

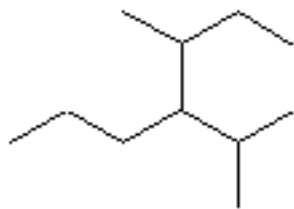
Your Name: _____

Student #: _____

Exercise	key
1	
2	
3	
4	
5	
6	
7	
8	
9	
10	
11	
12	

1. Deliver only the solution key in the above table. You can keep the text. The solution key will be posted today on the web.
2. You must respond to all exercises.

Question 1



An IUPAC name for is:

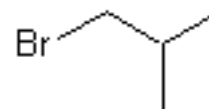
- 5-Methyl-4-(1-methylethyl)heptane
- 5-Methyl-4-(1-methylpropyl)hexane
- 2-Methyl-3-(1-methylpropyl)hexane
- 2-Methyl-3-(2-methylpropyl)hexane
- 3-Methyl-4-(1-methylethyl)heptane

Question 2

Which of the following pairs of compounds represent pairs of constitutional isomers?

- 2-Methylbutane and pentane
- 2-Chlorohexane and 3-chlorohexane
- sec-Butyl bromide and tert-butyl bromide
- Propyl chloride and isopropyl chloride
- All of the above

Question 3

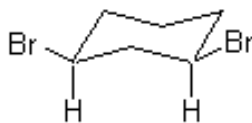


What is the common name for this compound?

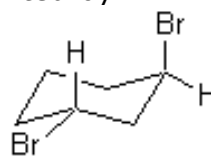
- Isobutyl bromide
- tert-Butyl bromide
- sec-Butyl bromide
- Butyl bromide
- Bromo-sec-butane

Question 4

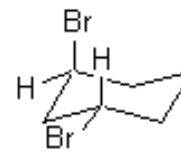
cis-1,3-Dibromocyclohexane is represented by



I



II



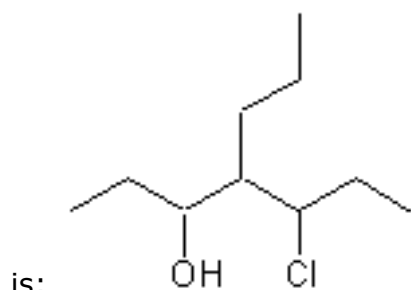
III

structure(s):

- I
- II
- III
- II and III
- I and II

Question 5

A correct IUPAC name for the following compound



- 4-propyl-5-chloro-3-heptanol
- 3-hydroxy-4-propyl-5-chloroheptane
- 4-propyl-3-chloro-5-heptanol
- 4-(1-chloropropyl)-3-heptanol
- 5-chloro-4-propyl-3-heptanol

Question 6

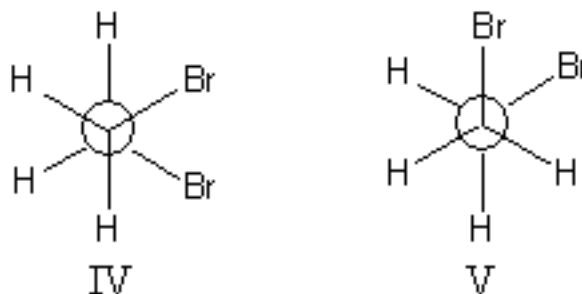
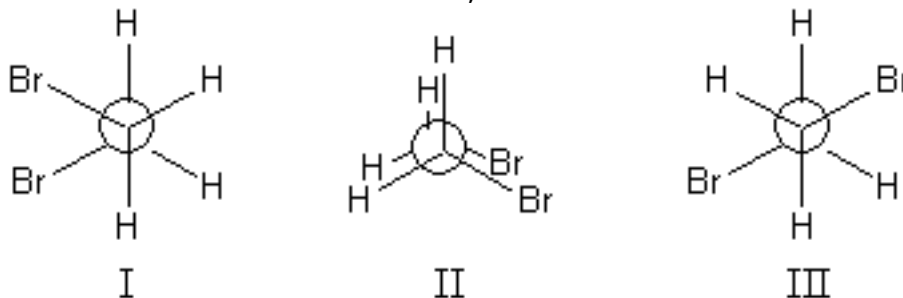
Which cycloalkane has the greatest ring strain?

- Cyclobutane
- Cyclohexane
- Cyclopropane
- Cycloheptane

Cyclopentane

Question 7

The most stable conformation of 1,2-dibromoethane



is:

- I
- II
- III
- IV
- V

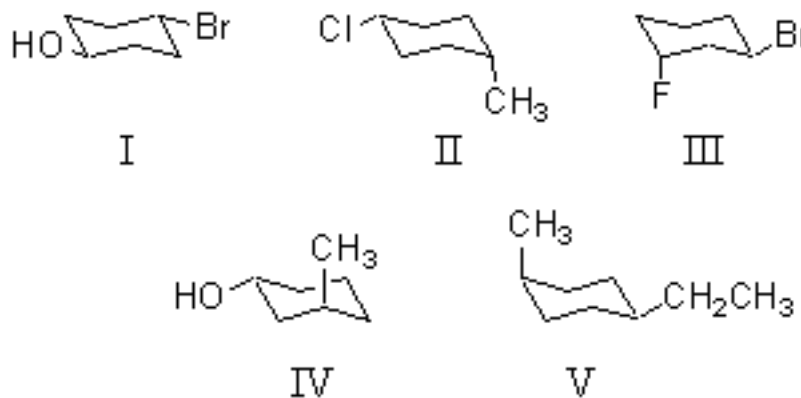
Question 8

The most stable conformation of cis-1-tert-butyl-2-methylcyclohexane is the one in which:

- the tert-butyl group is axial and the methyl group is equatorial.
- both groups are equatorial.
- the twist boat conformation is adopted.
- the methyl group is axial and the tert-butyl group is equatorial.
- both groups are axial.

Question 9

Which of the following can be described as *cis*

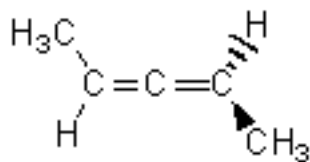


isomers?

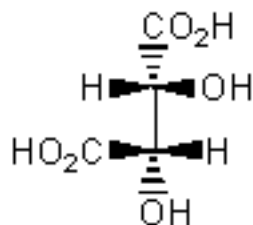
- I
- II, V
- III, IV
- I, III and IV
- None of the above are *cis* isomers.

Question 10

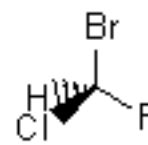
Which of the following molecules is



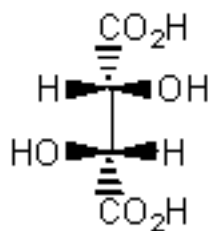
I



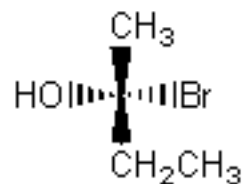
II



III



IV



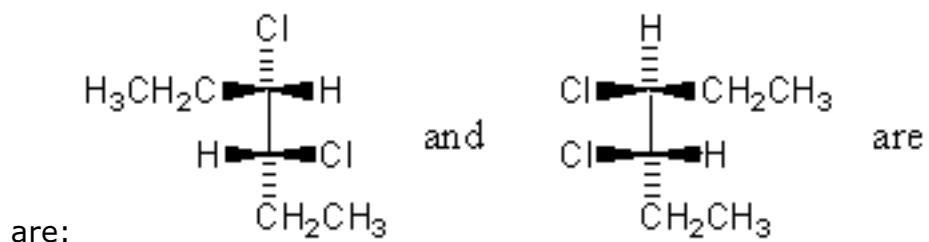
V

achiral?

- I
- II
- III
- IV
- V

Question 11

The molecules shown



- not isomeric.
- enantiomers.
- diastereomers.
- constitutional isomers.
- two conformations of the same molecule.

Question 12

(2R,4S) -2,4-Dichloropentane and (2S,4R)-2,4-dichloropentane are:

- conformational isomers
- constitutional isomers
- diastereomers
- enantiomers
- identical