

LECTURE 2

Scientific Method

- 1.) *Observe and Measure*
- 2.) *Form and Test a Hypothesis*
- 3.) *Develop a Theory*

Nomothetic Approach

Generalizes and comes up with theories that generally work

Idiographic Approach

Explains a single event, does not consider the entire system

Inductivism

Uses empirical data to generalize a theory

Use observations and should not contradict other theories

Presents a theory, waits for data to prove wrong

Falsification

Deduction from other theories

Uses experiments to prove other theories wrong

Uses for explanation until fitter theories are proven

System Approach

A set of variables and their relationship, distinct from relationships outside the system

Why? Because it stresses relationships between variables

It assists in developing theories

Open System

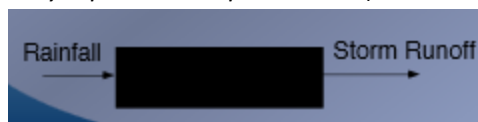
Inputs and Outputs (global energy entering and exiting the Earth's system)

Closed System

No input or output from the system (only internal) like the hydrological cycle of the Earth...water never leaves the Earth

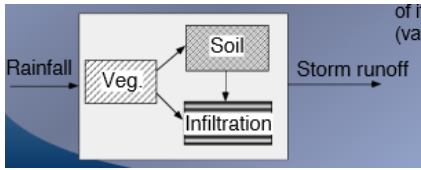
Black Box System

Only input and output matter (we don't know what goes on internally)



Grey Box

Partial understanding



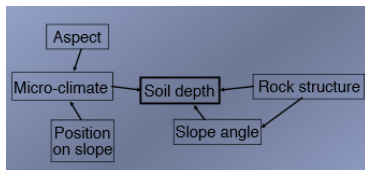
White Box

Comprehensive understanding of the internal system

Morphologic System

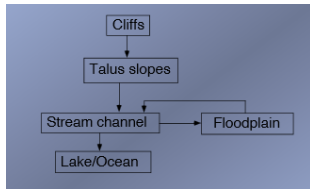
Vague relationships between elements

We do not understand the transfer of energy or matter between elements



Cascading

Relationship between elements understood through flux of matter



Process/response

Combination of morphologic and cascading

GEYSER

Control

Process response modified by human activity

FLOOD CHANNEL

Ecosystem

Process-response with physical, chemical, biological interactions

Positive Feedback

Small disturbance causes large change

A produces more B which in turn produces more A

Warmer atmosphere melts ice which changes the albedo and further warms the atmosphere

Negative Feedback

Disturbance reduces changes

Stable, self-regulating systems

Lake level increases, runoff increases, lake level gets back to normal

LECTURE 4

Black Body

Idealized body that emits radiation solely based on temperature, and irrelevant of shape or size

- Laws of Thermodynamics:**
- Energy can be transformed but not destroyed
 - Heat can never pass spontaneously from a colder to hotter body
 - Isolated system will reach equilibrium (Entropy)

Stefan Boltzmann Law

How much energy is emitted?

Measures intensity of radiation emitted from a body based on temperature

Wien's Displacement Law

What kind of energy is emitted?

Calculates the wavelength of emission from temperature of body

Inverse Square Law

Intensity decreases with distance

$$\text{Intensity} = 1/d^2$$

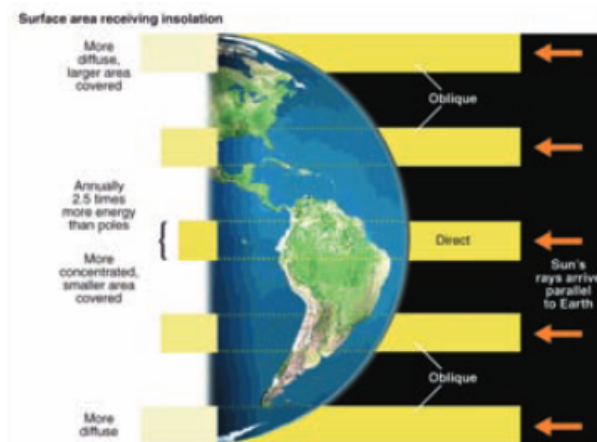
Solar Constant

Average amount of radiation received at the top of the atmosphere at right angles to solar beams

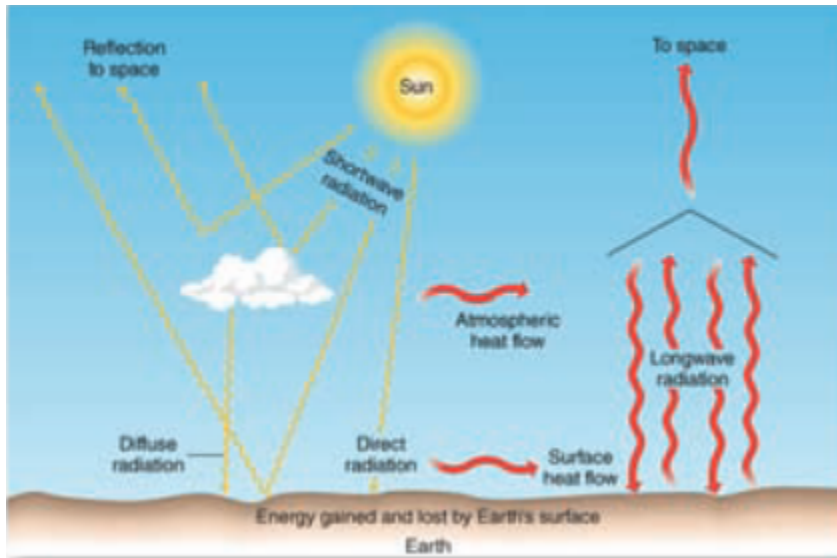
Calculated by the Stefan Boltzmann Law

Insolation

Amount of solar radiation received on a surface area



LECTURE 5



Scattering

Light changes direction without altering wave length
Hits molecules in the atmosphere

Refraction

Change in speed and direction of light
Rainbows created by raindrops refracting (change wavelength making colours)



Albedo

Percentage of incoming shortwave radiation that is reflected
High Albedo - → most radiation is reflected (Fresh snow)
Low Albedo → Absorbs a lot of shortwave radiation (wet ground)

Fresh Snow	90%
Dirt Snow	40%
Desert	25%
Meadows	15%
Water	5%

Gasses

Selectively absorb radiation

As a result, some wavelengths pass through the atmosphere while some are absorbed by gasses

Atmospheric Window

8 to 14 μm

Infrared wavelength that is emitted from earth and allowed to pass through the atmosphere into space without being absorbed

Conduction

Molecule transfer

Convection

Fluid movement energy transfer

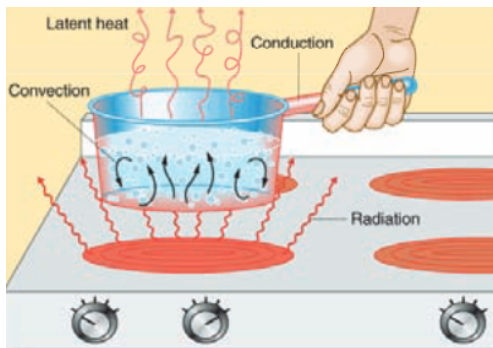
High density to low density

High pressure to low pressure

Latent Heat

Energy transferred during phase changes (not measurable)

When water evaporates in an ocean, latent heat is absorbed by water vapour resulting in LOWER AIR TEMPERATURE. Greater evaporation results in greater latent heat LOSS \rightarrow Colder air

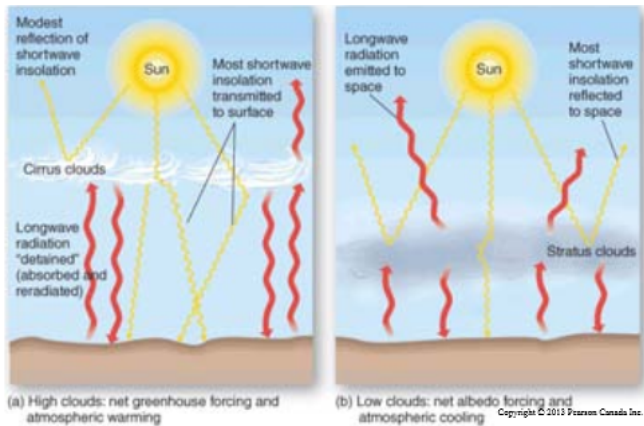


Sensible Heat

Heat that can be measured by a change in temperature

Greenhouse Effect

Short wave radiation passes through clouds, absorbed by Earth; long wave radiation emitted from Earth, reflected or absorbed by clouds back to Earth

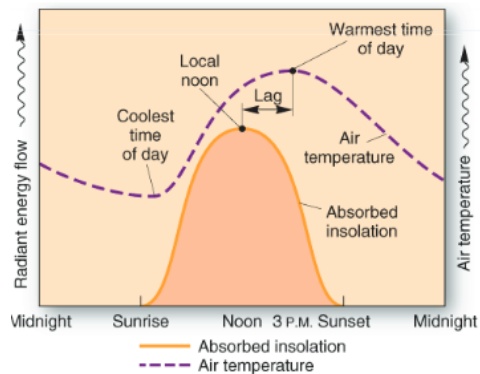


Radiation Balance Equation

$$Q = K(1 - \text{albedo}) + L_{\downarrow} - L_{\uparrow}$$

Daily Radiation Patterns

The lag is because the sun is still being absorbed but it takes a bit of time for the long wave radiation to be released by the surface. So we have a bit of lag because it is hotter when long wave radiation is released



Why do we have seasons?

Mainly because of the tilt of the Earth (decreases the intensity of solar beams)

Partly because on non-circular orbit

Electromagnetic Spectrum

Radio, Microwave, Infrared, Visible, UV, X-ray, Gamma

Atmospheric Layers

Thermosphere

Ionosphere (absorbs xrays, gamma rays, SOME UV rays)

Mesosphere

Stratosphere

→ bottom contains ozonosphere (reradiates longwave energy)

LECTURE 8

Covalent Bond

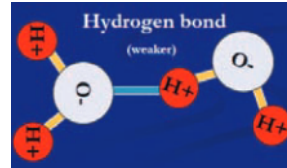
Strong bonds (H₂O)

Hydrogen Bond

Weak bond

Allow water to be liquid and solid

If we didn't have, water would only be a gas



Density

Maximum density at 4 degree Celsius

Solid ice is less dense than liquid water (crystalline structure)

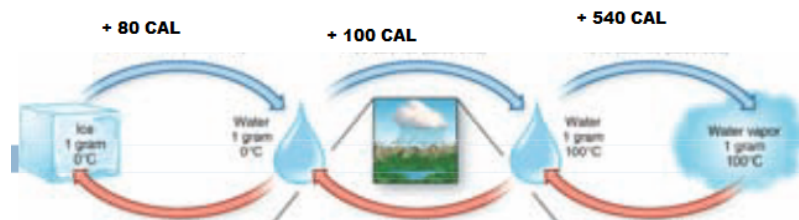
Phase Change

Function of temperature and air pressure

Triple point → At LOW pressure, and just above zero degrees, water is all phases at once

Specific Heat

Amount of energy required to increase the temperature by one degree



Latent Heat of Fusion

To melt ice → 80 cal

Latent Heat of Vaporization

To evaporate water → 540 cal

Absolute Humidity (e_a)

Varies with temperature

$$\frac{\text{Number of water molecules in air}}{\text{total mass of air}}$$

Vapour Pressure

The pressure that water vapor exerts on air

Saturation Vapour Pressure (e_s)

MAXIMUM absolute humidity

Air is holding all the water it can (for that temperature)

Relative Humidity (R_H)

$$R_H = \frac{e_a}{e_s}$$

Dewpoint Temperature

Temperature at which the air would be saturated for the current water content

Methods of Measuring Humidity

- 1.) Hair hygrometer → Change in tension of hair due to humidity (moves like a seismogram)
- 2.) Psychrometers → Dry vs. wet bulb temp (evaporate wet bulb and compare the temp change)
- 3.) Capacitive humidity sensors → (low accuracy) measures effect of humidity on metal oxide

Temperature Decrease in the Troposphere

Troposphere does not absorb solar energy, so temperature in the troposphere is only reflected from the ground

Environmental Lapse Rate

Rate of temperature change in troposphere due to adiabatic process

Atmospheric Stability

$ELR < DAR$ or MAR

Rising air parcel is cooler (denser) than surrounding air

Air parcel cools faster than the air cools

It will remain at same height

Clear skies

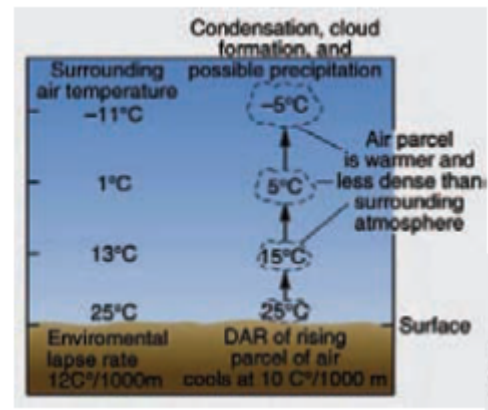
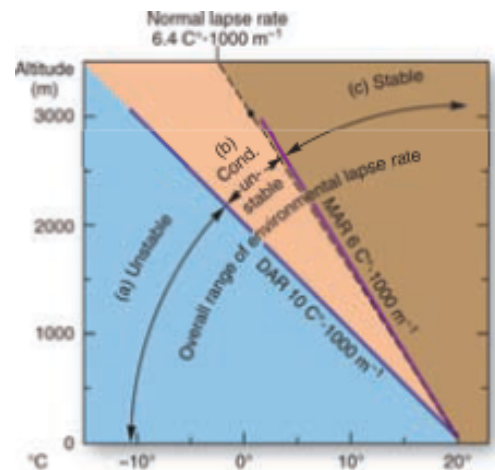
Atmospheric Instability

$ELR > DAR$ or MAR

Rising air parcel is warmer (less dense) than surrounding air

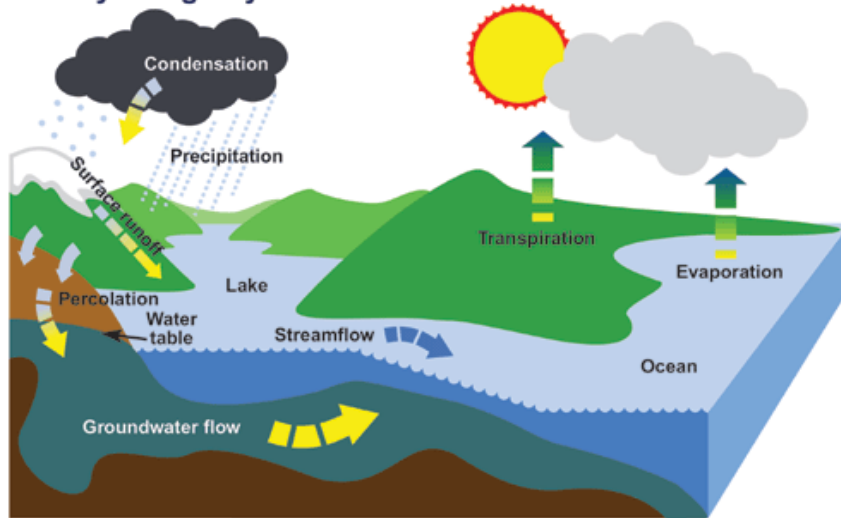
Air parcel cools slower than the lapse rate

Formation of clouds, condensation



Hydrologic Cycle

The hydrologic cycle



Evaporation

- Open water evaporation
- Interception loss (vegetation steals water and evaporates it)
- Transpiration (from leaves)
- Soil evaporation

Occurs when....

- 1.) Water is available
- 2.) Energy is available
- 3.) e_s at the surface is less than e_a above the surface

Cloud Formation

Cooling air increases its relative humidity close to 100%

Unstable conditions

Condensation

Vapour to liquid

Condensation Nuclei

Small particles that water vapour condensates onto

Dust, sea salt, sulfur, nitrogen oxide

Sea salt, sulfur (hygroscopic)

Soot, organic carbon (non-hygroscopic)

Cloud Types

- Stratus → Layer-like
- Cirrus → Thin, wispy, consist of ice particles

- *Cumulus* → *Thick puffy clouds*

Fog Types

- *Advection Fog*
- *Valley Fog*
- *Ice Crystal Fog*

LECTURE 9

Formation of Precipitation

- Cool air to almost 100% relative humidity
- Need condensation nuclei
- Cooling via rising air particles

Coalescence Process

Large water droplets collide and combine until their gravity force outweighs their buoyant and drag force

Ice-Crystal Process (Snow)

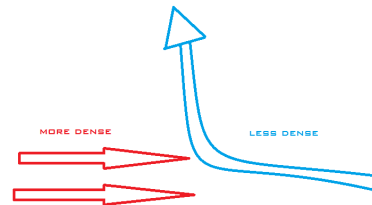
When a cloud contains both water droplets and ice particles

Water droplets evaporate and sublimate onto ice particles

1.) Convergent Lifting

Opposing air masses force the less dense mass to rise

Horizontal movement



2.) Convective Lifting

Convective heating of the lower atmosphere causes vertical lifting

Unstable atmosphere

Vertical movement

- Thunderstorms are convective precipitation
- Moist air rises and condensates (latent heat is released)

3.) Orographic Lifting

Air masses are forced to lift due to physical barriers (mountains)

Wet windward / dry leeward

4.) Frontal (Cyclonic) Lifting

Warm front → Light warm air overrides cool dense air

→ Small gradient boundary

→ Light precipitation (slow cooling)

→ Stratus clouds

Cold front → Cold dense air pushes warm air upwards by slipping under it

→ Steep boundary

→ Intense rainfall (quick cooling)

→ Cumulus clouds

Water Balance Equation

$$P = Q + E + \Delta S$$

Lecture 10

Climate

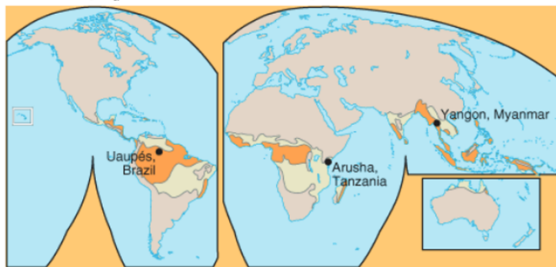
Average values of weather components over a 30-year period

- Function of:
- Latitude (influences solar intensity)
 - Altitude (environmental lapse rate, temperature)
 - Location of global high and low-pressure zones (influences winds)
 - Heat exchange from ocean currents (influences air temperature and evaporation)
 - Prevailing winds (if winds flow over water, will have maritime climate)
 - Land/Sea ratio (land heats and cools faster than water, continental have large seasonal temperature range)
 - Topography (orographic precipitation)

Köppen-Geiger Climate Classification

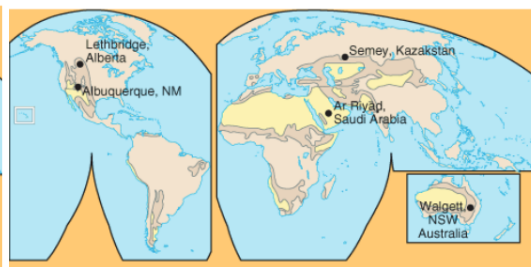
Type A: Tropical Climates

av. monthly $T_a > 18^\circ\text{C}$; $P > E$ so water surplus



Type B: Dry Climates

$PE > E$ so net water deficit



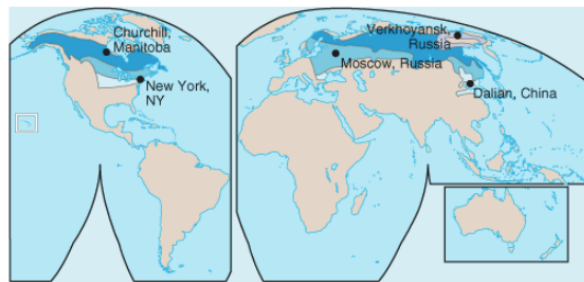
Type C: Mesothermal Climates

$-3^\circ\text{C} < T_a \text{ coldest month} < 18^\circ\text{C}$; at least one month $T_a > 10^\circ\text{C}$; strong seasonality



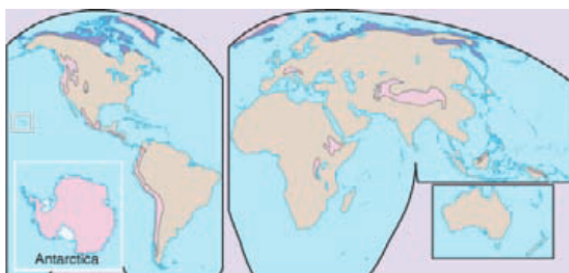
Type D: Microthermal Climates

$T_a \text{ coldest month} < -3^\circ\text{C}$; $T_a \text{ warmest month} > 10^\circ\text{C}$ (allows tree growth)



Type E: Polar and Highland Climates

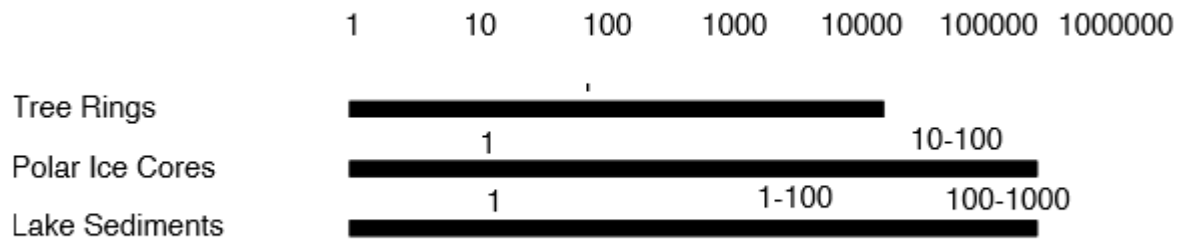
$T_a \text{ warmest month} < 10^\circ\text{C}$ (no tree growth)



And TYPE H – highlands

Methods of Documenting Past Climate

TIME RANGE AND RESOLUTION OF METHODS TO DOCUMENT PAST CLIMATES



Tree rings: → Ring width and density shows growth in a season (temperature and precipitation)

Lake cores: → Vegetation existence, moisture, show temperature precipitation

Glacial cores: → Warmer summers show more melt, colder winters show dense compaction

→ Very expensive

→ At lower depths, the pressure becomes very large

Temperature Variation Since Pliocene (5.5 million years ago)

5.5 million years ago → High frequency, low amplitude

→ Often had warmer and colder periods, but not severe change

1.5 million years ago → Low frequency, high amplitude

→ Doesn't change often, but when it does it's a fucking iceage

NOW → Same as 1.5 million years ago

→ Getting worse

Glacial Periods

Last 100 000 years, 10 000 in-between

Milankovitch Theory

Eccentricity of Orbit → 100 000 year cycle

→ Circular = no seasons

→ Elliptical = larger seasonality in one hemisphere, less seasonal in other

Tilt of Axis → 41 000 year cycle

→ 22 to 24.4 degrees (going towards 22)

→ Greater tilt = greater seasonality

Wobble → 22 000 year cycle

→ Changes equinoxes (dates of perihelion and aphelion change)

→ *Affects seasonality in different hemispheres (more in one, less in the other)*

Infiltration Capacity

Maximum rate at which a soil can absorb water

Lecture 11

Relative Time

Older rocks at the bottom, younger layers on top

Absolute Time

Radioactive dating

Core

Inner core is solid iron

Outer core is liquid iron that changes the magnetic field of the Earth

Mantle

Lower mantle, Fe Mg Si...

- Upper mantle :
- Solid upper mantle
 - Plastic asthenosphere
 - Solid lithosphere

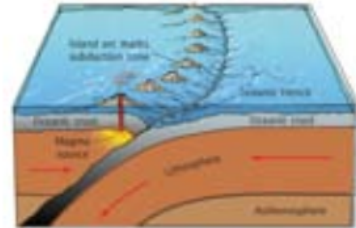
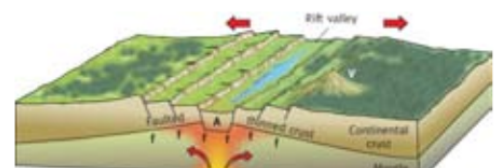
Tectonics:

Divergence

- Mid-oceanic ridges
- Plates spread out and magma comes out

Subduction

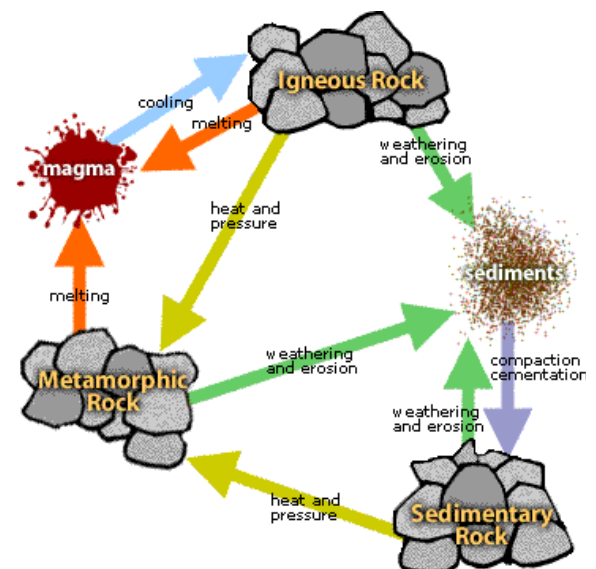
- Denser plate wedges under lighter one
- Lots of earthquakes and volcanoes
- Creates mountains and oceanic trenches



Types of Rocks:

- Igneous (magma cools and becomes a rock)
 - o Basalt
- Metamorphic (change from heat / pressure)
 - o Gneiss
 - o Marble
- Sedimentary (from sedimentation)
 - o Limestone
 - o Shale

Rock Cycle:



Weathering:

- **Physical**
 - *Dilatation (Spalling from stress in rocks)*
 - *Thermal stress (expansion/contraction)*
 - *Congelifraction (rocks break because of freezing water in pores)*
 - *Salt weathering (salt brought to pores by water, and expands during heat)*
- **Chemical**
 - *Depends on the rock composition and water content*
 - *Acid in water dissolves rocks (chemical reaction)*
 - *(stalactites in caves)*
- **Biological**
 - *Small bacterium can eat away the rock*

- *More weathering in lower latitudes with high heat, rainfall, humidity*

Weathering Zone

Area weathered above the underlying undamaged bedrock

Shallow in hot and cold regions

Deep in wet and humid regions

Weathering Front

Zone that the bedrock and weathered rock meet

Erosion

Once the particle is broken off from weathering, erosion moves the particles and removes it from the area

- *Glacial Erosion*
- *Fluvial Erosion*
- *Karst Erosion (Sinkholes, Canyons)*
- *Coastal Erosion*

1.) Detachment (rocks particles detached)

- a. *Weathering*
- b. *Glaciation (ice glaciers freeze rock, cause water to expand, rock cracks)*
- c. *Abrasion (suspended rocks hit the rock)*
- d. *Raindrop impact*

2.) Entrainment (velocity of water is large enough to cause motion of the particle)

- *Small particles need high velocity because they have cohesion to the original rock*
- *Large particles are heavy and require high entrainment*

3.) Transport

- a. *Suspension*

- b. *Saltation (bouncing)*
- c. *Traction (dragging)*
- d. *Solution (dissolved)*

4.) **Deposition**

- *Happens when the velocity of the flow slows*

Landform

Single unit (one mountain, a sinkhole, a sand dune)

Landscape

Aggregation of landforms (Rocky Mountains, a whole desert)

Degradational Process

Forces that work to flatten the landscape

- *Avalanches, mudslides, landslides*

Aggradational Process

Land height is levelled out or increased

- *Sedimentation*

Factors Affecting Landscape

- *Running water (erosion)*
- *Glaciers (erosion)*
- *Wind (erosion)*
- *Coastal waves (how much land there will be)*
- *Chemical dissolution (weathering, karstic landscape)*

Lecture 12

Soil Components:

- Minerals (45%)
- Voids (25%)
- Water (25%)
- Organic matter (5%)
 - Humus (broken down organic matter)

Water in Soil

- Surface water
- Gravitational water (drains to GWT)
- Capillary water
- Hygroscopic water (held to the surface of particles from tension)
- Chemically bonded water

Factors for Soil Formation:

Parent Material

Residual soils (minerals from underlying rock)

Transported soils

Coversands (sands transported by winds)

Climate

Soil moisture

Organisms

Micro-organisms eat and create organic content

Macro-organisms

Worms create trails that allow water and air to flow, and their shit creates bacteria

Termites remove all vegetation for their nests, accelerate organic decay

Topography

Amount of precipitation

Amount of erosion (thickness of soil)

Infiltration volumes

Time

Takes time for shit to happen

Describing Soil

Soil Colour

Grain Size Distribution

Soil Structure

Soil Porosity

Soil Permeability

Field Capacity

Soil Processes

Addition

Gains made by the soil when sun, water, organic matter is added

Transformation

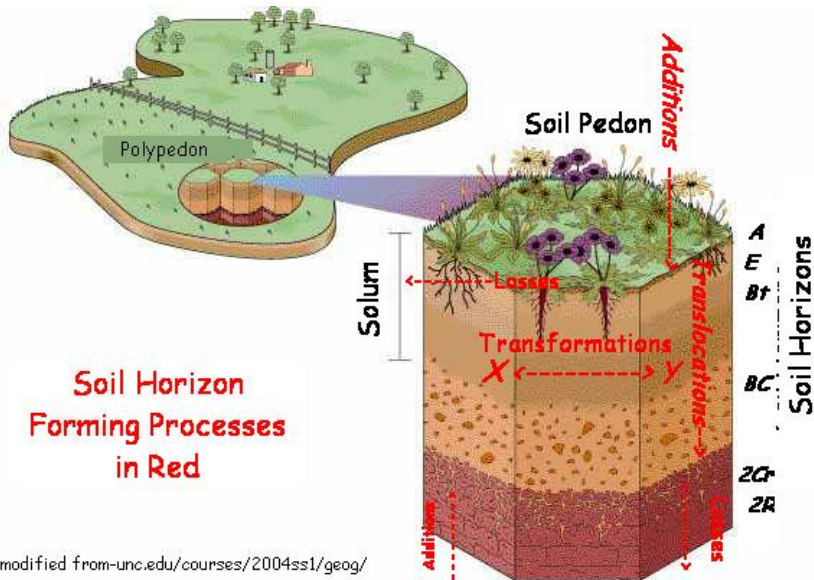
Rock weathering and decomposition of organic material

Depletion (losses)

Water percolates downwards and brings particles with it from upper layers

Translocation

Depletion by means of physical movement (worms and shit)



Soil Structure

Platy

Little cinnamon toast crunches

Prismatic

Vertical strength

Angular

Considerable strength

Spherical

Porous, weak



Lecture 13

Interception

Water falls onto vegetation and evaporates before reaching the soil

Infiltration

Water seeping into the soil

Runoff

Occurs when there is precipitation and the soil is fully saturated

Factors affecting infiltration capacity

Soil type (porosity, water content)

Vegetation (amount of cover)

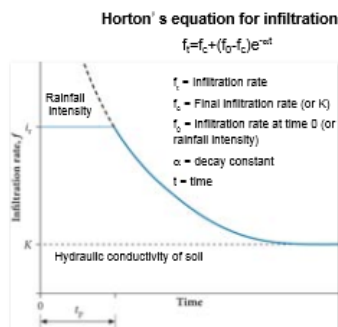
Land Use

Weather prior to rainfall (if ground is saturated)

Duration and intensity of pcp

Infiltration capacity (during rain event)

1. At the beginning, infiltration rate = rainfall intensity (i_r)
2. After some time, due to packing of the soil, swelling of the soil, in-washing of fine particles, the soil porosity decreases
3. This allows for surface ponding. After ponding occurs (at t_p), infiltration rate decreases until it reaches the hydraulic conductivity (K) of the soil.



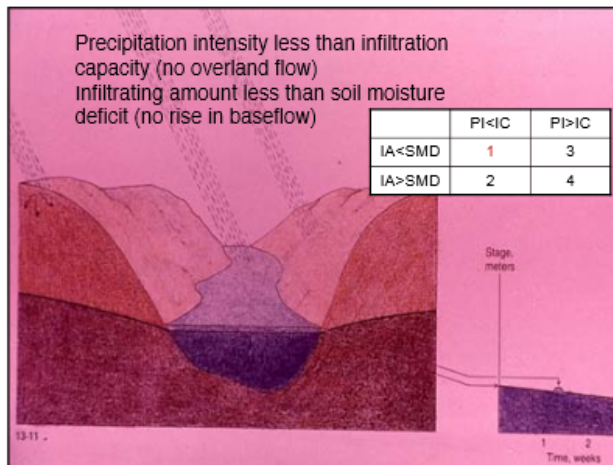
Horton's Theory

Applies to semi-arid areas with low vegetation cover (low infiltration capacity) and high intensity storms

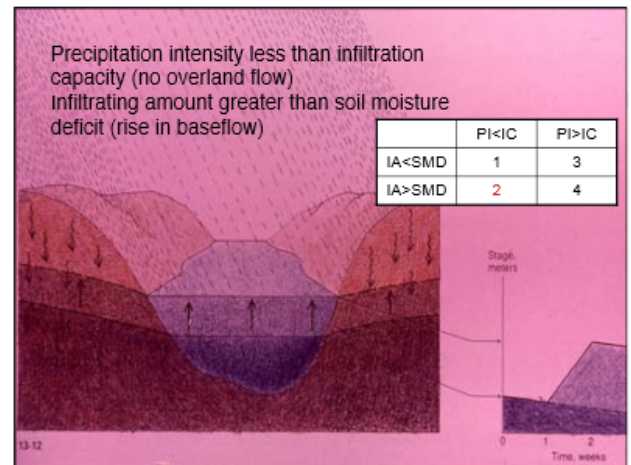
Surface Runoff (SR) controlled by **Precipitation Intensity (PI)** vs. **Infiltration Capacity (IC)**

Baseflow controlled by **Infiltrating Amount (IA)** vs. **Soil Moisture Deficit (SMD)**

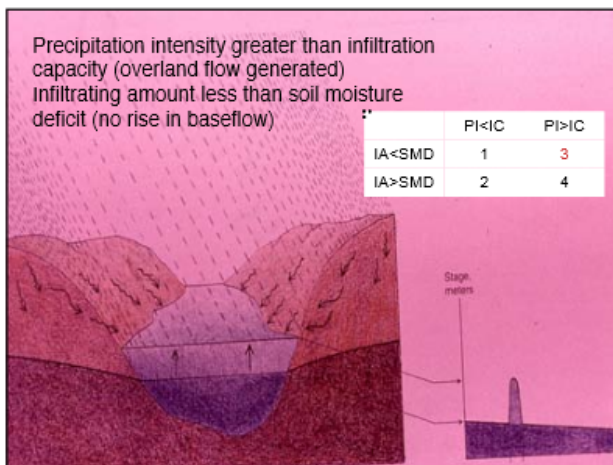
	$PI < IC$	$PI > IC$
$IA < SMD$	1	3
$IA > SMD$	2	4



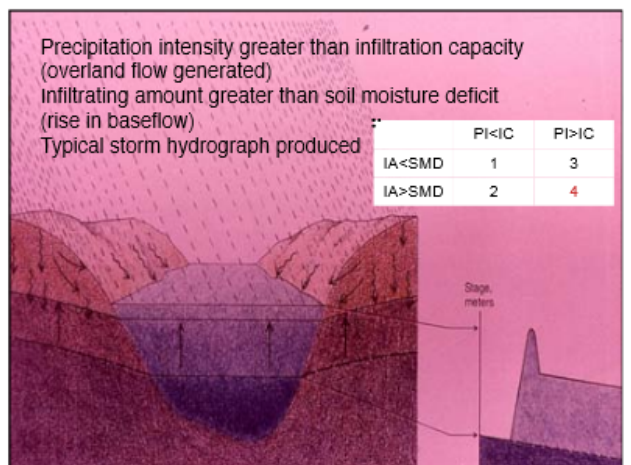
Hortonian type 1



Hortonian type 2



Hortonian type 3



Hortonian type 4

Formation of Lakes

Glacial Activity

Tectonic Processes (faulting of crust)

Volcanic Activity (crater lakes)

Meteorites

Solution lakes (sinkholes)

Organic (beavers)

Human (dams)

Etc..

Thermal Stratification of Lakes

Epilimnion

Surface layer that is warm and less dense from the Sun

Thermocline

Transition layer that has large thermal gradient

Hypolimnion

Bottom of lake, cool dense water

Holomictic Lake

Entire lake is involved in the mixing

Seasonal mix

Meromictic Lake

Could be centuries before the lake mixes

Pink Lake

Oligotrophic Lake

Clear, lack nutrients

Mesotrophic Lake

Clear, average nutrients

Eutrophic Lake

A lot of nutrients

Algae bloom

Hypertrophic Lakes

Too much nutrients

Turbid

Shit-tonne of algae bloom, and low DO