

# Assignment #1 - Solutions

1. (1.4 in text)

$$a) {}^2C_1 = \frac{2!}{(2-1)!1!} = 2, \text{ There are 2 ways.}$$

$$b) {}^{20}C_{10} = \frac{20!}{10!(20-10)!} = 184756, \text{ There are 184,756 ways.}$$

$$c) {}^{2 \times 10^{23}}C_{1 \times 10^{23}} = \frac{(2 \times 10^{23})!}{(1 \times 10^{23})!(2 \times 10^{23} - 1 \times 10^{23})!} = \frac{(2 \times 10^{23})!}{[(1 \times 10^{23})!]^2}$$

These are too big to calculate directly  $\rightarrow$  Stirling's approximation:

$$\ln(2 \times 10^{23}!) \approx (2 \times 10^{23}) \ln(2 \times 10^{23}) - 2 \times 10^{23} = 1.053 \times 10^{25}$$

$$\ln[(1 \times 10^{23}!)]^2 = 2 \ln(1 \times 10^{23}!) \approx 2 [1 \times 10^{23} \ln(1 \times 10^{23}) - 1 \times 10^{23}] = 1.039 \times 10^{25}$$

$$\text{So, } \ln({}^{2 \times 10^{23}}C_{1 \times 10^{23}}) = \ln(2 \times 10^{23}!) - \ln[(1 \times 10^{23}!)]^2 \approx 1.386 \times 10^{23}$$

$$\text{and, } {}^{2 \times 10^{23}}C_{1 \times 10^{23}} \approx e^{1.386 \times 10^{23}} = 10^{6.02 \times 10^{22}}$$

Thus, there are approximately  $10^{6.02 \times 10^{22}}$  ways of arranging  $10^{23}$  quanta among  $2 \times 10^{23}$  atoms.

2. (2.5 in text)

We assume  $T_2 > T_1$  for simplicity (works either way). From conservation of energy we must have heat flow into body 1 equal to heat flow out of body 2, i.e.,

$$\Delta Q_1 + \Delta Q_2 = 0$$

$$C_1 \Delta T_1 + C_2 \Delta T_2 = 0$$

Both bodies have same final temps, so,

$$\Delta T_1 = T_F - T_1, \quad \Delta T_2 = T_F - T_2$$

We thus have,

$$C_1(T_F - T_1) + C_2(T_F - T_2) = 0$$

Rearrange for  $T_F$ ,

$$T_F = \frac{C_1 T_1 + C_2 T_2}{C_1 + C_2}$$

We now assume  $C_1 \gg C_2$ . Rewrite the above as,

$$T_F = \frac{C_1 T_1 + C_2 T_2}{C_1 + C_2} \cdot \frac{1/C_1}{1/C_1} = \frac{T_1 + C_2/C_1 T_2}{1 + C_2/C_1}$$

Since  $C_2 \ll C_1$ , we have  $C_2/C_1 \ll 1$ . We can thus expand the denominator above. With  $x = C_2/C_1$ ,

$$(1+x)^{-1} = 1 - x + x^2 - x^3 + \dots \approx 1 - x \quad (\text{since } x \ll 1)$$

We therefore have,

$$T_F = \frac{T_1 + C_2/C_1 T_2}{1 + C_2/C_1} \approx (T_1 + \frac{C_2}{C_1} T_2) (1 - \frac{C_2}{C_1}) = T_1 + \frac{C_2}{C_1} (T_2 - T_1)$$

Which is the desired result.

3. (3.3 in text)

We have the Poisson distribution,

$$P(x) = \frac{e^{-m} m^x}{x!}$$

We will use the power series representation of  $e^m$ ,

$$e^m = \sum_{x=0}^{\infty} \frac{m^x}{x!}$$

a) We have,

$$\sum_{x=0}^{\infty} P(x) = \sum_{x=0}^{\infty} \frac{e^{-m} m^x}{x!} = e^{-m} \sum_{x=0}^{\infty} \frac{m^x}{x!} = e^{-m} \cdot e^m = 1$$

b) We have,

$$\begin{aligned} \sum_{x=0}^{\infty} xP(x) &= \sum_{x=0}^{\infty} \frac{e^{-m} m^x}{x!} \cdot x = e^{-m} \sum_{x=0}^{\infty} x \frac{m^x}{x!} = e^{-m} \sum_{x=1}^{\infty} x \frac{m^x}{x!} \\ &= m e^{-m} \sum_{x=1}^{\infty} \frac{m^{x-1}}{(x-1)!} \end{aligned}$$

Let  $k = x-1$ ,

$$\sum_{x=0}^{\infty} xP(x) = m e^{-m} \sum_{k=0}^{\infty} \frac{m^k}{k!} = m e^{-m} \cdot e^m = m$$

Thus,  $\langle x \rangle = \sum_{x=0}^{\infty} xP(x) = m$

c) The average number of deaths/yr/corps ( $\lambda$ ) is,

$$\begin{aligned} \lambda &= (0) \binom{109}{200} + (1) \binom{65}{200} + (2) \binom{22}{200} + (3) \binom{3}{200} + (4) \binom{1}{200} \\ &= 122/200 \\ &= 0.61 \end{aligned}$$

To calculate expected frequencies multiply probability of observing  $n$  deaths/yr/corps by number of observations. For  $n=0$  we have,

$$\text{Expected Frequency} = 200 \cdot \frac{e^{-0.61} \cdot (0.61)^0}{0!} = 108.7$$

To calculate for  $n \geq 5$  deaths/yr/corps use the fact that,

$$P(n \geq 5) = 1 - P(n < 5)$$

# of deaths per year per corps	Observed Frequency	Calculated Frequency
0	109	108.7
1	65	66.3
2	22	20.2
3	3	4.1
4	1	0.6
≥5	0	0.1
TOTAL	200	200

4. (3.6 in text)

This is almost identical to example 3.7 in text. We have,

$$Y = \frac{1}{n} (X_1 + X_2 + \dots + X_n)$$

the mean of  $Y$  is thus,

$$\langle Y \rangle = \frac{1}{n} (\langle X_1 \rangle + \langle X_2 \rangle + \dots + \langle X_n \rangle)$$

All  $X_i$  have the same mean, so,

$$\langle Y \rangle = \frac{1}{n} \cdot n \langle X \rangle = \langle X \rangle$$

To find  $\sigma_y^2$  we will use,

$$\sigma_y^2 = \langle Y^2 \rangle - \langle Y \rangle^2$$

Thus,

$$\begin{aligned} \langle Y^2 \rangle &= \frac{1}{n^2} \langle (X_1 + X_2 + \dots + X_n)^2 \rangle \\ &= \frac{1}{n^2} \langle X_1^2 + \dots + X_n^2 + X_1 X_2 + X_2 X_1 + X_1 X_3 + \dots \rangle \\ &= \frac{1}{n^2} [ \langle X_1^2 \rangle + \dots + \langle X_n^2 \rangle + \langle X_1 X_2 \rangle + \langle X_2 X_1 \rangle + \langle X_1 X_3 \rangle + \dots ] \\ &= \frac{1}{n^2} [ n \langle X^2 \rangle + n(n-1) \langle X \rangle^2 ] \end{aligned}$$

So,

$$\begin{aligned}\sigma_y^2 &= \frac{1}{n} [\langle x^2 \rangle - (n-1)\langle x \rangle^2] - \langle x \rangle^2 \\ &= \frac{1}{n} [\langle x^2 \rangle - (n-1)\langle x \rangle^2 - n\langle x \rangle^2] \\ &= \frac{1}{n} [\langle x^2 \rangle - \langle x \rangle^2] \\ &= \frac{1}{n} \sigma_x^2\end{aligned}$$

So,

$$\sigma_y = \frac{1}{\sqrt{n}} \sigma_x$$

5. (3.7 in text)

The binomial distribution is,

$$P(n, k) = \frac{n!}{k!(n-k)!} p^k (1-p)^{n-k}$$

a) Let  $m = np$ , so,

$$\begin{aligned}P(n, k) &= \frac{n!}{(n-k)! k!} \left(\frac{m}{n}\right)^k \left(1 - \frac{m}{n}\right)^{n-k} \\ &= \left[ \frac{n!}{(n-k)! n^k} \right] \cdot \frac{m^k}{k!} \left(1 - \frac{m}{n}\right)^{n-k}\end{aligned}$$

The term in square brackets has a limit as  $n \rightarrow \infty$  of,

$$\lim_{n \rightarrow \infty} \frac{n!}{(n-k)! n^k} = \lim_{n \rightarrow \infty} \frac{n(n-1)(n-2)\dots(n-k+1)}{n^k}$$

$$= \lim_{n \rightarrow \infty} \frac{n^k + an^{k-1} + bn^{k-2} + \dots + zn}{n^k} \quad (a, b, \dots, z \text{ just numbers})$$

$$= \lim_{n \rightarrow \infty} 1 + \frac{a}{n} + \frac{b}{n^2} + \dots + \frac{z}{n^{k-1}} = 1$$

We also have the two terms,

$$\left(1 - \frac{m}{n}\right)^{n-k} = \left(1 - \frac{m}{n}\right)^n \left(1 - \frac{m}{n}\right)^{-k}$$

Remember that,

$$e^{-\frac{m}{n}} = \sum_{x=0}^{\infty} \frac{\left(-\frac{m}{n}\right)^x}{x!} = 1 - \frac{m}{n} + \frac{m^2}{2n^2} - \dots$$

As  $n \rightarrow \infty$ ,  $\frac{m}{n} \rightarrow 0$  so we can take  $e^{-\frac{m}{n}} \approx 1 - \frac{m}{n}$  and,

$$\lim_{n \rightarrow \infty} \left(1 - \frac{m}{n}\right)^n = \left(e^{-\frac{m}{n}}\right)^n = e^{-m}$$

and,

$$\lim_{n \rightarrow \infty} \left(1 - \frac{m}{n}\right)^{-k} = \lim_{n \rightarrow \infty} \left(e^{-\frac{m}{n}}\right)^{-k} = \lim_{n \rightarrow \infty} e^{\frac{mk}{n}} = e^0 = 1$$

Thus,

$$\begin{aligned} \lim_{n \rightarrow \infty} P(n, k) &= \frac{m^k}{k!} \left[ \lim_{n \rightarrow \infty} \frac{n!}{(n-k)! n^k} \right] \left[ \lim_{n \rightarrow \infty} \left(1 - \frac{m}{n}\right)^{n-k} \right] \\ &= \frac{m^k e^{-m}}{k!} \end{aligned}$$

Which is what we wanted to show.

6.

We are given the length of the mercury column is  $l = 5 \text{ cm}$  at the triple point of water,  $T_T = 0.01^\circ\text{C}$ . We will assume that the ideal gas law (IGL) applies, so we can write,

$$PV = nRT$$

Here  $V$  is the volume of the Hg column, which we can write as,

$$V = \text{height} \cdot \text{area} \\ = l \cdot A$$



So, rearranging for  $l$  in the IGL gives,

$$l = \frac{nRT}{PA}$$

For what we are interested in  $n$ ,  $R$ ,  $P$ , and  $A$  are all constant, so let,

$$\alpha = nR/PA$$

and we have,

$$l = \alpha T$$

We can determine  $\alpha$  since  $l = 5 \text{ cm}$  at  $T = 0.01^\circ\text{C} = 273.16 \text{ K}$ ,

$$\alpha = \frac{l}{T} = \frac{5 \text{ cm}}{273.16 \text{ K}} = 1.83 \times 10^{-2} \text{ cm/K}$$

a) We want  $T$  when  $l = 6.0 \text{ cm}$ ;

$$T = \frac{l}{\alpha} = \frac{6.0 \text{ cm}}{1.83 \times 10^{-2} \text{ cm/K}} = 328 \text{ K}$$

b) We want  $l$  when  $T = 373.15 \text{ K}$ ,

$$l = \alpha T = (1.83 \times 10^{-2} \frac{\text{cm}}{\text{K}})(373.15 \text{ K}) \\ = 6.8 \text{ cm}$$

c) The length precision is  $\Delta l = 0.01 \text{ cm}$ . Since  $l$  is proportional to  $T$  we can simply write the temperature precision as,

$$\Delta T = \frac{\Delta l}{\alpha} = \frac{0.01 \text{ cm}}{1.83 \times 10^{-2} \text{ cm/K}}$$

$$= 0.5 \text{ K}$$

This is much greater than the separation between the triple point & freezing point, thus the thermometer could not be used.