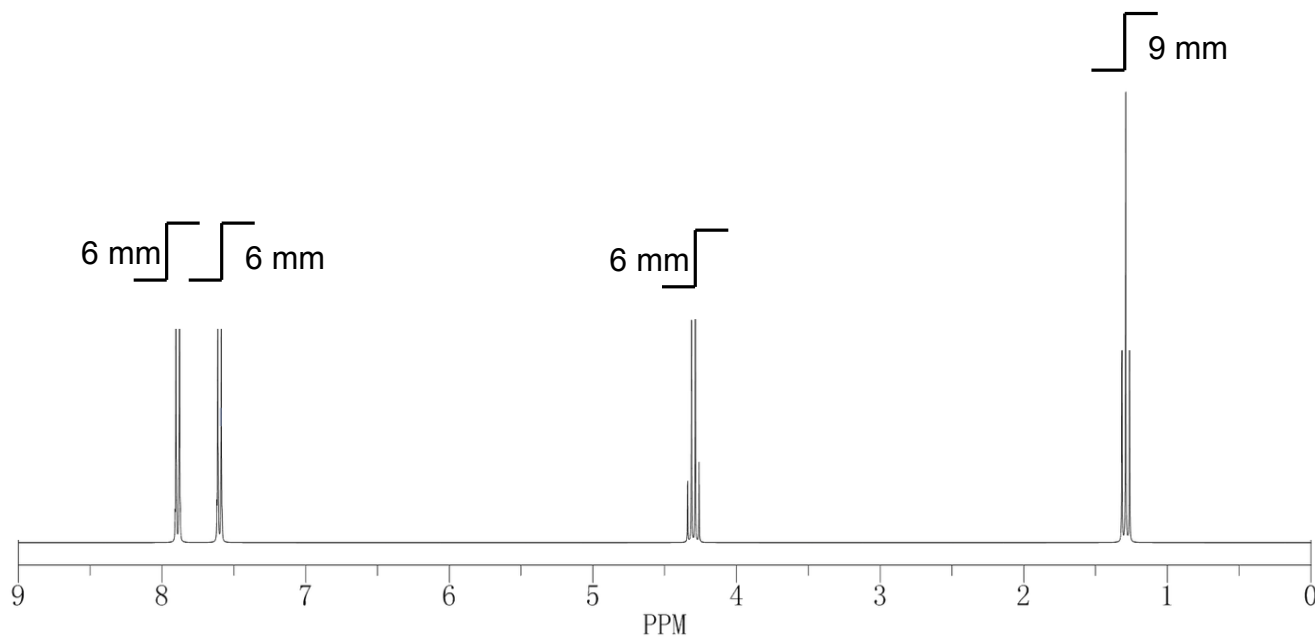
Signal	δ	Integration	Multiplicity	Ideas
A	4.7	1H	q, n=3	O-CH-CH ₃
B	3.5	4H	q, n=3	O-CH ₂ -CH ₃
C	1.4	3H	d, n=1	CH ₃ -CH
D	1.2	6H	t, n=2	CH ₃ -CH ₂

1 mark for filling in chemical shift column correctly, 1 mark for filling in integration column correctly, 1 mark for filling in multiplicity column correctly, plus up to 2 marks for any correct ideas. If answer is wrong, up to 2 points for close ideas, or putting fragments together correctly.

Correct structure:



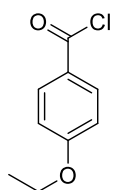
5. An unknown compound has the formula $C_9H_9O_2Cl$. An IR spectrum of the compound shows a sharp peak at 1750 cm^{-1} . Fill in the table (**5 marks**) and elucidate the structure of the molecule by analyzing the following 1H NMR spectrum. (**5 marks**)



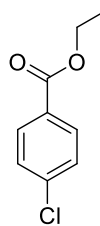
Signal	δ	Integration	Multiplicity	Ideas
A	7.9	2H	d, n = 1	Aromatic, symmetry, 2x Ar-H
B	7.6	2H	d, n = 1	Aromatic, symmetry, 2x Ar-H
C	4.3	2H	q, n = 3	CH2-CH3 connected to electroneg atom
D	1.3	3H	t, n = 2	CH2-CH3

1 mark for filling in chemical shift column correctly, 1 mark for filling in integration column correctly, 1 mark for filling in multiplicity column correctly, plus up to 2 marks for any correct ideas. If answer is wrong, up to 2 points for close ideas, or putting fragments together correctly.

Correct answer

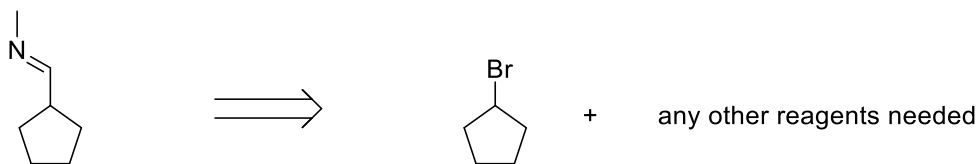


Also accepted for full marks:

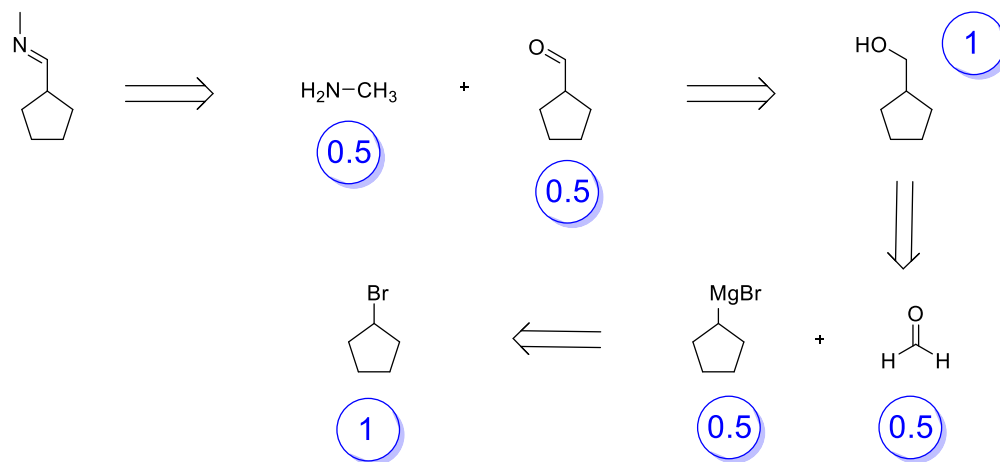


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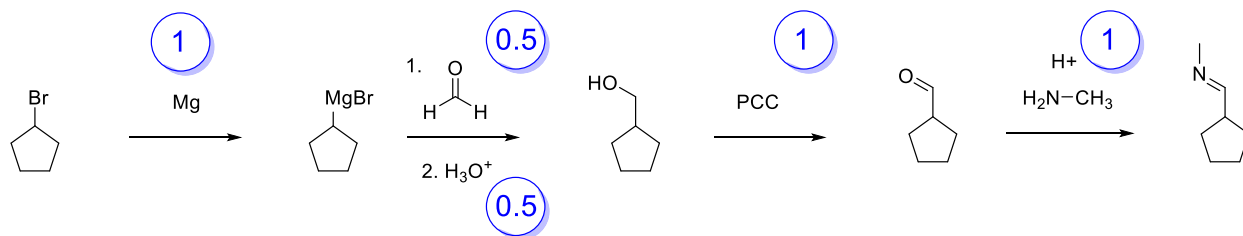
6. Design a synthesis of N-(cyclopentylmethylidene)-methanamine from bromocyclopentane. Include a retrosynthesis (4 marks), and the forward synthesis (4 marks) including any other reagents needed. Mechanisms are not required.




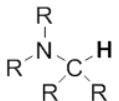

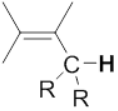
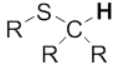
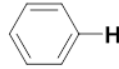
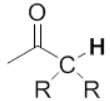


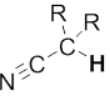

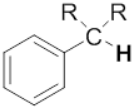

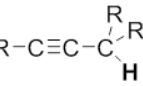

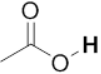
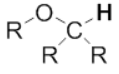
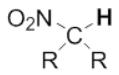
Retro



Synthesis



Typical ^1H chemical shifts

	0.7 - 1.7		2.2 - 2.9		4.5 - 7.0
	1.6 - 2.6		2.0 - 3.0		6.5 - 8.0
	2.1 - 2.5		2.0 - 4.0		9.0 - 10.0
	2.1 - 3.0		2.7 - 4.1		
	2.3 - 2.7		3.1 - 4.1		
	1.7 - 2.7		4.2 - 4.8		
	11.0 - 12.0		3.0 - 5.0		
			4.1 - 4.3		

*** All are approximate**