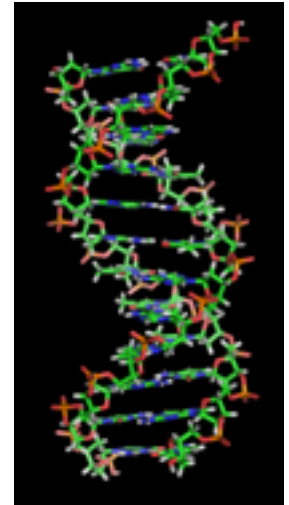


Proteins

CHEM112

Week 8

Dr J Carran



Lab 7

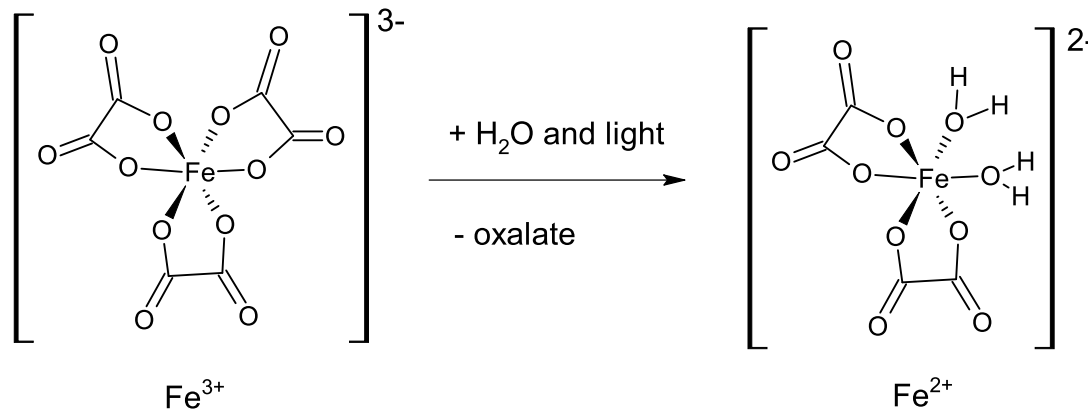
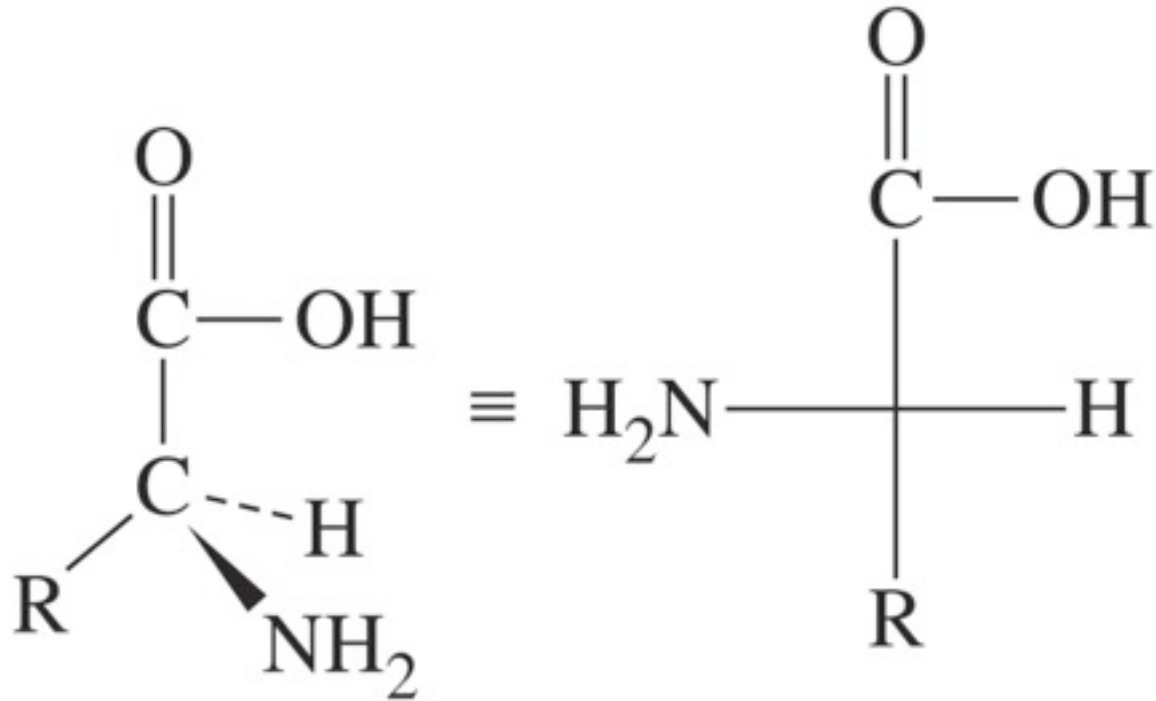


Figure 32: Reduction of tris(oxalato)ferrate(III) (aka ferrioxalate) to diaquabis(oxalato)ferrate(II) ion.

Purpose

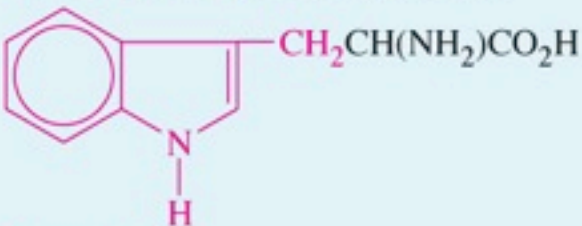
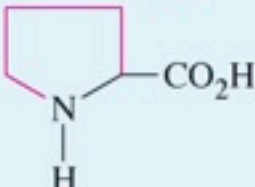
You will prepare and examine the properties of photosensitive solutions (i.e. sensitivity to light and heat). You will gain experience with chemicals that are sensitive to handling and determine its properties by spectrophotometry.

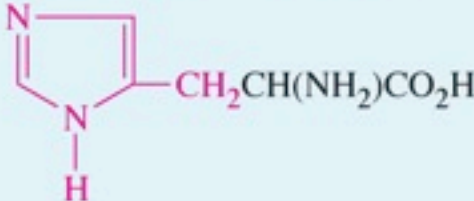
28-4 Proteins



An L-amino acid

TABLE 28.3 Some Common Amino Acids

Name	Symbol	Formula	pI ^a
Neutral Amino Acids			
Glycine	Gly	$\text{HCH}(\text{NH}_2)\text{CO}_2\text{H}$	6.03
Alanine	Ala	$\text{CH}_3\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	6.10
Valine ^b	Val	$(\text{CH}_3)_2\text{CHCH}(\text{NH}_2)\text{CO}_2\text{H}$	6.04
Leucine ^b	Leu	$(\text{CH}_3)_2\text{CHCH}_2\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	6.04
Isoleucine ^b	Ileu or Ile	$\text{CH}_3\text{CH}_2\text{CH}(\text{CH}_3)\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	6.04
Serine	Ser	$\text{HOCH}_2\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	5.70
Threonine ^b	Thr	$\text{CH}_3\text{CH}(\text{OH})\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	5.6
Phenylalanine ^b	Phe	$\text{C}_6\text{H}_5\text{CH}_2\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	5.74
Methionine ^b	Met	$\text{CH}_3\text{SCH}_2\text{CH}_2\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	5.71
Cysteine	Cys	$\text{HSCH}_2\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	5.05
Cystine	(Cys) ₂	$[\text{SCH}_2\text{CH}(\text{NH}_2)\text{CO}_2\text{H}]_2$	5.1
Tyrosine	Tyr	$4\text{-HOC}_6\text{H}_4\text{CH}_2\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	5.70
Tryptophan ^b	Trp		5.89
Proline ^c	Pro		6.21

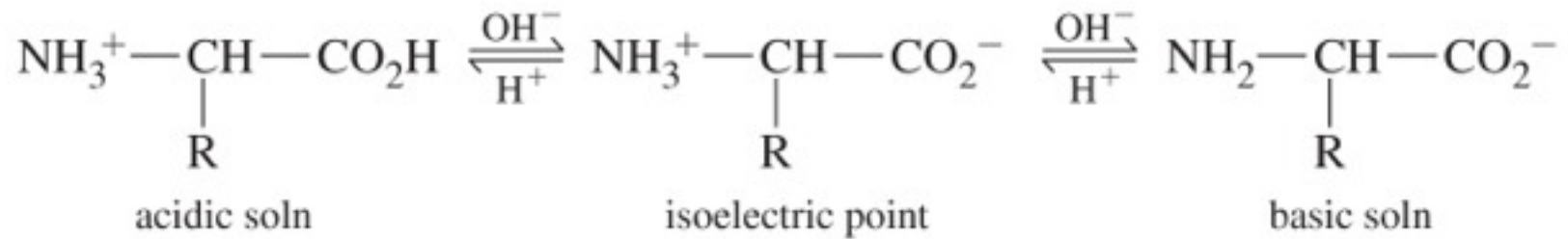
Acidic Amino Acids			
Aspartic acid	Asp	$\text{HO}_2\text{CCH}_2\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	2.96
Glutamic acid	Glu	$\text{HO}_2\text{CCH}_2\text{CH}_2\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	3.22
Basic Amino Acids			
Lysine ^b	Lys	$\text{H}_2\text{N}(\text{CH}_2)_4\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	9.74
Arginine	Arg	$\text{H}_2\text{NC}(=\text{NH})\text{NH}(\text{CH}_2)_3\text{CH}(\text{NH}_2)\text{CO}_2\text{H}$	10.73
Histidine	His		7.58

^apH of isoelectric point.

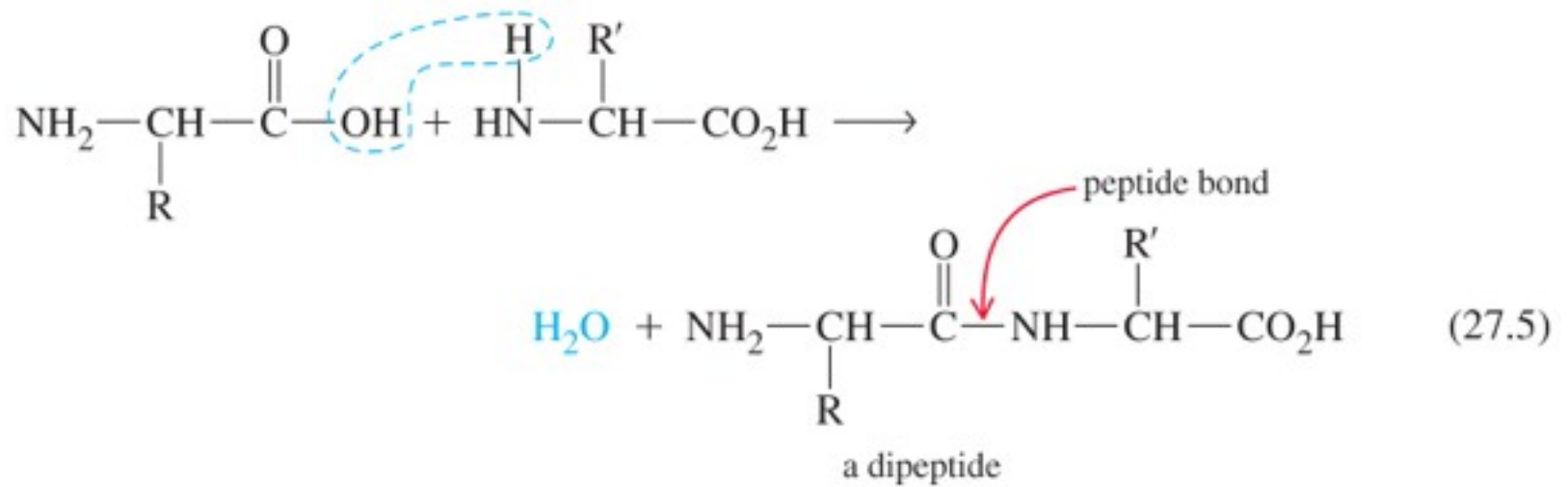
^bEssential amino acids. In addition, arginine and glycine are required by the chick, arginine by the rat, and histidine by human infants.

^cThe secondary amino group makes proline an α -imino acid. Nevertheless, it is commonly listed with amino acids.

Zwitterion



Peptides



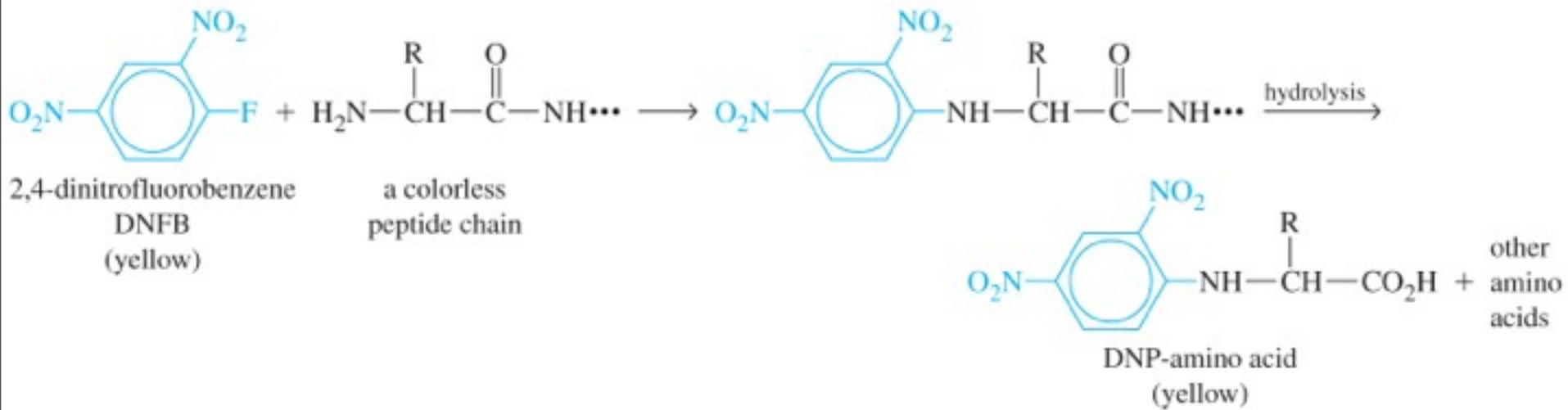


FIGURE 28-10
Experimental determination of amino acid sequence

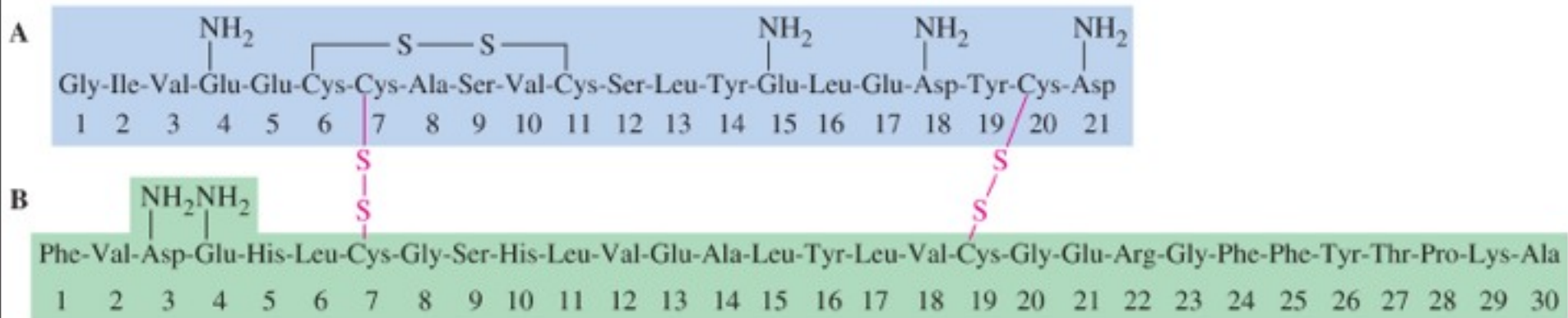
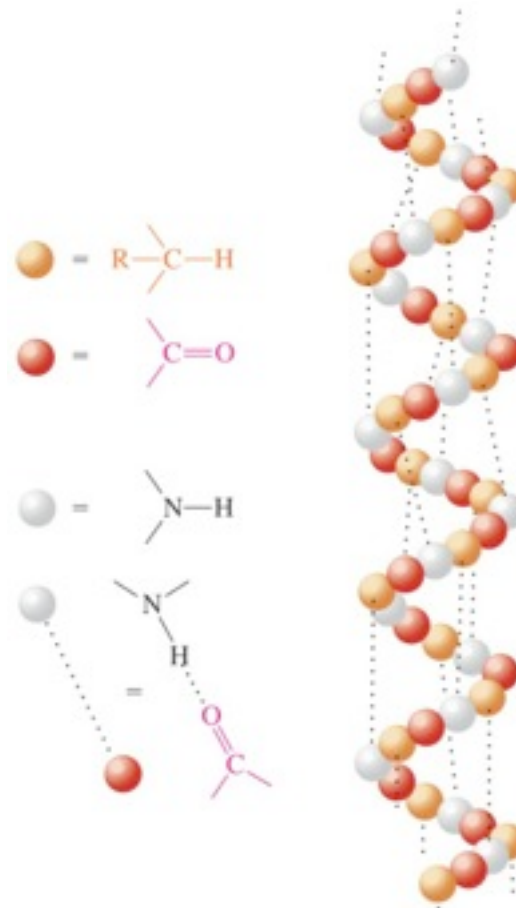


FIGURE 21-11

Amino acid sequence in beef insulin – primary structure

The Structure of Proteins



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▲ FIGURE 28-12

•Secondary structure of a protein – an α helix

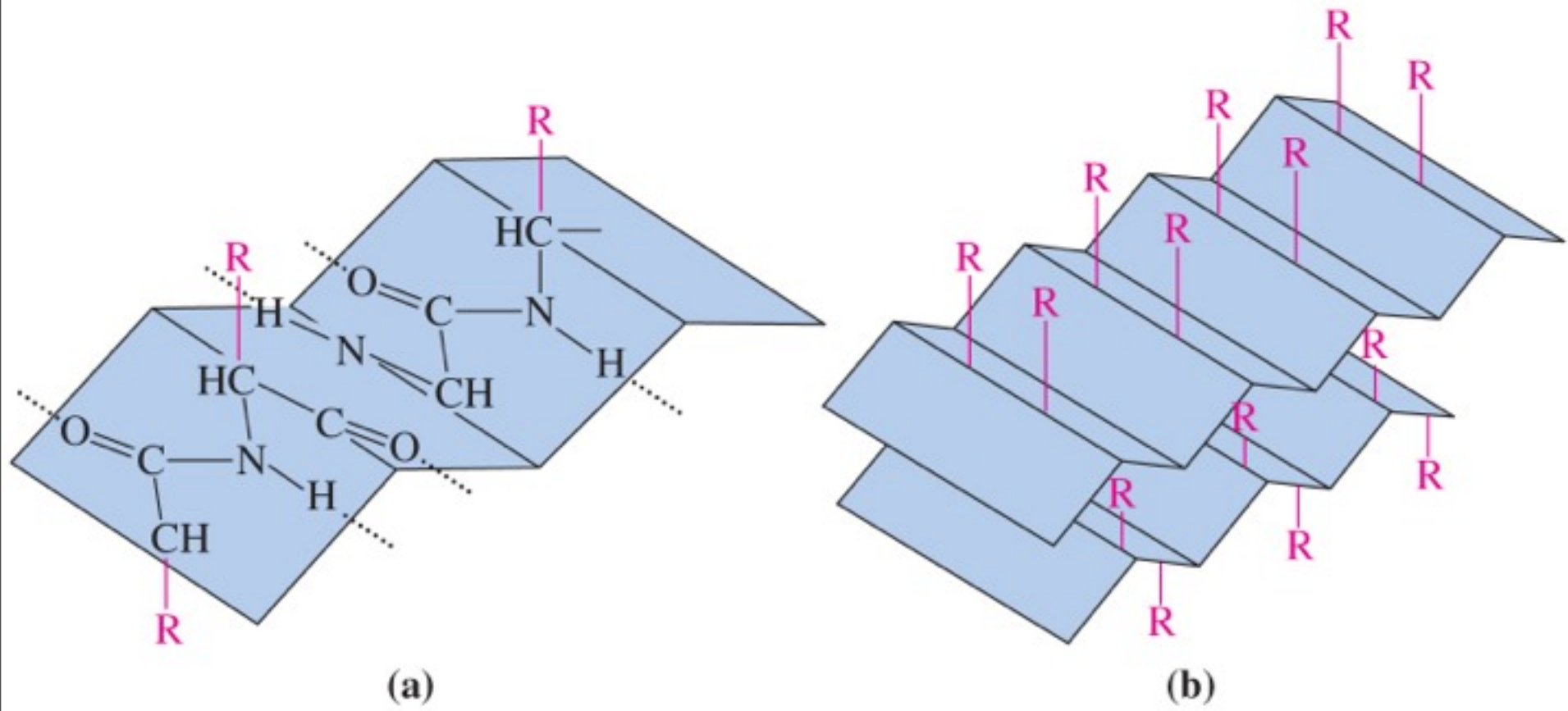


FIGURE 28-13
Pleated-sheet model of β -keratin

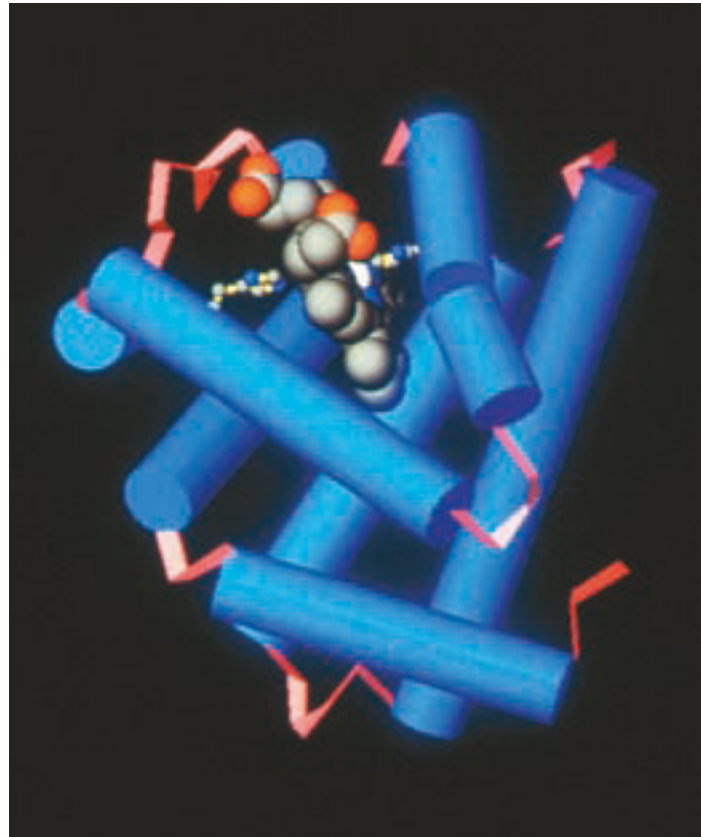


FIGURE 28-14
Representation of the tertiary structure of myoglobin

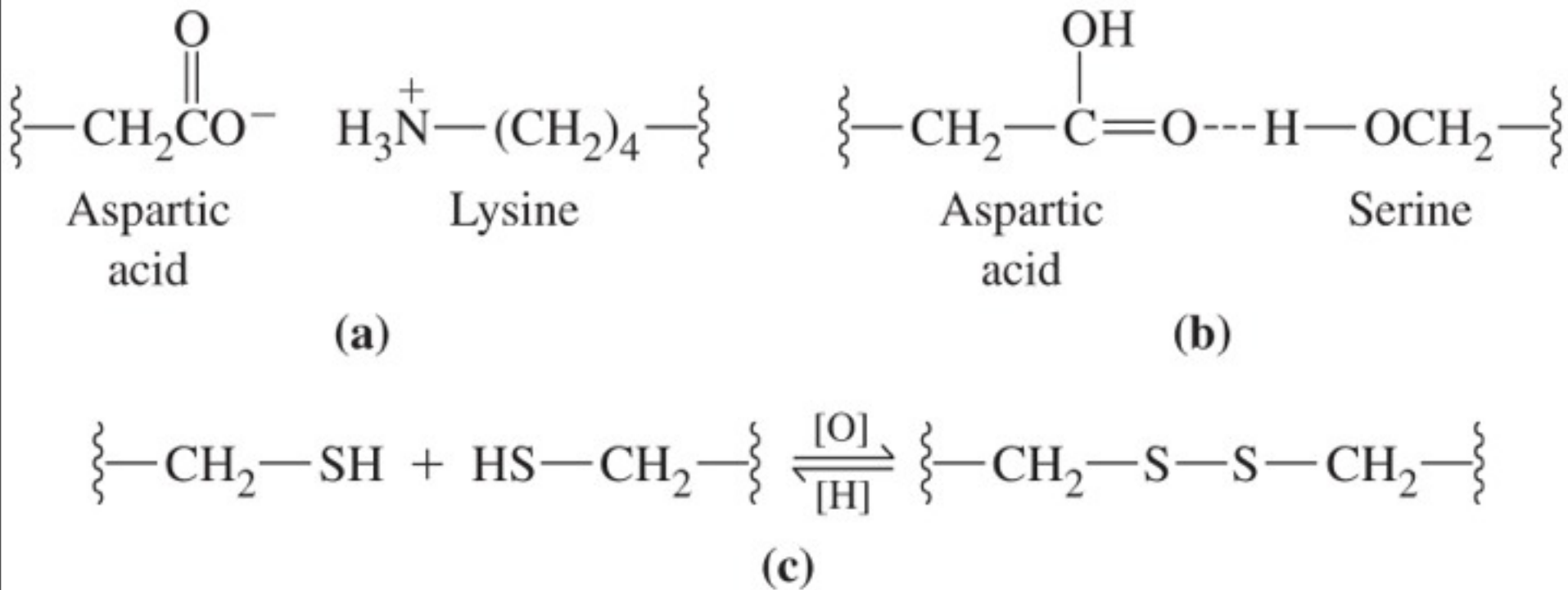
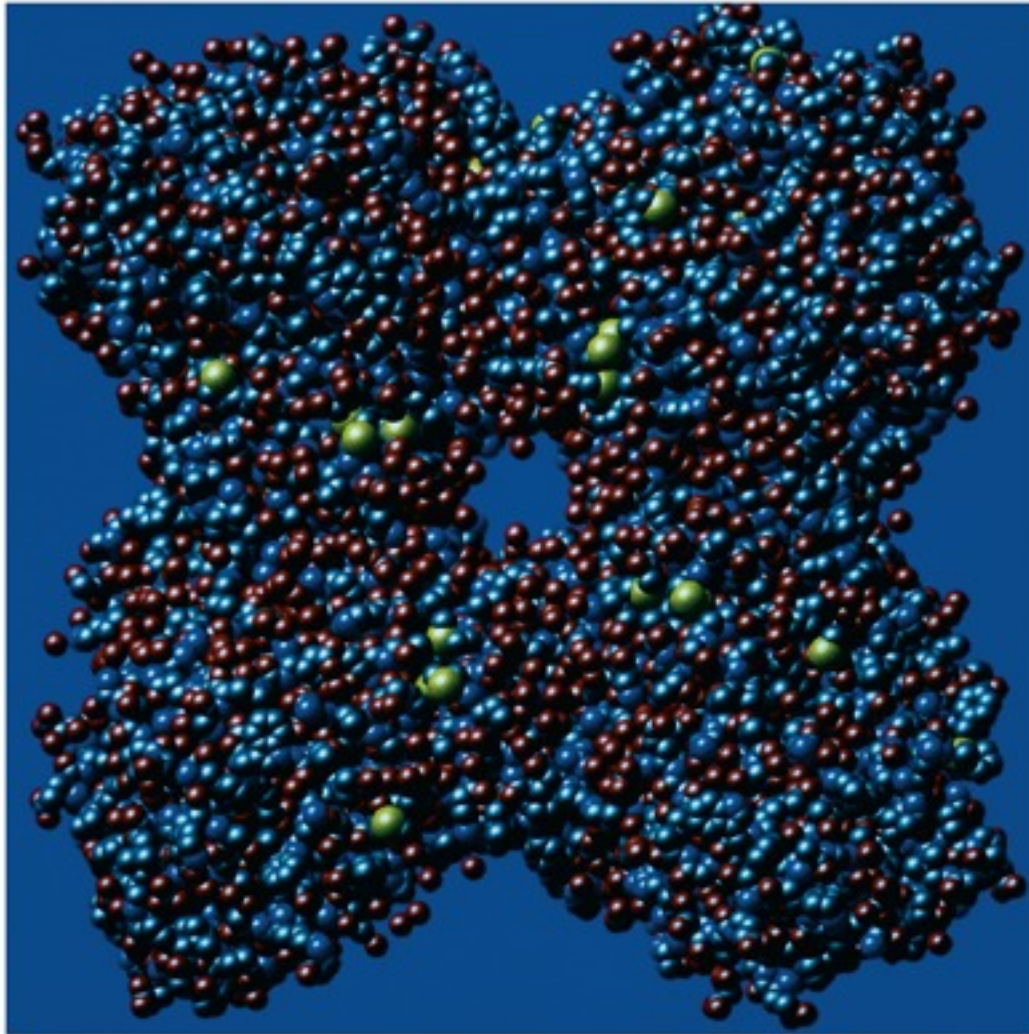


FIGURE 28-15
Linkages contributing to tertiary structure of proteins

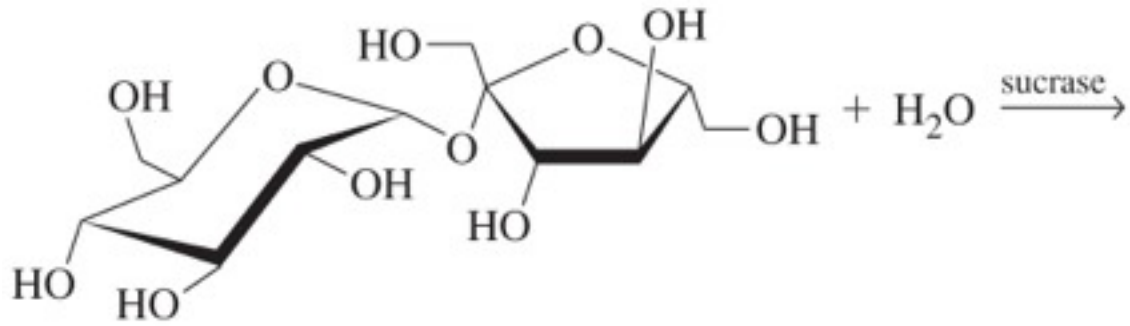


▲ A computer image of the three dimensional structure of ribulose-1,5-biphosphate carboxylase-oxygenase (RuBisCo)

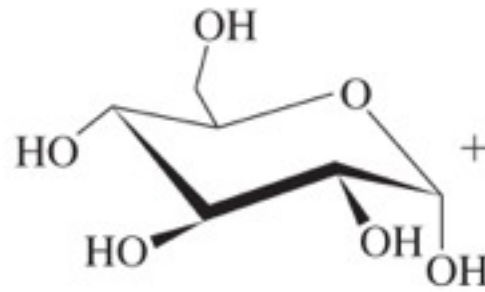
Enzymes



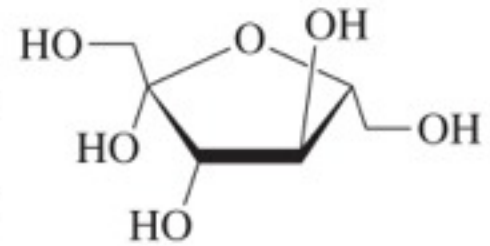
Enzymes



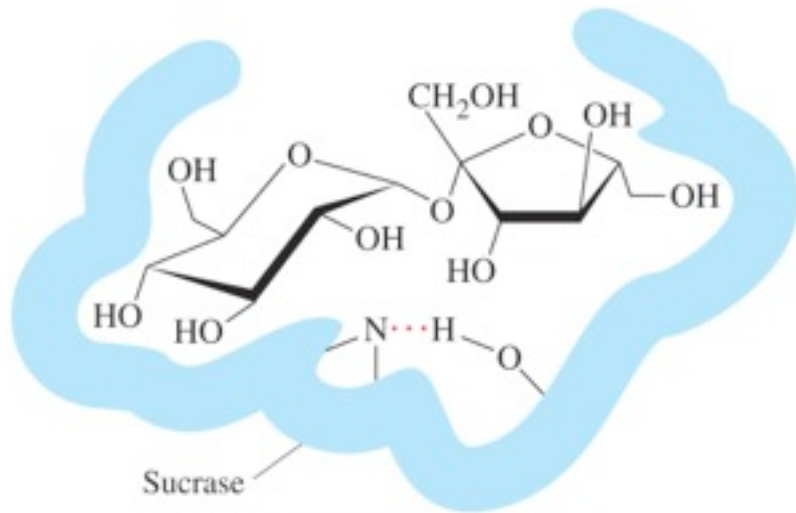
sucrose
 $C_{12}H_{22}O_{11}$



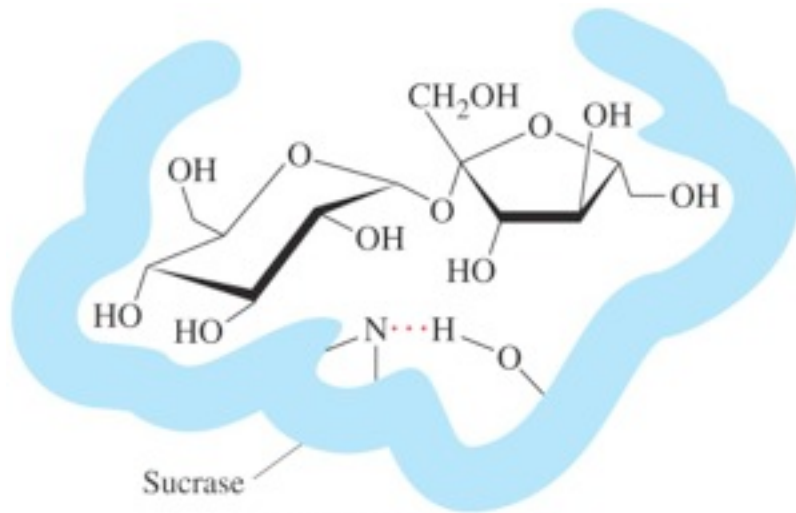
glucose
 $C_6H_{12}O_6$



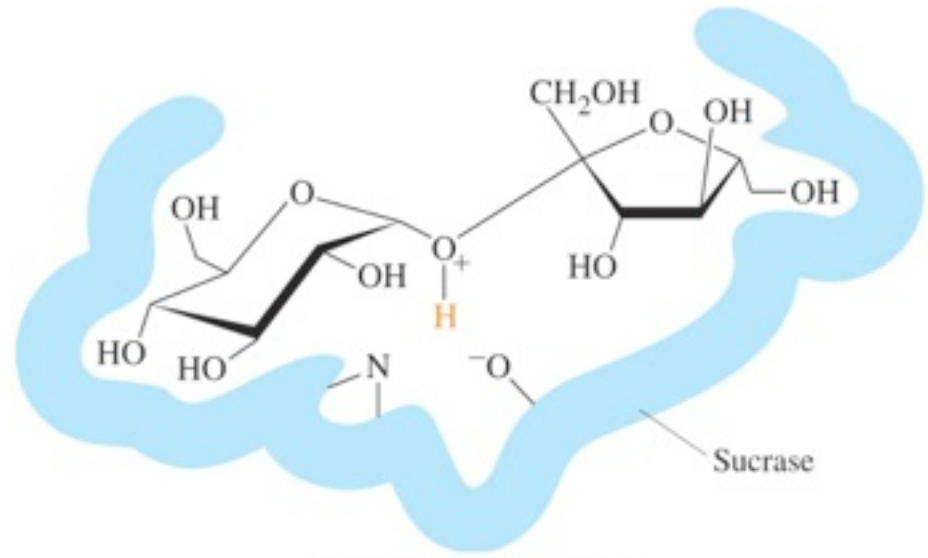
fructose
 $C_6H_{12}O_6$



▲ FIGURE 28-19
Sucrase binding a sucrose molecule



▲ FIGURE 28-19
Sucrase binding a sucrose molecule



▲ FIGURE 28-20
Change in enzyme conformation after binding of substrate

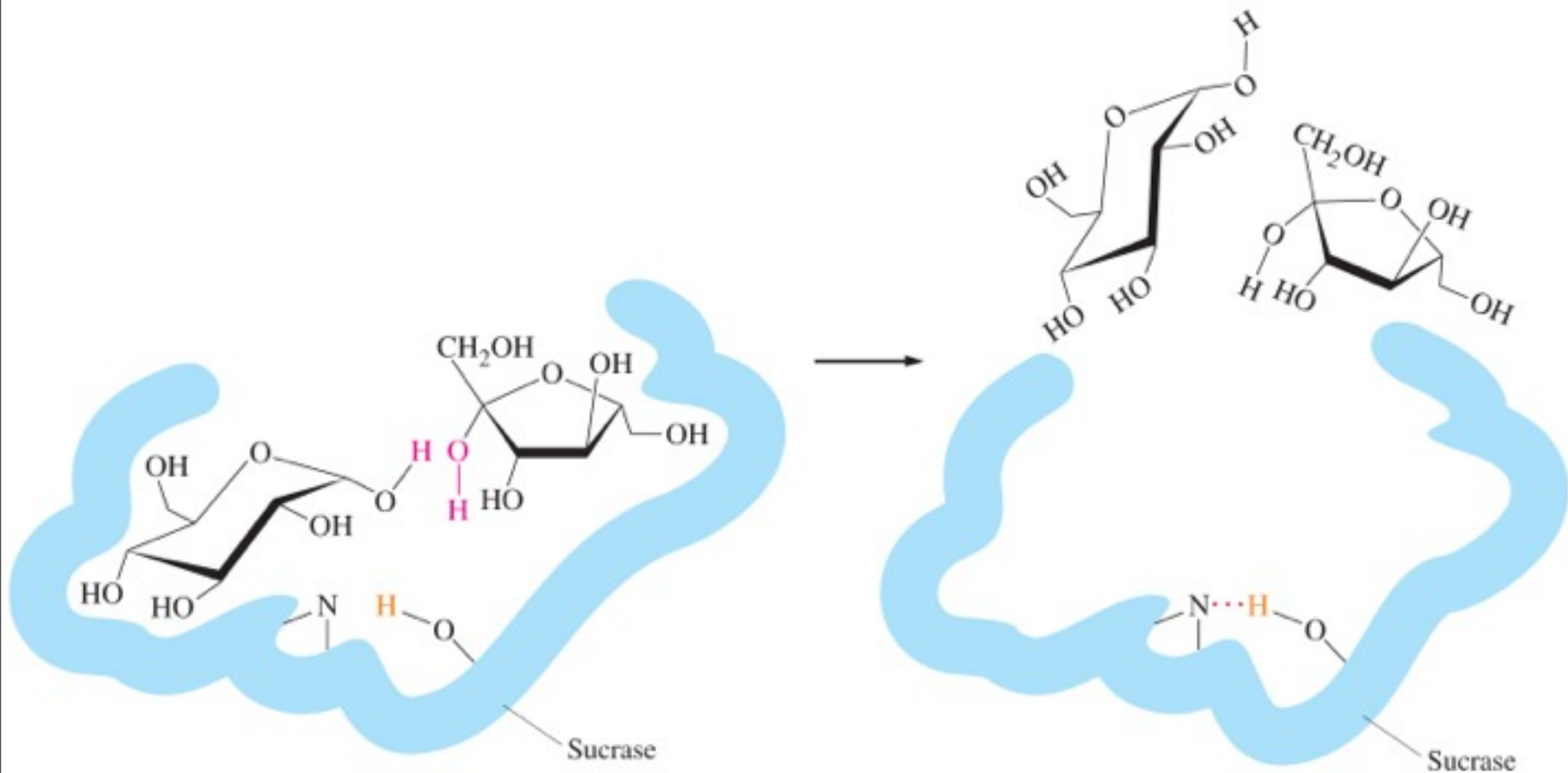
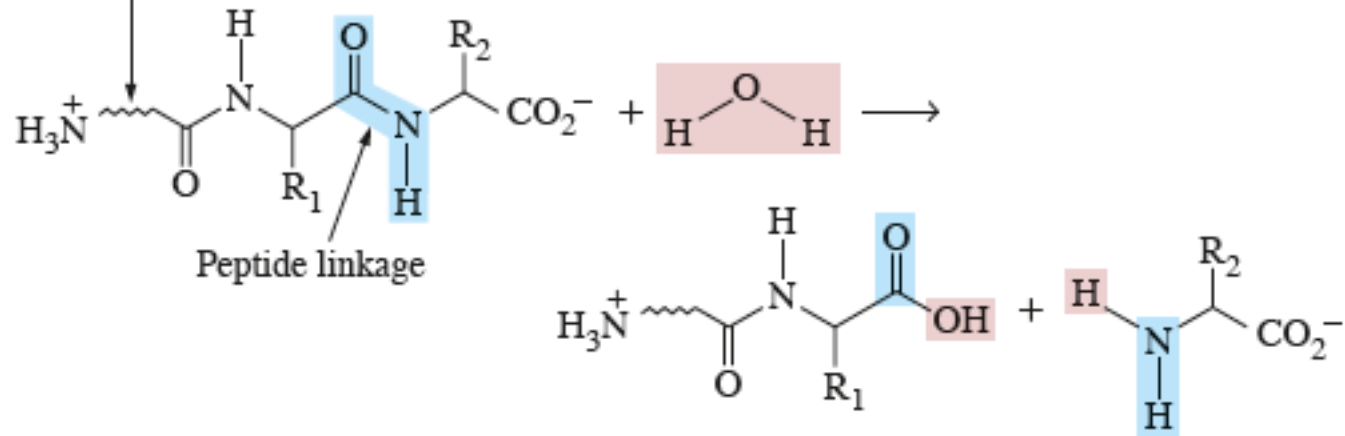
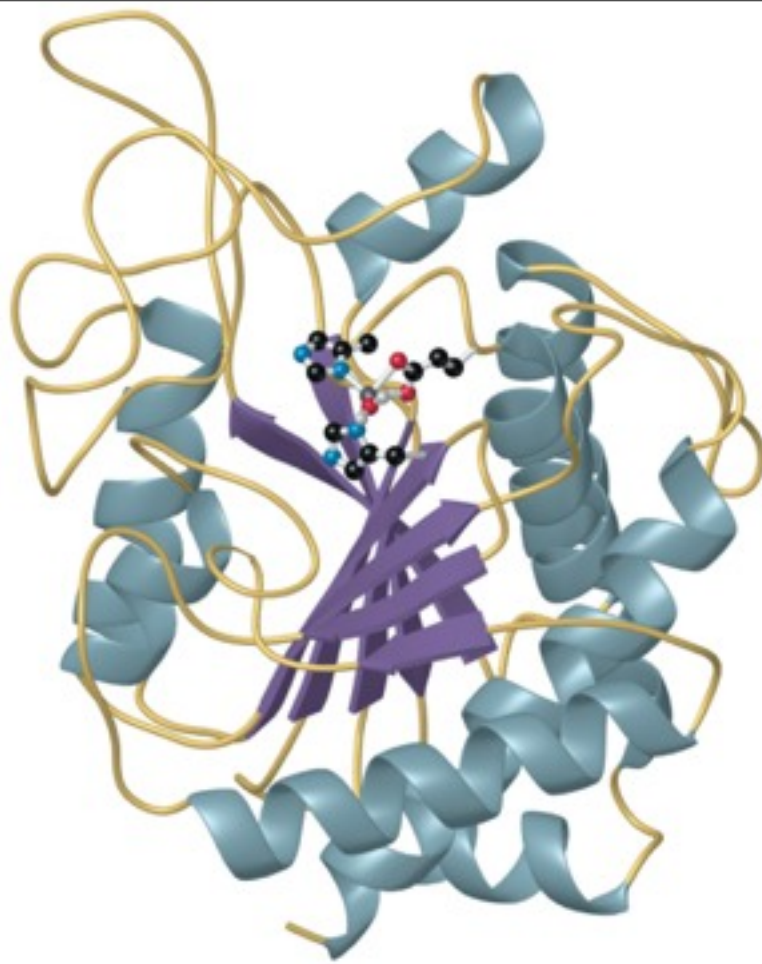


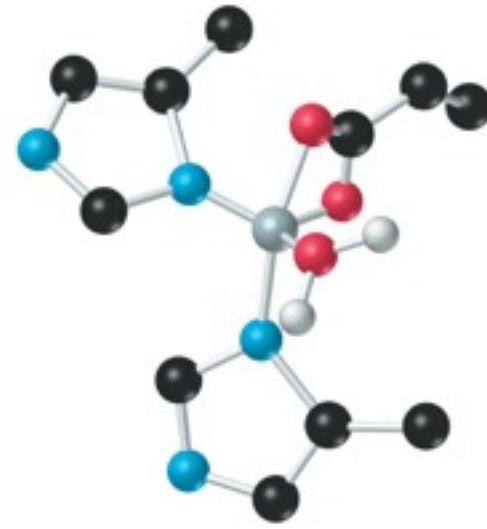
FIGURE 28-21
Completion of the enzyme-catalyzed reaction

Many amino acids
leading to the amino
terminal group





(a)



(b)

FIGURE 28-22
Structure of carboxypeptidase

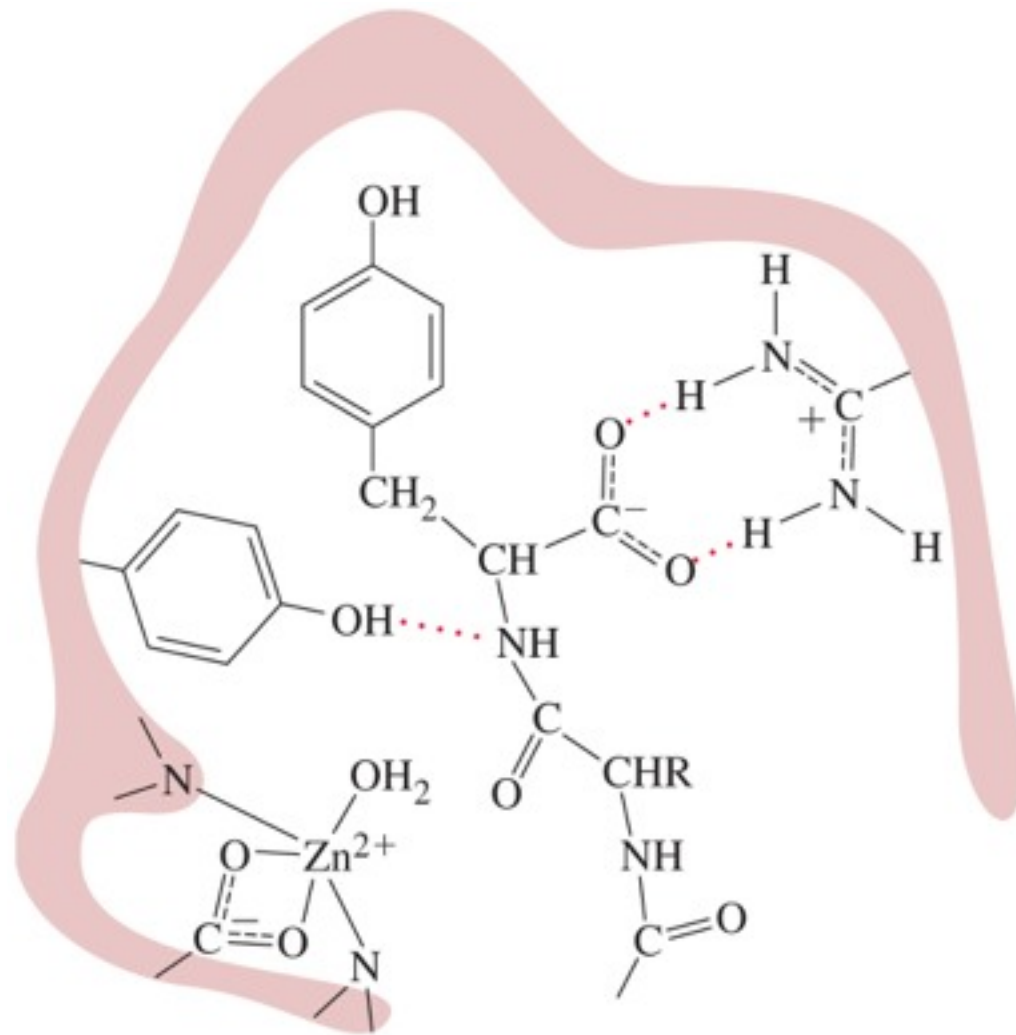


FIGURE 28-23

Binding of protein to the active site of carboxypeptidase