

## General

$$\rho = \frac{m}{V} \quad \text{specific density} = \rho = V/m$$

$$1 \text{ bar} = 100000 \text{ Pa}$$

$$T (\text{K}) = T (^{\circ}\text{C}) + 273 \text{ K}$$

## Ideal Gas

$$pV = nRT$$

$$\rho = \frac{RT}{p}$$

$$p = \frac{p}{RT}$$

$$\text{useful arrangement: } \frac{p_1 V_1}{T_1} = \frac{p_2 V_2}{T_2}$$

## Gases

### Energy for all systems

$$\text{Change in internal energy: } (U_2 - U_1) = m (C_v (T_2 - T_1))$$

$$\text{" " " " per kg: } (u_2 - u_1) = C_v (T_2 - T_1)$$

$$\text{" " enthalpy: } (H_2 - H_1) = m C_p (T_2 - T_1)$$

$$\text{" " " " per kg: } h_2 - h_1 = C_p (T_2 - T_1)$$

$$\text{Gas constant } R = C_p - C_v$$

$$\gamma = C_p / C_v$$

### For Polytropic process

$$p_1 V_1^n = p_2 V_2^n$$

$$\frac{T_2}{T_1} = \left( \frac{p_2}{p_1} \right)^{\frac{n-1}{n}} = \left( \frac{V_1}{V_2} \right)^{n-1}$$

### For adiabatic process (Q=0)

$$p_1 V_1^\gamma = p_2 V_2^\gamma, \quad \frac{T_2}{T_1} = \left( \frac{p_2}{p_1} \right)^{\frac{\gamma-1}{\gamma}} = \left( \frac{V_1}{V_2} \right)^{\gamma-1}$$

$$\text{For constant volume heating: } Q = U_2 - U_1 = m C_v (T_2 - T_1)$$

$$\text{For constant pressure heating: } Q = H_2 - H_1 = m C_p (T_2 - T_1)$$

### For Mixed Saturation

$$u = x u_g + (1-x) u_f$$

$$h = x h_g + (1-x) h_f$$

$$v = x v_g + (1-x) v_f$$

### Non-Flow Problems

$$Q - W = U_2 - U_1$$

$$\text{For unit mass: } q - w = u_2 - u_1$$

$$W = \int p dV$$

When	Work
$V = C$	$W = 0$
$p = C$	$W = p (V_2 - V_1) = W$
$pV = C$	$p_1 V_1 \ln (V_2 / V_1) = W$
$pV^n = C$	$W = (p_1 V_1 - p_2 V_2) / (n-1)$
$pV^\gamma = C$	$W = (p_1 V_1 - p_2 V_2) / (\gamma-1)$

→

## Steady Flow Problems

$$\dot{Q} - \dot{W} = \dot{m}_{\text{out}} \left( h_1 + \frac{1}{2} V^2 + z_1 \right) - \dot{m}_{\text{in}} \left( h_2 + \frac{1}{2} V^2 + z_2 \right)$$

## Entropy

### Closed system entropy balance

$$s_2 - s_1 = \int_1^2 \frac{\delta Q}{T} + \sigma$$

entropy change      entropy transfer      entropy balance

### Closed system entropy rate balance

$$s_2 - s_1 = \sum \frac{\dot{Q}}{T} + \dot{\sigma}$$

### Control Volume entropy rate balance

$$\dot{s} = \sum \frac{\dot{Q}}{T} + \sum \dot{m}_e s_e - \sum \dot{m}_o s_o + \dot{\sigma}, \quad \text{steady state when } \dot{s} = 0$$

### Isentropic Turbine efficiency

$$\eta = \frac{\dot{W}_{\text{cv}} / \dot{m}}{(\dot{W}_{\text{cv}} / \dot{m})_s} = \frac{h_1 - h_2}{h_1 - h_{2s}}, \quad \boxed{s_1 = s_{2s}}$$

### Isentropic Compressor efficiency

$$\eta = \frac{(-\dot{W}_{\text{cv}} / \dot{m})_s}{(-\dot{W}_{\text{cv}} / \dot{m})} = \frac{h_{2s} - h_1}{h_2 - h_1}$$

### Change in entropy

$$s(T_2, V_2) - s(T_1, V_1) = \int_{T_1}^{T_2} C_v(T) \frac{dT}{T} + R \ln \frac{V_2}{V_1}$$

If  $C_v = \text{constant}$ ,  $s_2 - s_1 = C_v \ln \left( \frac{T_2}{T_1} \right) + R \ln \left( \frac{V_2}{V_1} \right)$

$$s_2 - s_1 = s^\circ(T_2) - s^\circ(T_1) - R \ln \left( \frac{P_2}{P_1} \right)$$

$$s_2 - s_1 = C_p \ln \left( \frac{T_2}{T_1} \right) - R \ln \left( \frac{P_2}{P_1} \right)$$

### When $s_1 = s_2$

$$\frac{P_2}{P_1} = \frac{P_{r2}}{P_{r1}}, \quad \frac{V_2}{V_1} = \frac{V_{r2}}{V_{r1}}$$

### When $s_1 = s_2$ , constant heat ratio $k$

$$\frac{T_2}{T_1} = \left( \frac{P_2}{P_1} \right)^{\frac{k-1}{k}}, \quad \frac{T_2}{T_1} = \left( \frac{V_1}{V_2} \right)^{k-1}$$

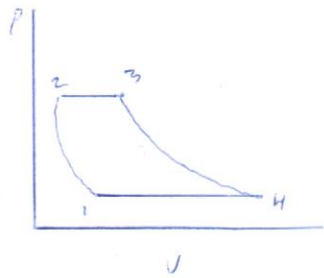
$$\frac{P_2}{P_1} = \left( \frac{V_1}{V_2} \right)^k$$

use for turbines & compressors

### Heat exchanger

$$\eta_{\text{HX}} = \frac{T_2 - T_1}{T_x - T_1}$$

# Brayton Cycle

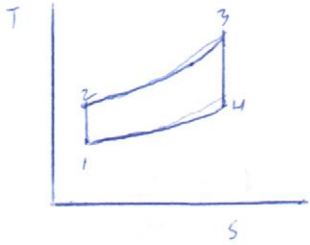


$$s_1 = s_2$$

$$s_3 = s_4$$

$$P_2 = P_3$$

$$P_4 = P_1$$

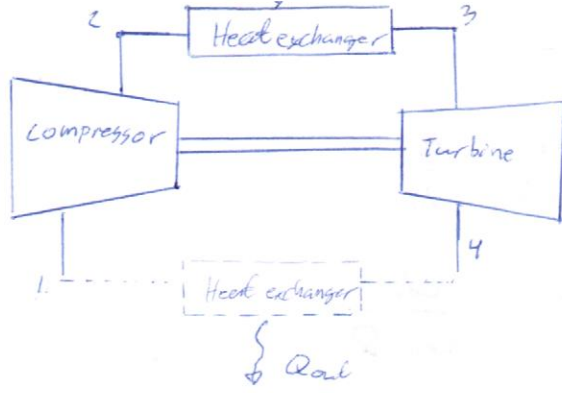


$$1. W_2 = h_2 - h_1$$

$$2. Q_{23} = h_3 - h_2$$

$$3. W_4 = h_3 - h_4$$

$$4. Q_{41} = h_4 - h_1$$



$$\frac{T_4}{T_1} = \frac{T_3}{T_2}$$

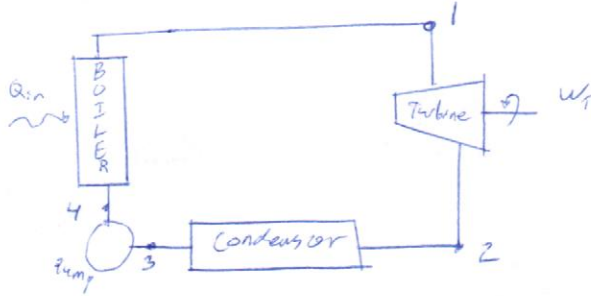
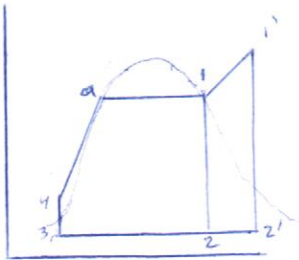
$$\eta_{cycle} = \frac{W_{net}}{Q_{in}} = 1 - \frac{h_4 - h_1}{h_3 - h_2}$$

$$\eta_{cycle} = 1 - \frac{1}{(P_2/P_1)^{\frac{\gamma-1}{\gamma}}} = 1 - \frac{1}{\left(\frac{T_2}{T_1}\right)^{\frac{\gamma-1}{\gamma}}}, \quad r = P_2/P_1$$

$$\eta_{comp} = \frac{h_2 - h_1}{h_2 - h_1}$$

$$\eta_{turb} = \frac{h_3 - h_4}{h_3 - h_4}$$

# Rankin



$$\eta = \frac{W_{out}}{Q_{in}} = 1 - \frac{h_2 - h_3}{h_1 - h_4}$$

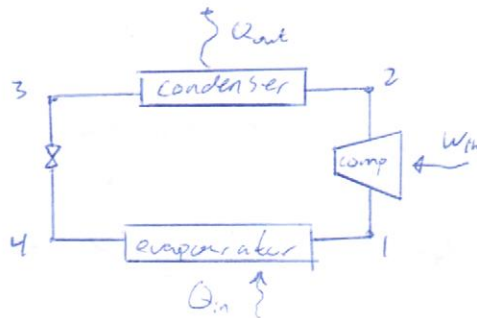
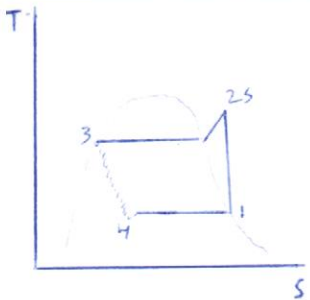
$$\text{Back work ratio} = BWR = \frac{h_4 - h_3}{h_1 - h_2}$$

$$W_L = h_1 - h_2 \quad W_P = h_4 - h_3$$

$$Q_{out} = h_2 - h_3 \quad Q_{in} = h_1 - h_4$$

$$P_2 = P_3 \quad P_4 = P_1 \quad W_{comp} = h(P_2 - P_1)$$

# Rankin VC Refrigeration



$$COP_{ref} = \frac{h_1 - h_4}{h_2 - h_1}$$

$$COP_{hp} = \frac{h_2 - h_3}{h_2 - h_1}$$

$$P_2 = P_3 \quad q_{23} = 0 \quad W_2 = h_2 - h_1$$

$$h_4 = h_3 \quad q_{34} = 0 \quad W_3 = 0$$

$$q_{41} = 0 \quad W_4 = 0$$

$$q_{12} = h_1 - h_4 \quad W_1 = 0$$

# Ketodiyasyon Cycle

Coefficient of Performance

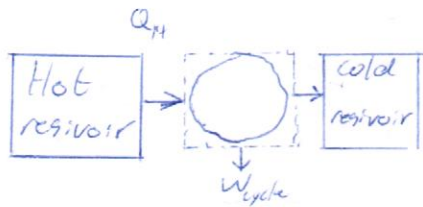
Refrigeration:  $COP = \frac{Q_c}{W_{cycle}} = \frac{Q_c}{Q_H - Q_c}$

Heat Pump:  $COP = \frac{Q_H}{W_{cycle}} = \frac{Q_H}{Q_H - Q_c}$

## Power Cycle

$E_2 - E_1 = Q_c - W_2 = 0$

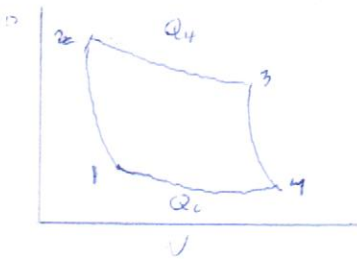
$W_{cycle} = Q_H - Q_c$



$\eta_{max} = 1 - \frac{T_c}{T_H}$

$\eta = \frac{W_{cycle}}{Q_H} = \frac{Q_H - Q_c}{Q_H} = 1 - \frac{Q_c}{Q_H}$

## Carnot Cycle

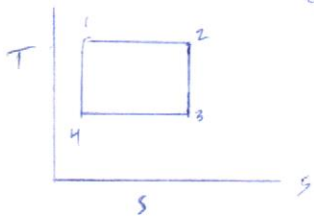


- 1→2 - adiabatic compression until gas reaches  $T_H$
- 2→3 - isothermal expansion receiving  $Q_H$  ( $T_2 = T_3 = T_H$ )
- 3→4 - adiabatic expansion until gas reaches  $T_c$
- 4→1 - isothermal compression discharging  $Q_c$  ( $T_4 = T_1 = T_c$ )

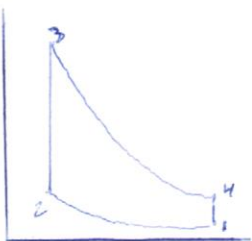
$\eta = 1 - \frac{T_c}{T_H} = \frac{W_{cycle}}{Q_H}$  } Carnot power cycle

Carnot refrigeration cycle }  $\eta = \beta_{carnot} = \frac{T_c}{T_H - T_c} = \frac{Q_H}{W_{cycle}}$

Carnot heat pump }  $\eta_{carnot} = \frac{T_H}{T_H - T_c}$



## Otto Cycle



$s_1 = s_2$   
 $s_3 = s_4$

${}_1Q_2 = 0$   
 ${}_3Q_4 = 0$

$V_2 = V_3$   
 $V_3 = V_4$

${}_1W_2 = u_2 - u_1$

${}_2Q_3 = u_3 - u_2$

${}_3W_4 = u_3 - u_4$

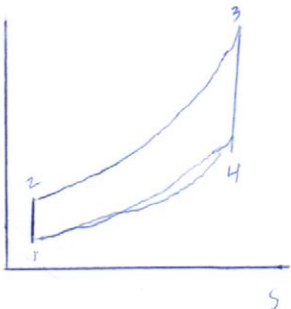
${}_4Q_1 = u_4 - u_1$

$\frac{T_4}{T_1} = \frac{T_3}{T_2}$

$\eta = 1 - \frac{1}{r^{k-1}}$   
 $r = \frac{V_1}{V_2}$

$\eta = \frac{W_{34} - W_{12}}{Q_{23}}$   
 $= \frac{(u_3 - u_2) - (u_4 - u_1)}{u_3 - u_2}$

$\eta = 1 - \frac{1}{r^{k-1}}$ ,  $r = \frac{V_1}{V_2}$



## Heat Transfer: Conduction

$$\dot{Q} = -kA \frac{dT}{dx} \Rightarrow \dot{Q}_z = kA \frac{(T_1 - T_2)}{x}$$

$x$  = distance through material, some times shown as  $l$   
 $A = xL$  area

## Convection

$$\dot{Q}_{conv} = hA(T_s - T_\infty), \quad h = \text{convection heat transfer coefficient}$$

$$\text{Nusselt} = \text{Nu}_x = \frac{hx}{k}$$

$$\text{Nu}_x = 0.332 \text{Re}^{1/2} \text{Pr}^{1/3}$$

$$\text{Reynolds} = \text{Re}_x = \frac{\rho U_\infty x}{\mu}$$

$x$  = distance from edge of plate  
 $\rho$  = density

$$\text{Prandtl Pr} = \frac{\nu}{\alpha} = \frac{(\mu/\rho)}{(k/\rho c_p)}$$

$\mu$  = fluid viscosity (m/s)

$\mu$  = dynamic viscosity (kg/m.s)

$\nu$  = kinematic viscosity (m<sup>2</sup>/s)

$T_f = 0.5(T_w + T_\infty)$ , used to get values

## Radiation Blackbody

$$\dot{Q}_{rad} = \sigma A T_s^4, \quad \text{where } \sigma = 5.6704 \times 10^{-8} \text{ W/m}^2 \text{K}^4, \quad T_s = \text{surface temp}$$
$$= 0.1714 \text{ Btu/hr} \cdot \text{ft}^2 \cdot \text{R}^4$$

~~Q<sub>absorbed</sub> = \alpha Q<sub>incident</sub>, \alpha = absorptivity~~

## Grey Body Radiation

$$\dot{Q}_{rad} = \epsilon \sigma A (T_s^4 - T_{surround}^4), \quad \epsilon = \text{emissivity}, \quad 0 < \epsilon < 1, \quad \text{black body } \epsilon = 1$$

$$\dot{Q}_{absorbed} = \alpha \dot{Q}_{incident}$$

$\alpha$  = absorptivity

~~Combined Transfer~~

$$\dot{Q}_{incident} = A_{sur} \sigma T_{sur}^4$$

for small surface  $\dot{Q}_{absorbed} = \epsilon A_s (\sigma T_{sur}^4)$

$$\dot{Q}_{absorbed} = \alpha A_s \frac{\dot{Q}_{incident}}{A_{surround}} = \epsilon A_s \alpha A_s (\sigma T_{sur}^4)$$

$$\dot{Q}_{net} = \epsilon \sigma A_s (T_s^4 - T_{sur}^4)$$

$$\dot{Q}_{total} = \epsilon \sigma A_s (T_s^4 - T_{sur}^4) + h A_s (T_s - T_\infty)$$

## Combined Transfer

Transfer methods at boundaries are equal